

Sulfur-Fluoride Exchange (SuFEx)

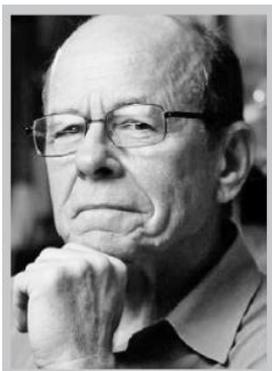
Literature Seminar
2019/11/28

M2 Katsuya Maruyama

Contents

- Introduction
- Synthesis of S(VI)-F unit
- Applications of SuFEx

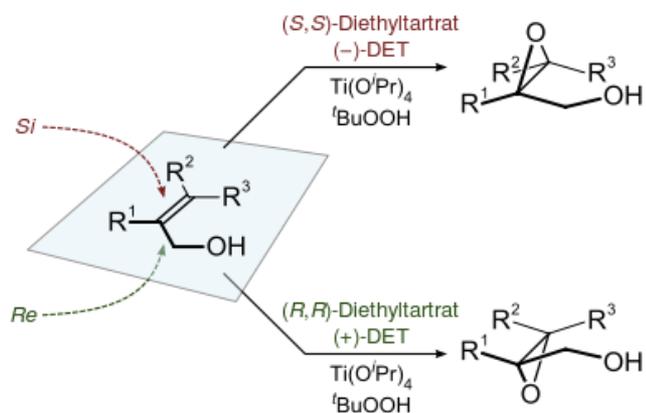
K. B. Sharpless



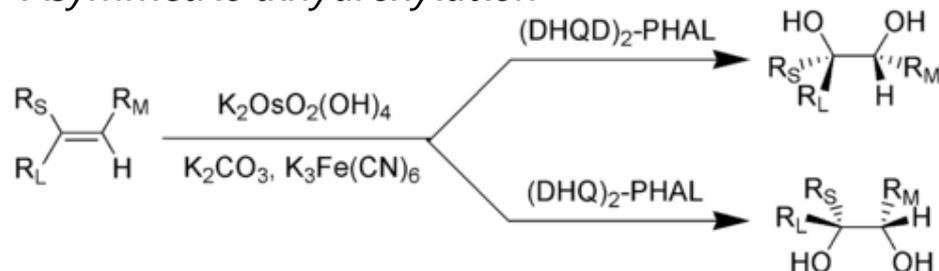
Nobel Laureate K. Barry Sharpless became W. M. Keck Professor of Chemistry at The Scripps Research Institute and The Skaggs Institute of Chemical Biology in 1990. Previously a professor at MIT and Stanford, he was educated at Dartmouth College (BA 1963), Stanford (PhD 1968 with E. E. van Tamelen; postdoc 1969 with J. P. Collman), and Harvard (postdoc 1970, K. E. Bloch).

The Nobel Prize in Chemistry (2001):
"for his work on chirally catalyzed oxidation reactions."

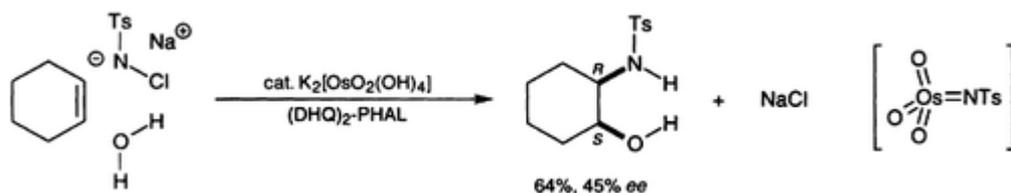
Asymmetric epoxidation



Asymmetric dihydroxylation

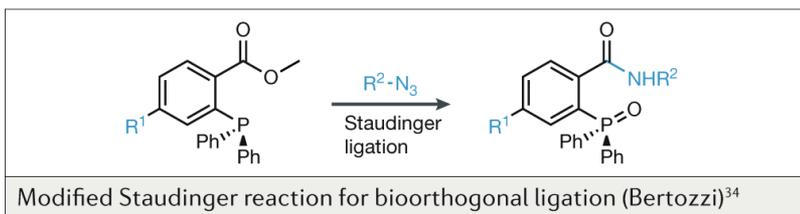
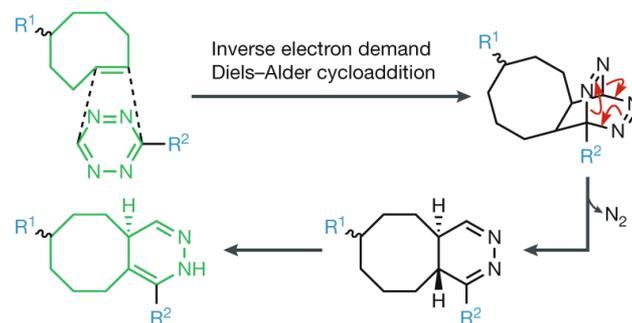


Asymmetric aminohydroxylation



Click Chemistry

Introduction of tetrazine–*trans*-cyclooctene as a fast copper-free bioorthogonal reaction (Fox)³⁷



2000

2001

2002

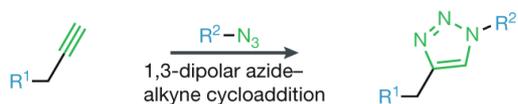
2007

2008

2014

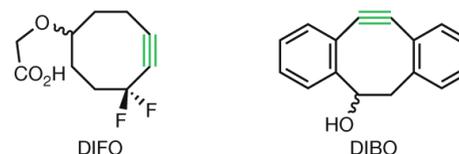
Introduction of click chemistry (Kolb–Finn–Sharpless)³¹

Copper(I)-catalysed azide–alkyne cycloaddition (Meldal–Sharpless)^{35,36}



Introduction of sulfur(vi) fluoride exchange as a promising new class of click chemistry reaction (Sharpless)⁵⁵

Development of DIFO for copper-free click chemistry (Bertozzi)⁵⁷

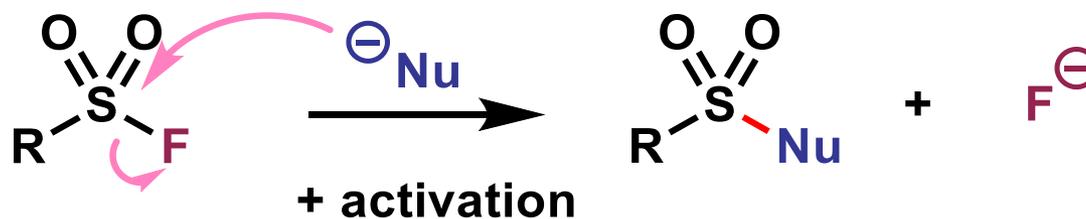


Nat. Rev. Chem. **2018**, 2, 202.

- **Robust connection**
- **High functional group tolerance**
- **Fast kinetics**
- **Easy operation**

SuFEx Reaction

General scheme



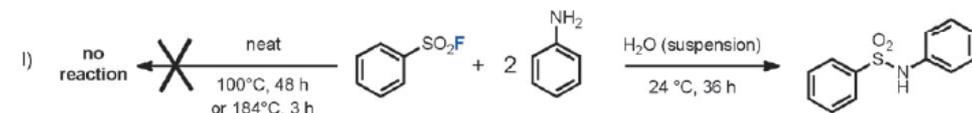
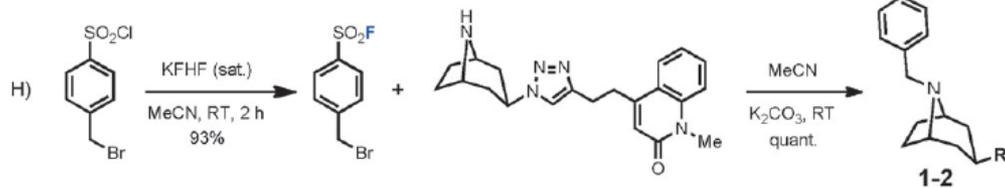
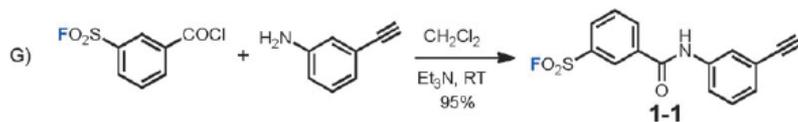
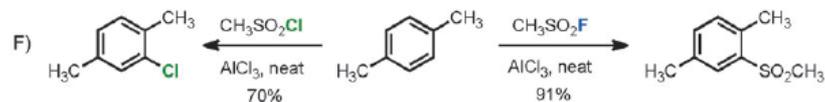
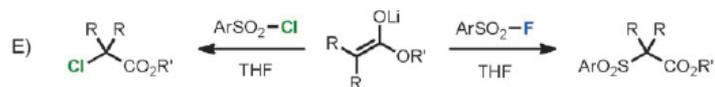
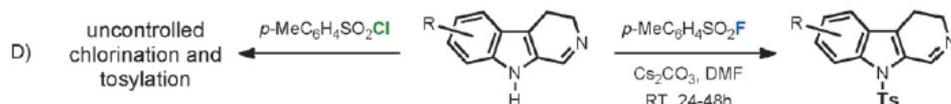
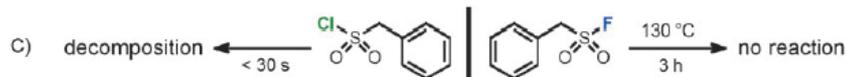
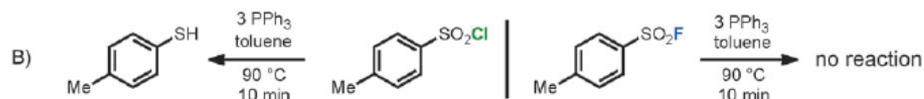
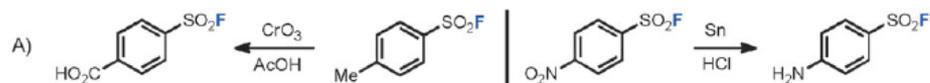
Stable S-F bond (necessity of activation)

Metal free

diversity of linkage

“Context dependent” Click reaction

Property of Sulfonyl Fluoride



Oxidation/reduction

Heating

BDE
 SO_2F_2 90.5 kcal/mol
 SO_2Cl_2 46 kcal/mol

Substitution at S(VI) center
 (no fluorination)

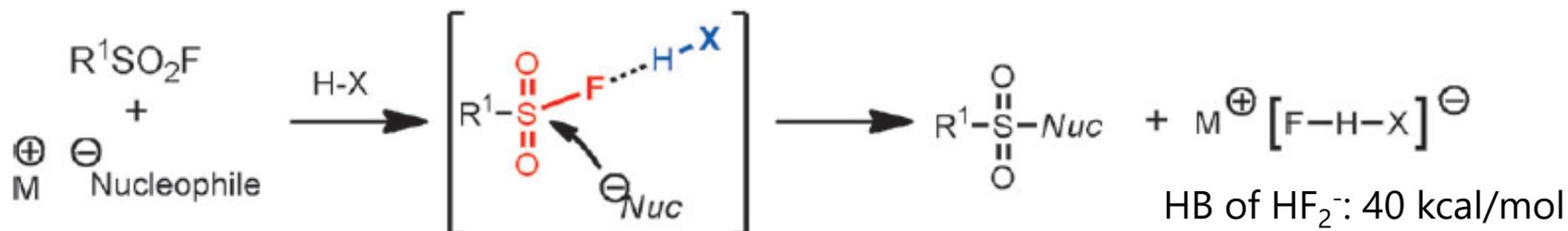
Nucleophilic attack

Activation with water

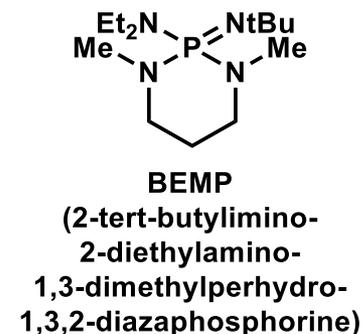
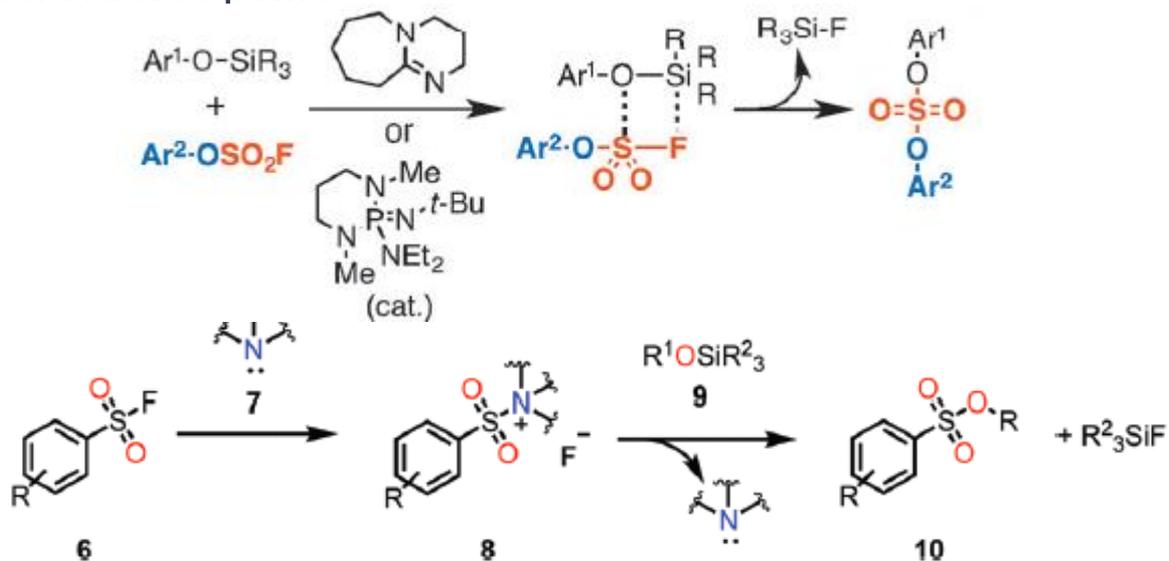
SuFEx Reaction

Activation of S(VI)-F

- Hydrogen bond donor (acid, HX)



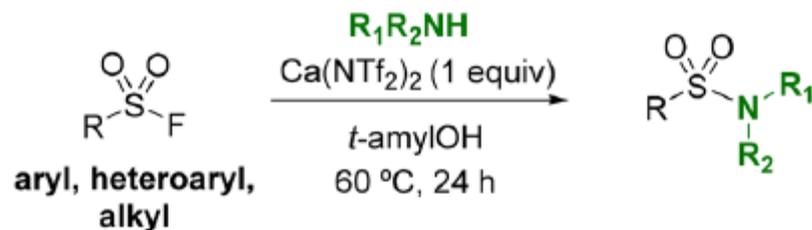
- Silylated nucleophile



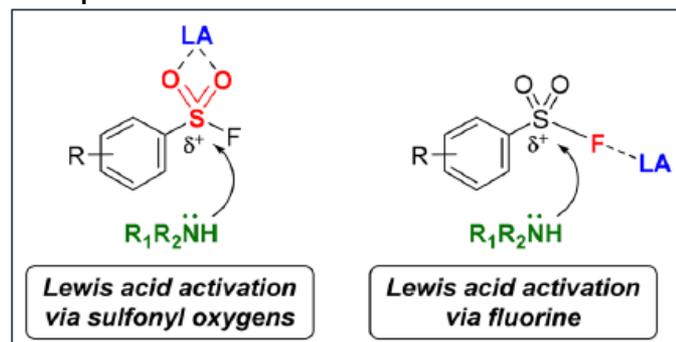
Reactivity: $\text{Et}_3\text{N} < \text{DBU} < \text{BEMP}$

SuFEx Reaction

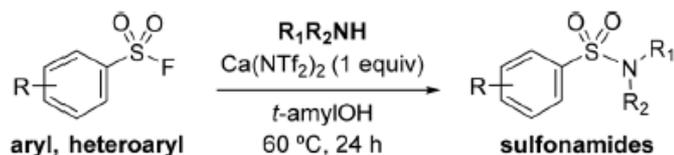
Activation with $\text{Ca}(\text{NTf}_2)_2$ (Lewis acid)



Proposed mode of activation

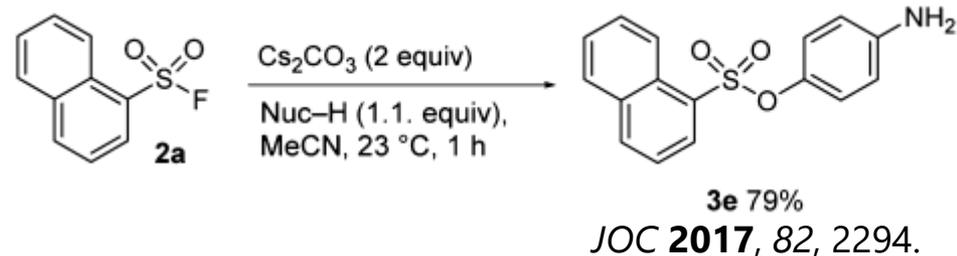
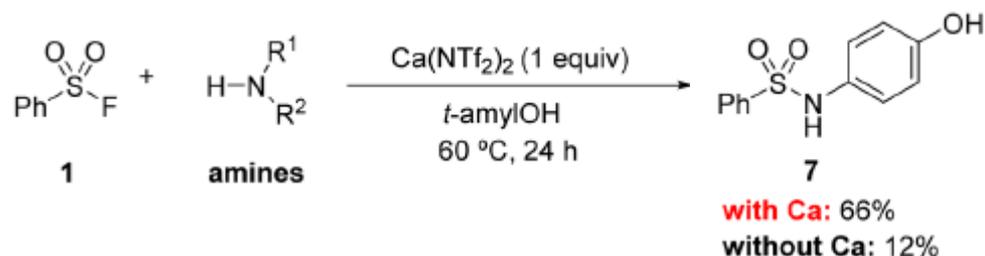


Effect of Ca salt



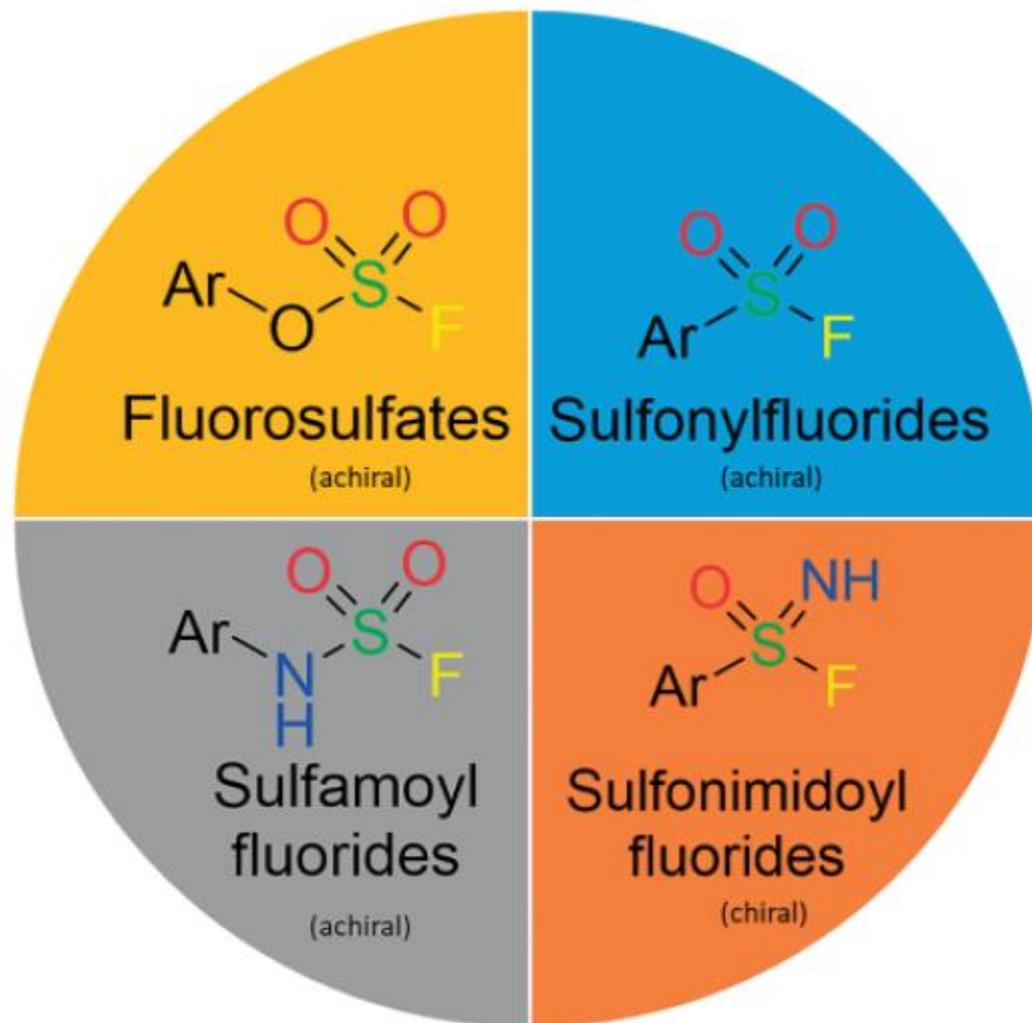
	15		2
	21 with Ca: 82% without Ca: 95%		26 with Ca: 85% without Ca: N/R ^a
	22 with Ca: 90% without Ca: 81%		27 with Ca: 88% without Ca: trace ^b
	23 with Ca: 95% without Ca: 90%		28 with Ca: 98% without Ca: trace ^b

Selectivity of nucleophile



Synthesis of SuFEx unit

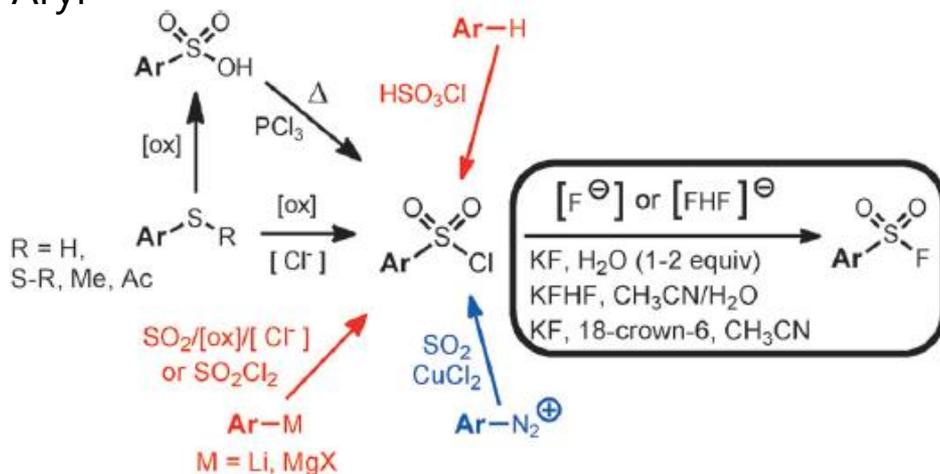
SuFEx Unit



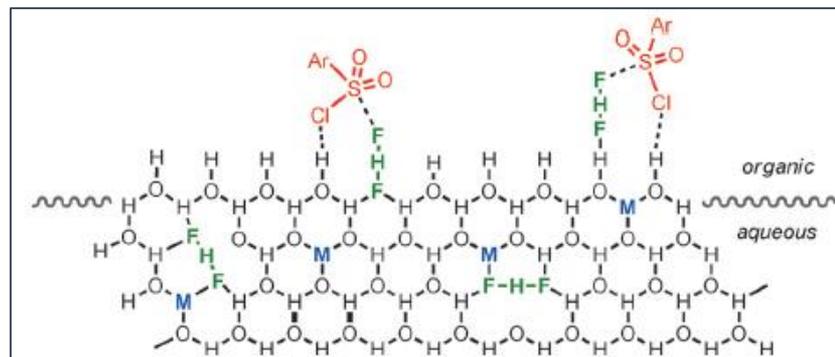
Sulfonylfluoride (RSO₂F)

Oldest conditions: via RSO₂Cl

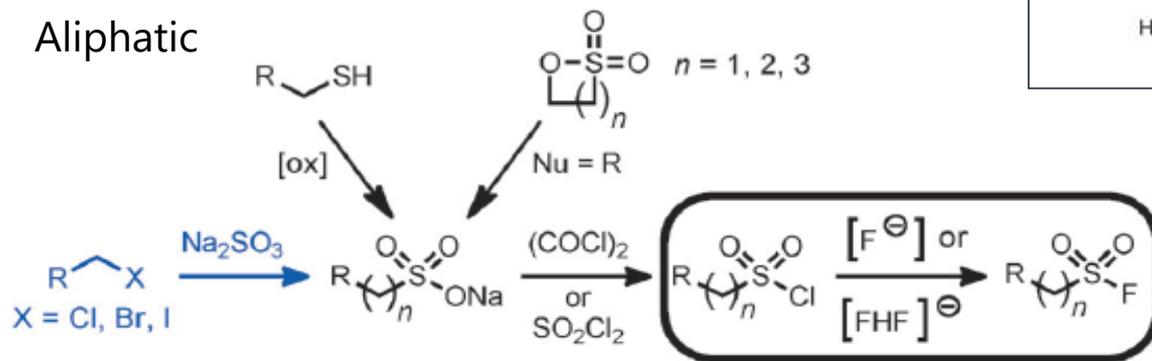
Aryl



Destabilized and nucleophilic F⁻



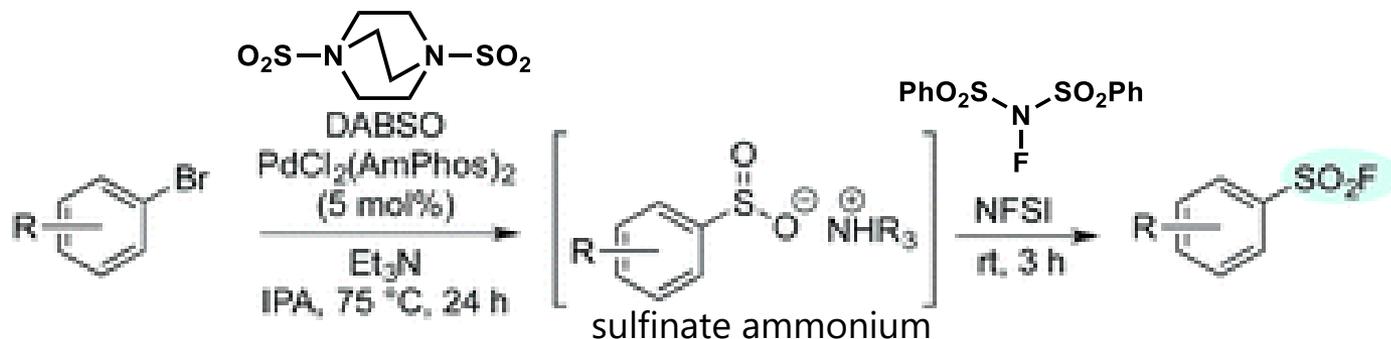
Aliphatic



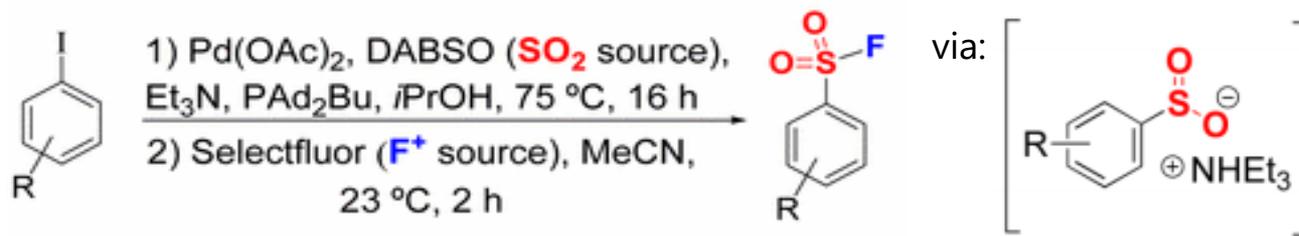
Harsh conditions

Sulfonylfluoride (RSO₂F)

Pd catalyzed SO₂ insertion

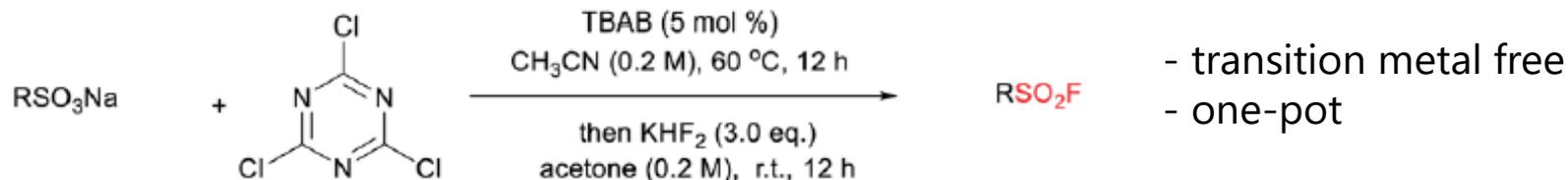


Bailev, S. W.; Willis, M. C. *et al. Chem. Sci.* **2017**, *8*, 1233.



Ball, N. D. *et al. J. Org. Chem.* **2017**, *82*, 2294.

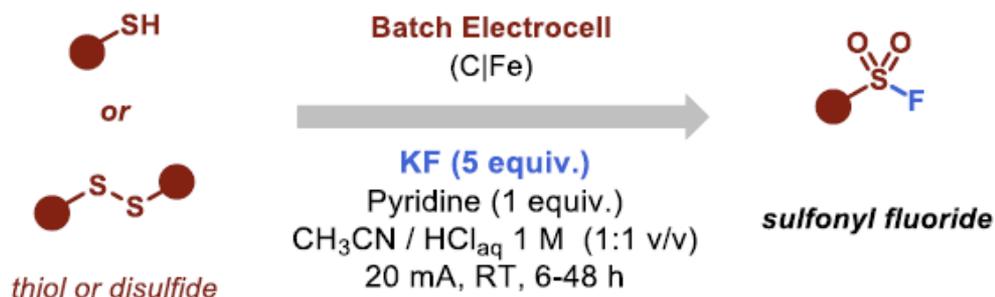
Via sulfonate salt



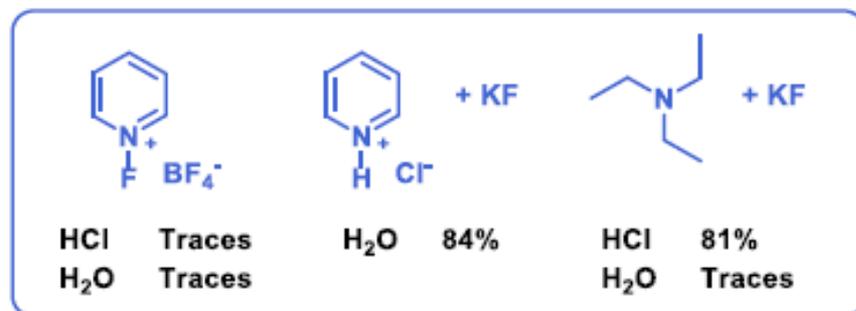
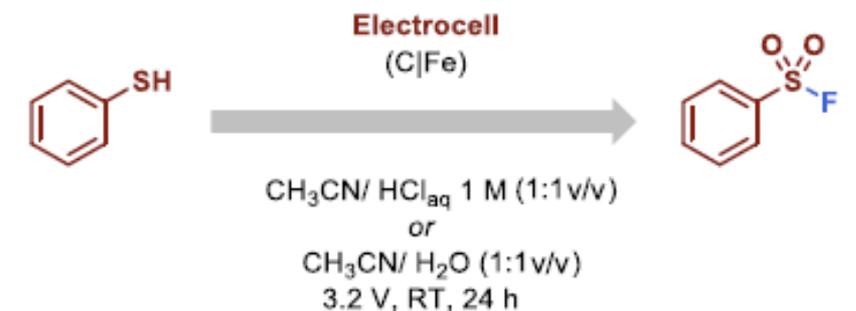
Sun, B.; Qin, H. L. *et al. RSC Adv.* **2019**, *9*, 13863. 12

Sulfonylfluoride (RSO₂F)

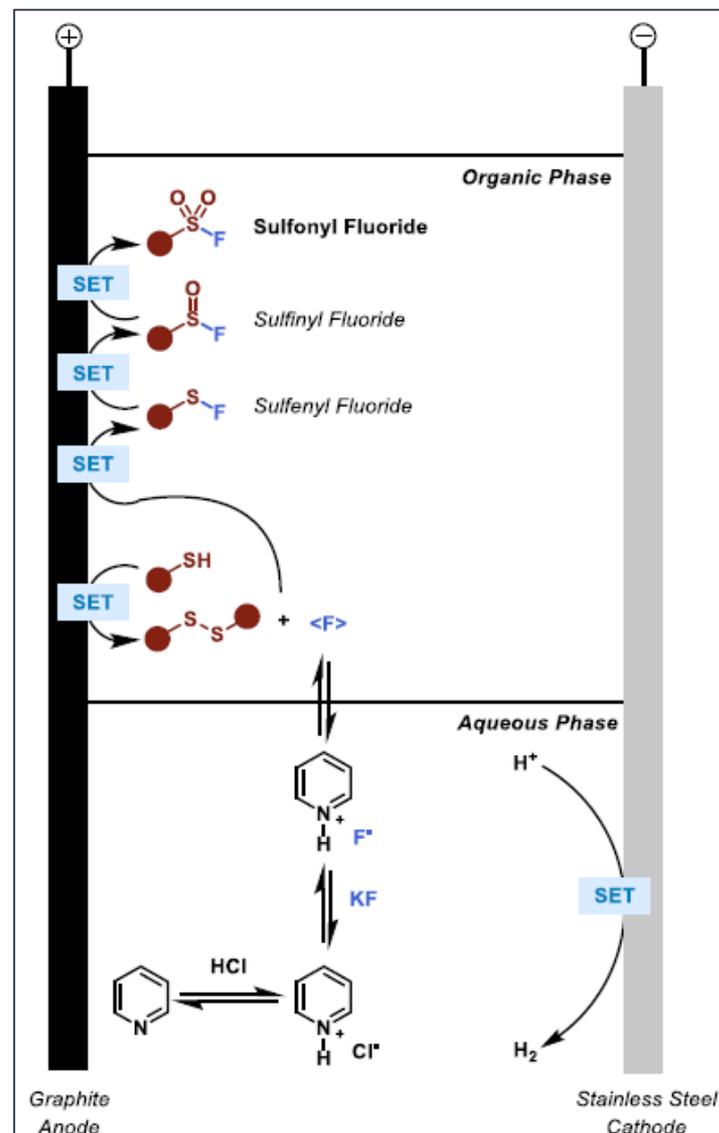
Electrochemical synthesis



Investigation of active species

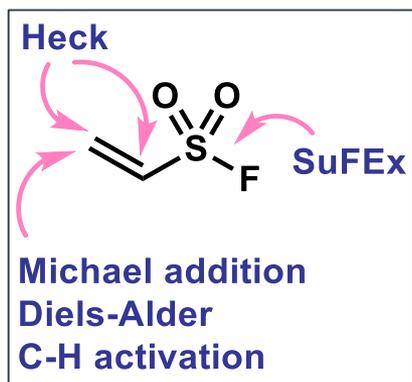


Mechanism



SuFEx Hubs

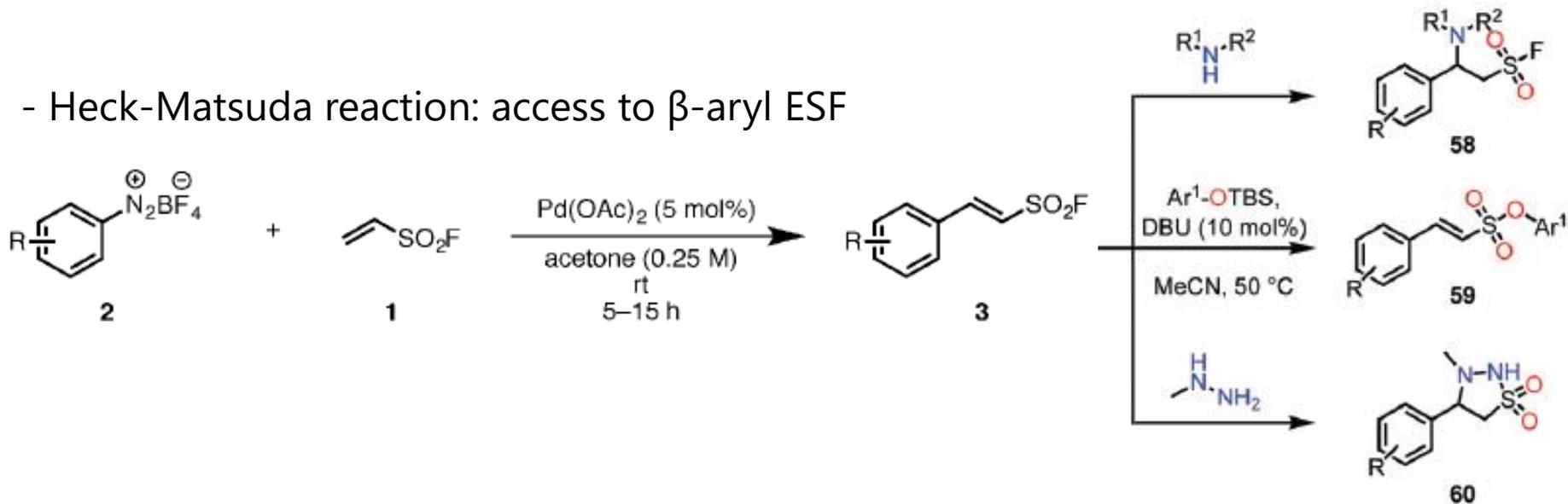
ESF: ethene sulfonyl fluoride



- Michael addition



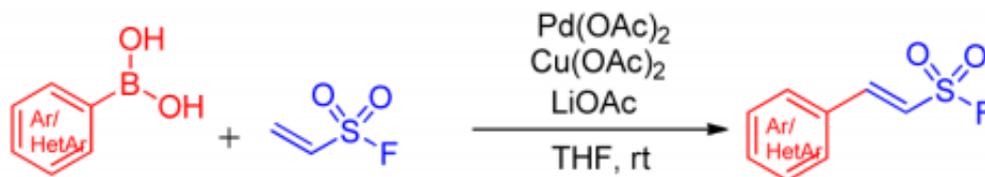
- Heck-Matsuda reaction: access to β -aryl ESF



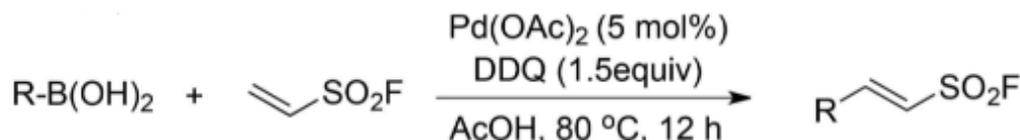
Wu, P.; Sharpless, K. B. *et al. Angew. Chem. Int. Ed.* **2016**, *55*, 14155.

SuFEx Hubs

- Oxidative Heck reaction

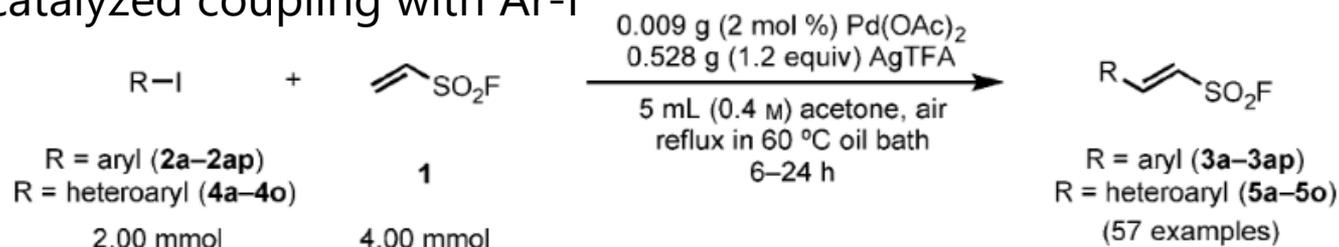


Arvidsson, P. I. *et al. Org. Lett.*, **2017**, 19, 480.



Qin, H. L. *et al. Adv. Synth. Catal.*, **2017**, 359, 3237.

- Pd catalyzed coupling with Ar-I



Qin, H. L.; Shalpley, K. B. *et al. Angew. Chem., Int. Ed.*, **2017**, 56, 4849.

- C-H activation



Org. Chem. Front., **2018**, 5, 1411.

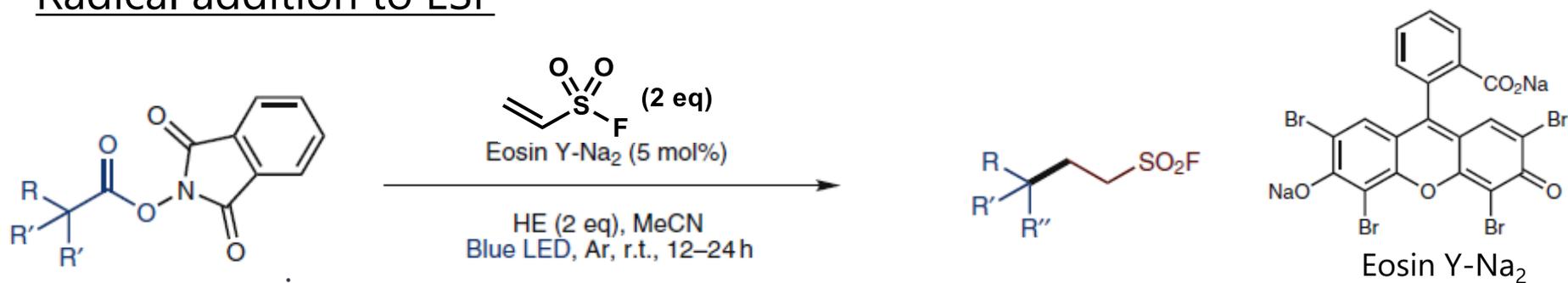
Eur. J. Org. Chem., **2018**, 4407.

Org. Lett., **2018**, 20, 4699.

Organometallics, **2019**, 38, 76.

SuFEx Hubs

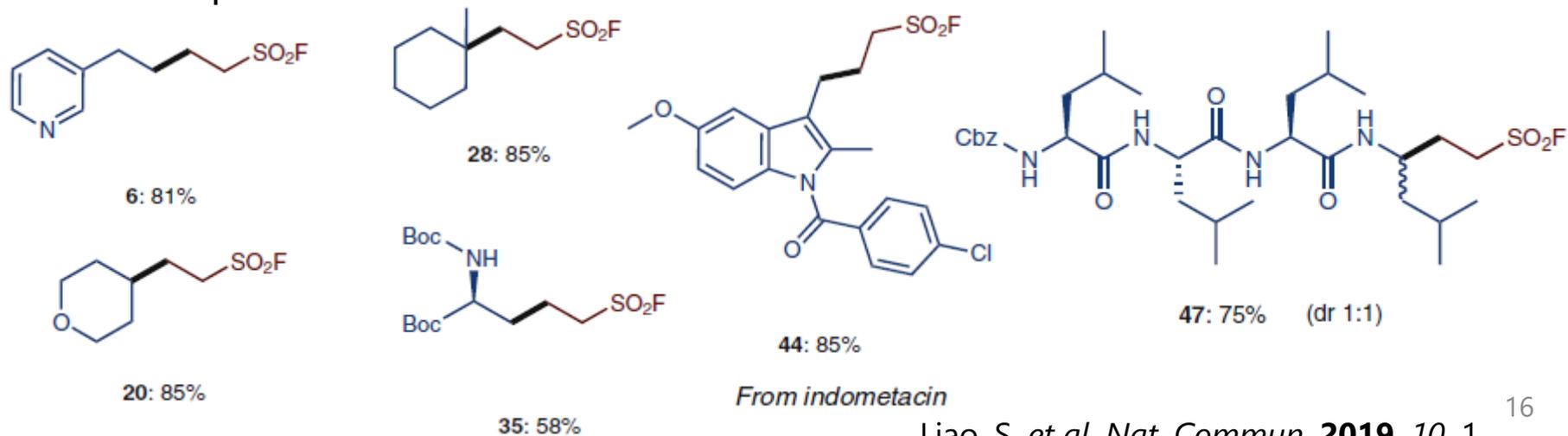
Radical addition to ESF



via:

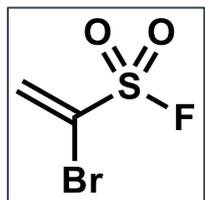


Selected scopes

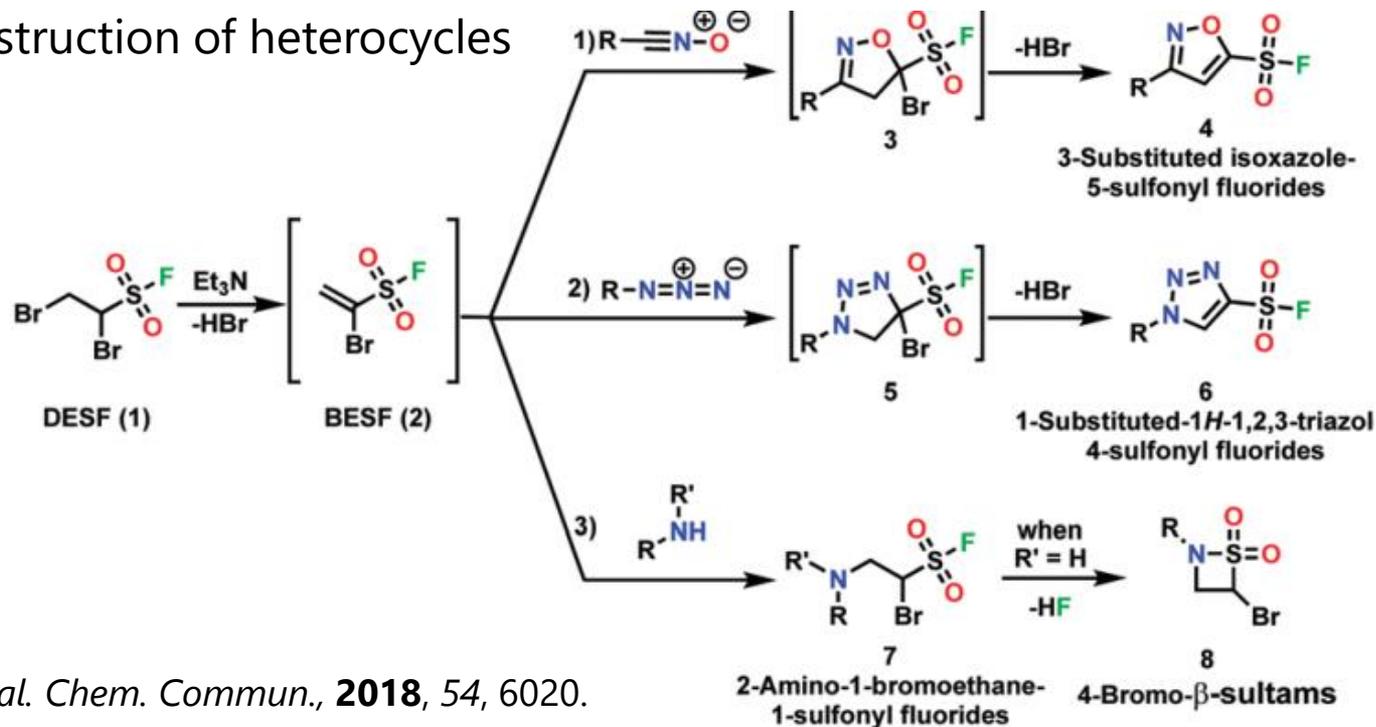


SuFEx Hubs

BESF: 1-bromoethene-1-sulfonyl fluoride

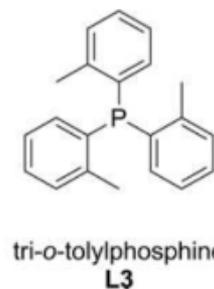
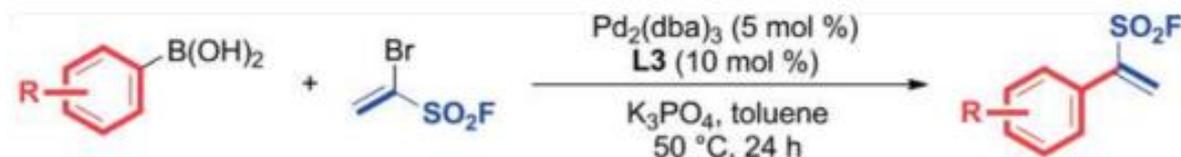


Construction of heterocycles



Moses, J. E. *et al. Chem. Commun.*, **2018**, 54, 6020.

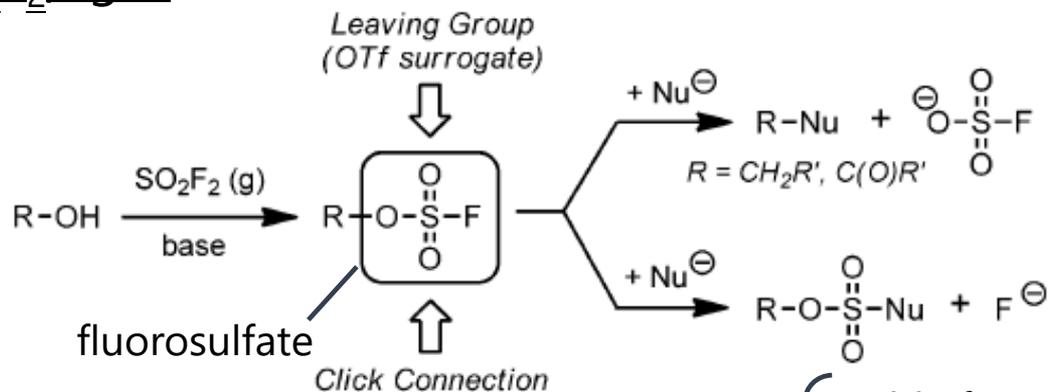
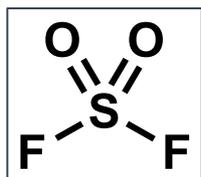
- Suzuki-Miyaura cross coupling: access to α -aryl ESF



Qin, H. L. *et al. Eur. J. Org. Chem.* **2019**, 6101.

Synthesis of Fluorosulfate (ROSO₂F)

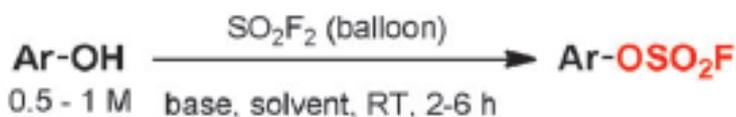
Sulfuryl fluoride (SO₂F₂): gas Reaction with alcohol/phenol



Aryl fluorosulfate is useful in terms of stability.

stable for

- months (neutral buffer)
- 2 weeks (buffer pH 10)



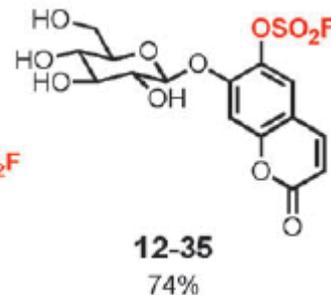
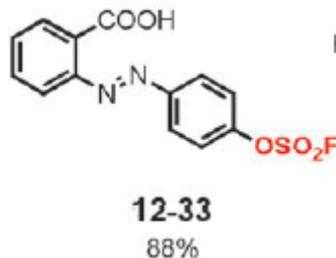
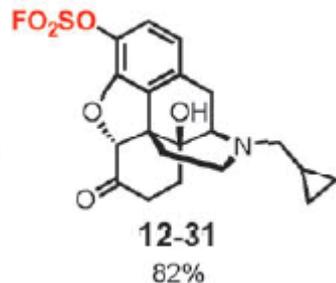
A) Et₃N in CH₂Cl₂,

:for simple molecules

B) Et₃N or *i*Pr₂NEt in biphasic mixture (CH₂Cl₂/water)

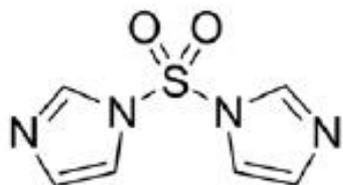
:for complex molecules

Selected scopes

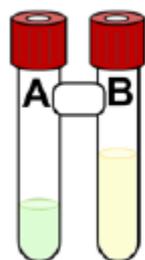


Ex Situ Generation of SO₂F₂ Gas

Precursor of SO₂F₂

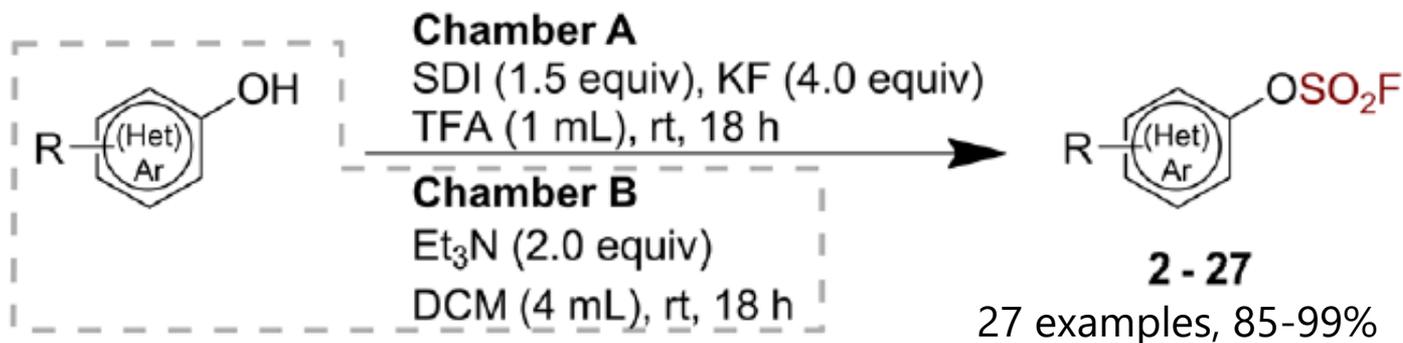


1,1'-sulfonyldiimidazole (SDI)



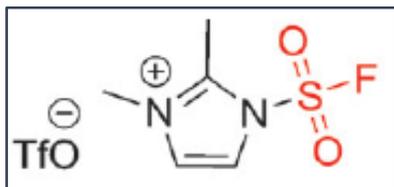
A: SO₂F₂ generation

B: (Het)ArOH functionalization

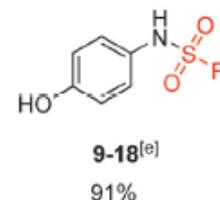
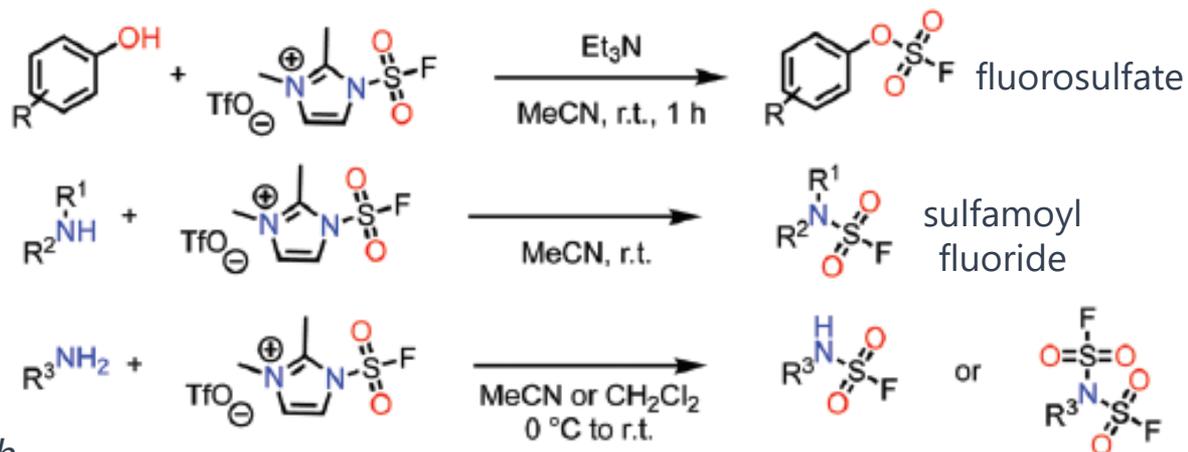


SO₂F₂ Equivalents

Fluorosulfuryl imidazolium salt



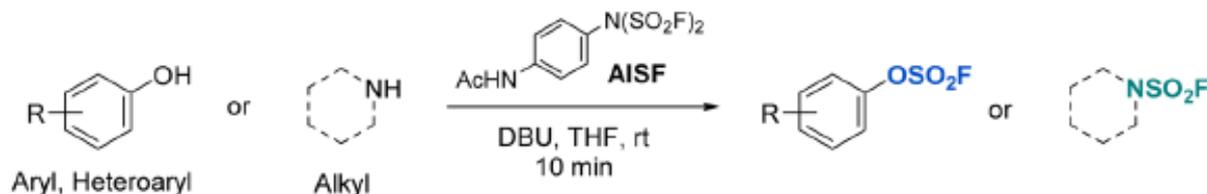
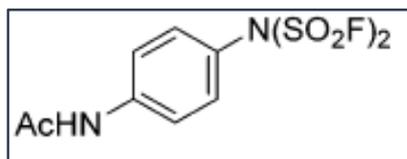
- bench-stable solid
- higher reactivity than SO₂F₂
- commercially available from Aldrich
- applicable to 1° amine



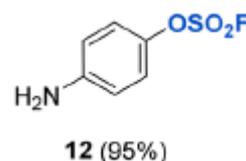
selectivity:
amine, aniline > phenol

Sharpless, K. B.; Dong, J. *et al. Angew. Chem. Int. Ed.* **2018**, 57, 2605.

AISF: [(acetylamino)phenyl]imidodisulfuryl difluoride



- no need of SO₂F₂ for preparation
- bench-stable solid
- higher reactivity than SO₂F₂
- rapid kinetics

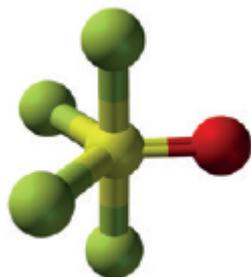
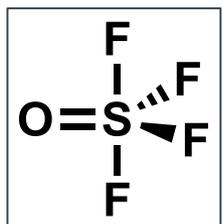


selectivity:
phenol > aniline

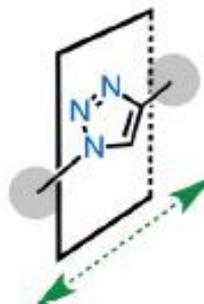
Am Ende, C. W. *et al. Org. Lett.* **2018**, 20, 812. ²⁰

Sulfonimidoyl Fluoride (R-SO₂NHF)

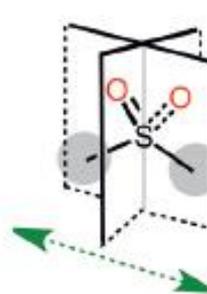
Thionyl tetrafluoride (SO₂F₂) gas



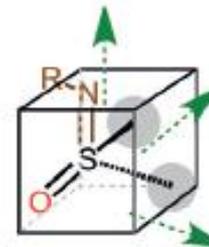
CuAAC-triazole
planar linker



SO₂F₂ derived planar
SuFEx linker



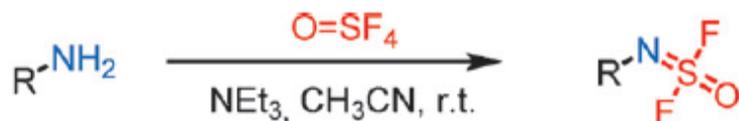
O=SF₄ derived
3D-SuFEx linker



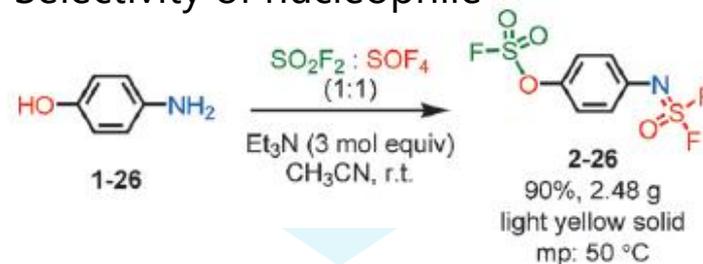
multidimensional hub

1st addition

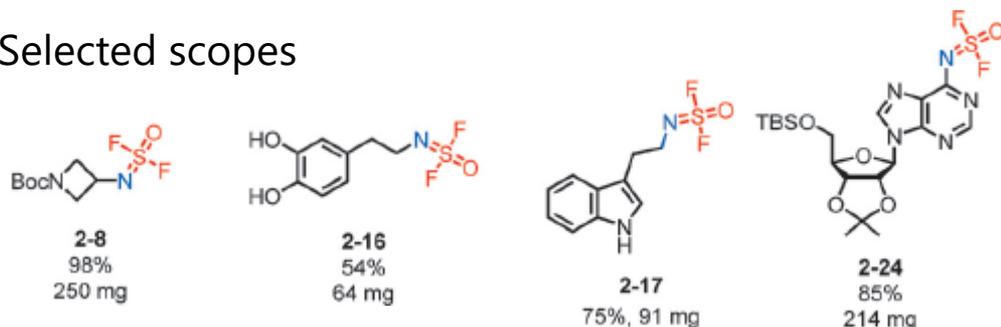
- 1° amine



Selectivity of nucleophile



Selected scopes

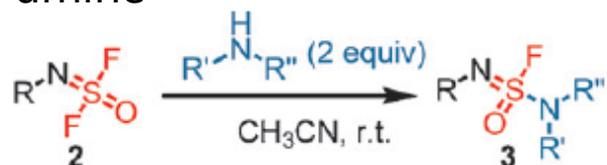


orthogonal reactivity
SO₂F₂: phenol > amine
SO₂F₄: amine > phenol

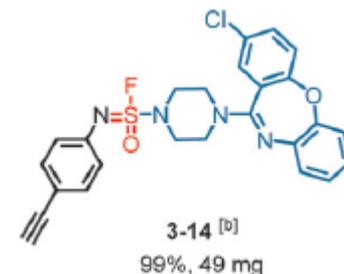
Sulfonimidoyl Fluoride (R-SO₂NHF)

2nd addition

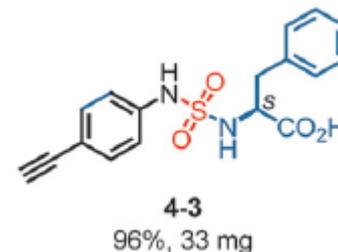
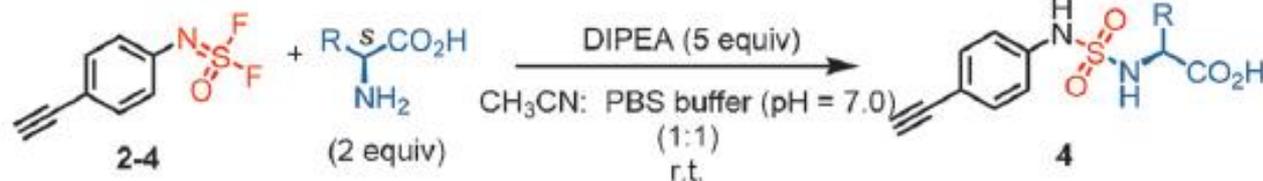
- 2° amine



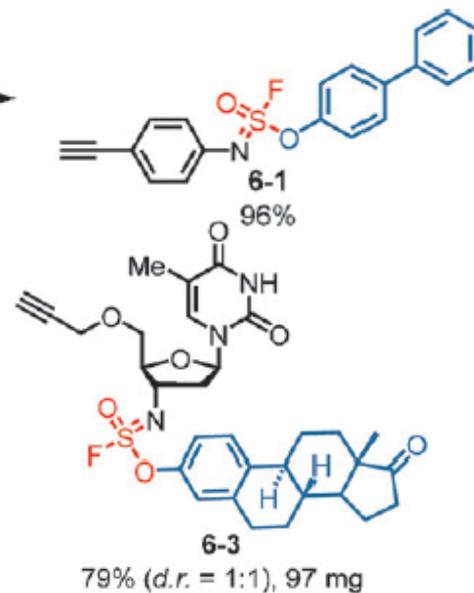
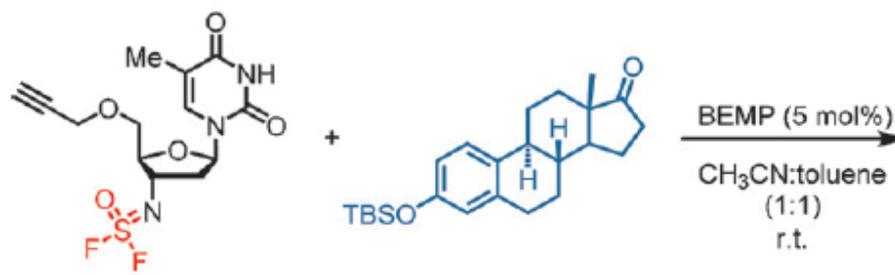
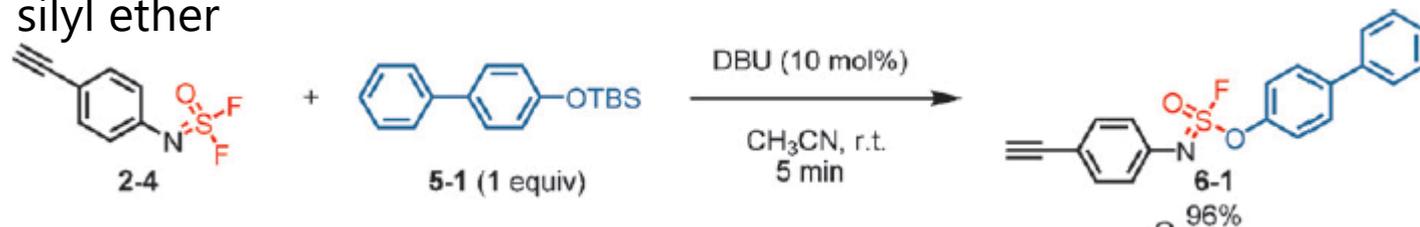
sulfonimidoyl fluoride



- Amino acid: hydrolytic loss of second S-F

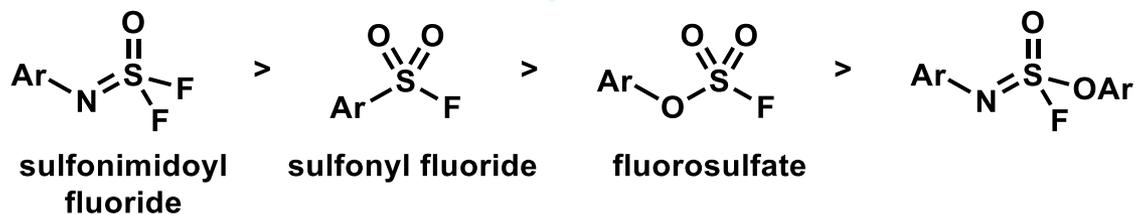
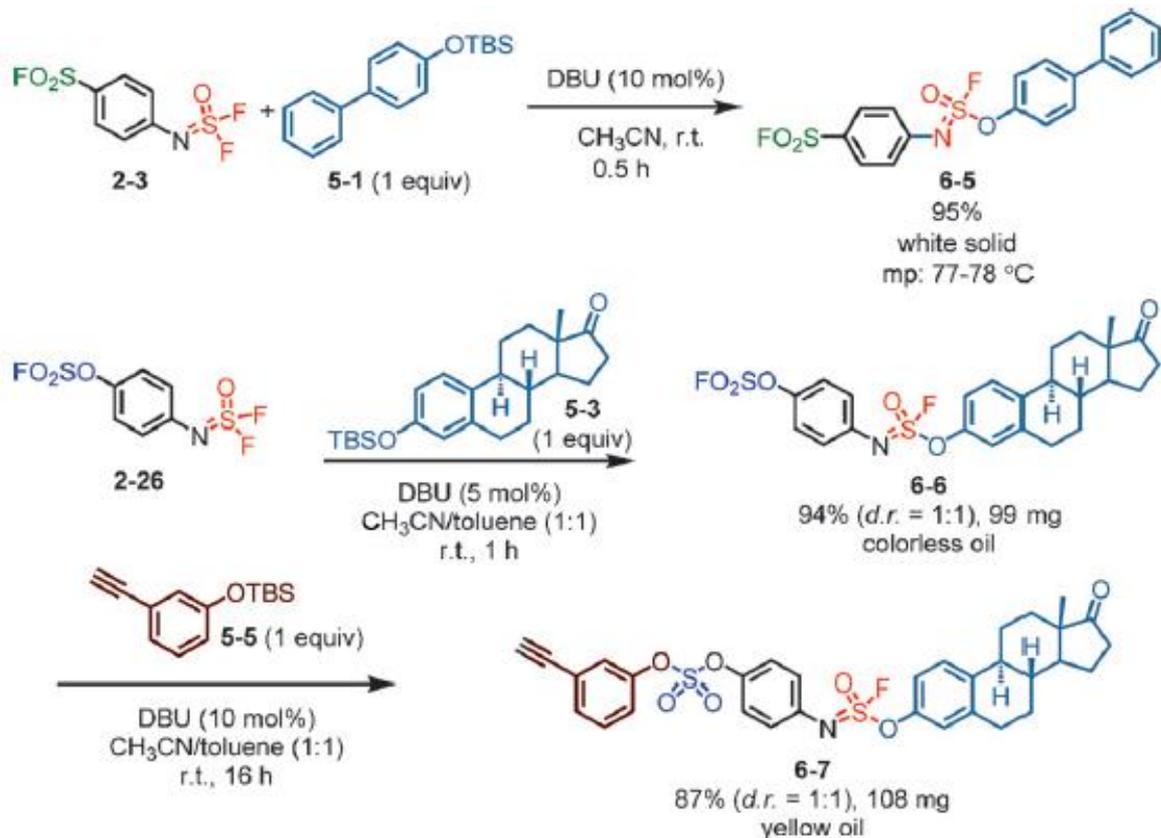


- Aryl silyl ether



Sulfonimidoyl Fluoride (R-N=S(O)F₂)

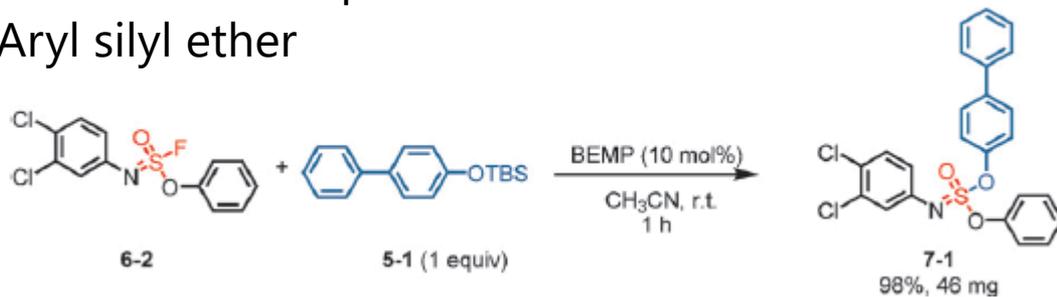
Comparison of reactivity



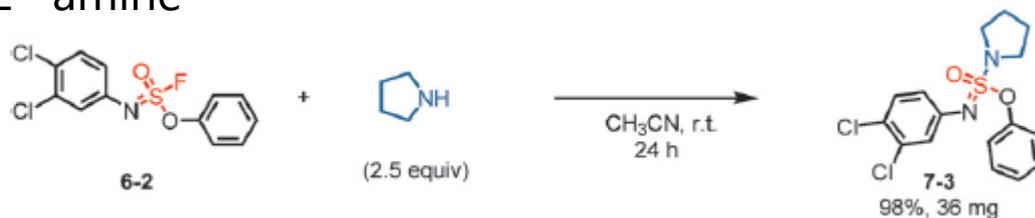
Sulfonimidoyl Fluoride (R-SO₂NHF)

3rd addition BEMP as potent base to activate inert S-F bond

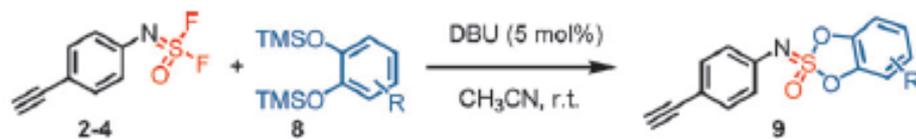
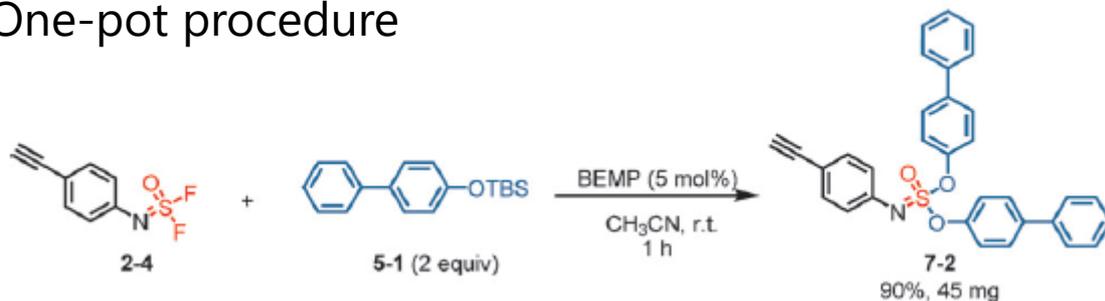
- Aryl silyl ether



- 2° amine



- One-pot procedure



Application

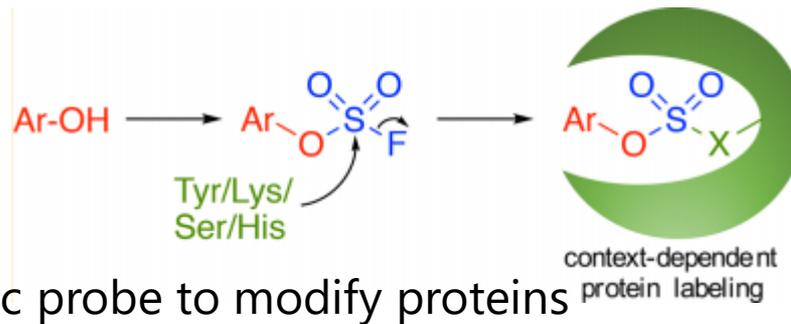
Medicinal chemistry

Chemical biology

New synthetic methodology

Material science

RSO₂F in Biological Context



H⁺ mediated switch-on

-> specificity to microenvironment

- Activation of S-F in special environment
- Proximity driven reaction
- Ser, Tyr, Lys, Thr, His, Cys

Examples

- Covalent drugs
- Target identification and validation
- Mapping of enzyme binding sites, substrates and protein–protein interactions.
- Late-stage functionalization (LSF) of bioactive molecules

Inertness of Sulfonyl Fluoride

Nucleophilic stability of electrophiles in chemical biology

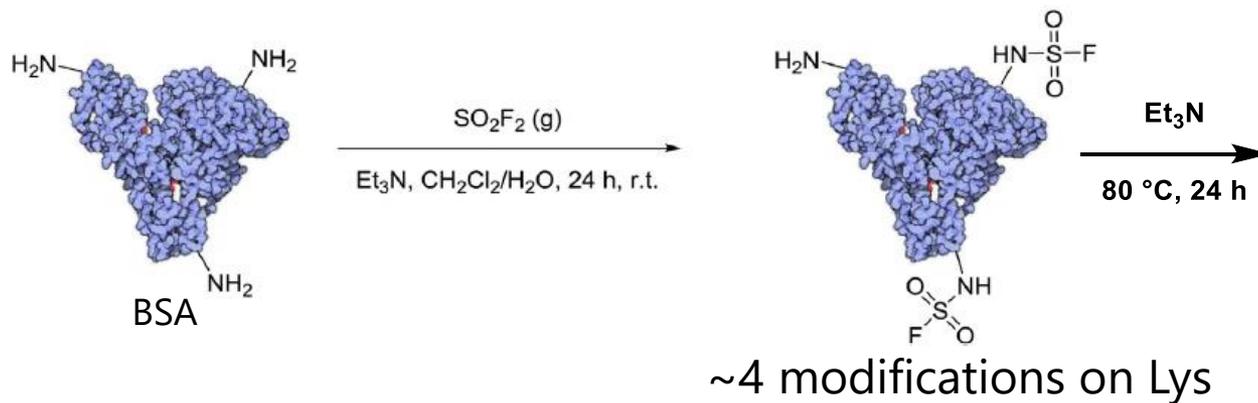
Entry	Electrophile	% remaining SM	Product (% HPLC yield)
1	(ii) Sulfonyl fluoride	100	N/A
2	(vii) Epoxide [(±)-styrene oxide]	0	Unidentified mixture
3	(viii) Acrylamide	59	(41)
4	(ix) Vinyl sulfone	0	(100)
5	(x) α -chloroamide	0	(>95)
6	(xi) β -lactam	0	Unidentified mixture

Entry	Electrophile	% remaining SM	Product (% HPLC yield)
7	(xii) maleimide	0	(27) (73)
8	(xiii) α -chloroketone	0	Unidentified mixture
9	(xiv) fluorophosphate	11	(55) (34)

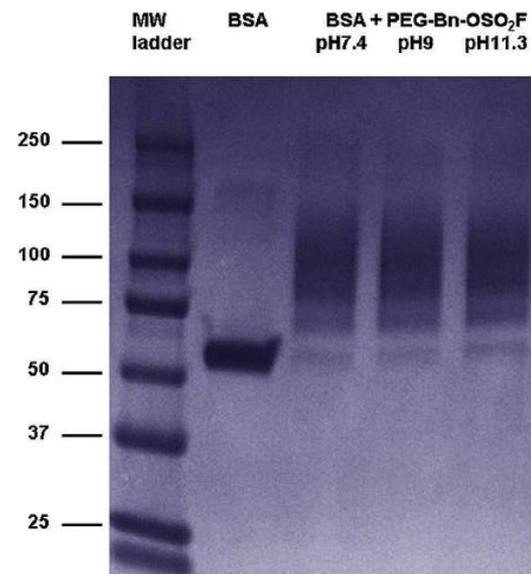
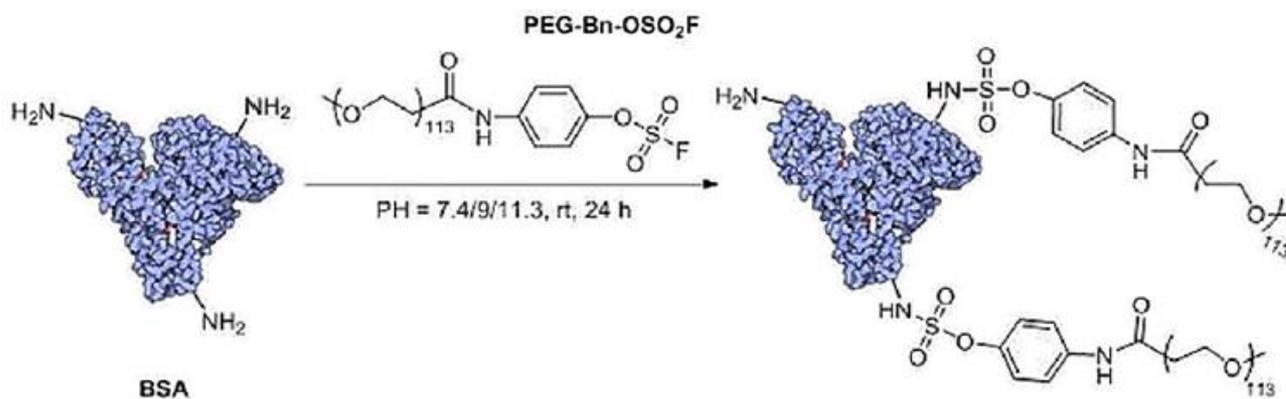
Conditions: electrophiles (1.0 mmol), aniline 1.3 mL (neat), 184 °C

Application to Bioconjugation

Modification of BSA with SO_2F_2 gas

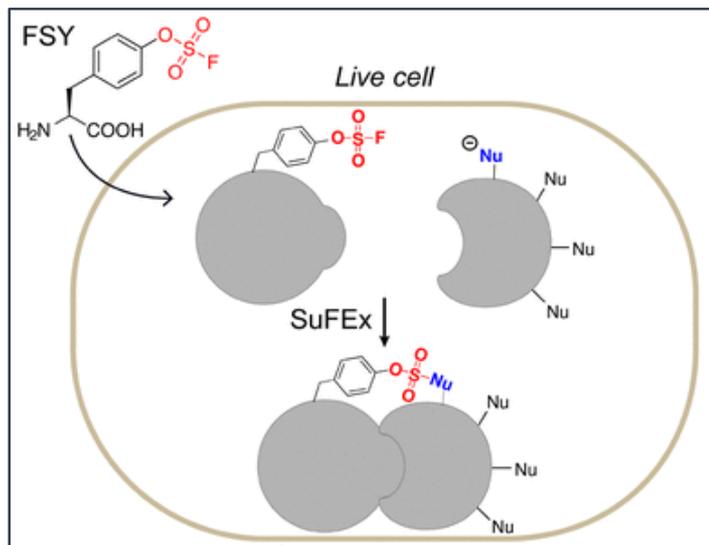


Modification of BSA with fluorosulfate



SuFEx with Protein *in Vivo*

Fluorosulfate-L-tyrosine (FSY)



Incorporation of FSY

Pyrrolysyl-tRNA synthetase (PylRS) mutated library
(mutation at A302, L305, Y306, L309, I322, N346,
C348, Y384, V401, W417)

selection

6 hits PylRS mutant specific for FSY
(mutation 302I, 346T, 348I, 384L, 417K)
-> FSYRS (fluorosulfate tyrosyl-tRNA synthetase)

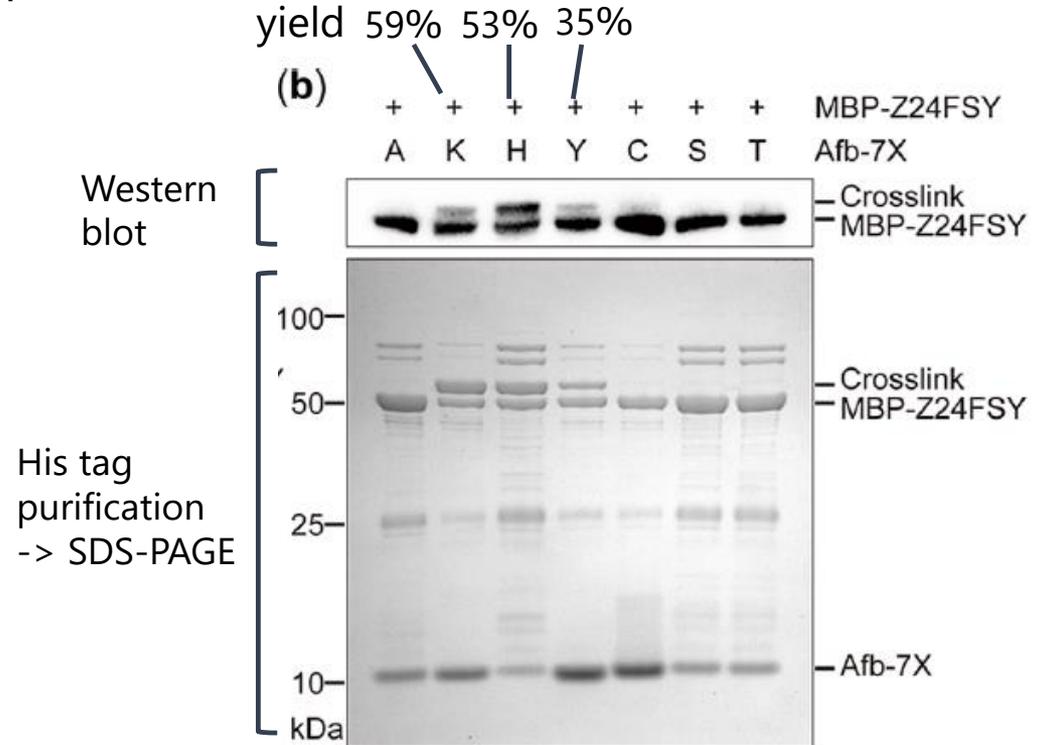
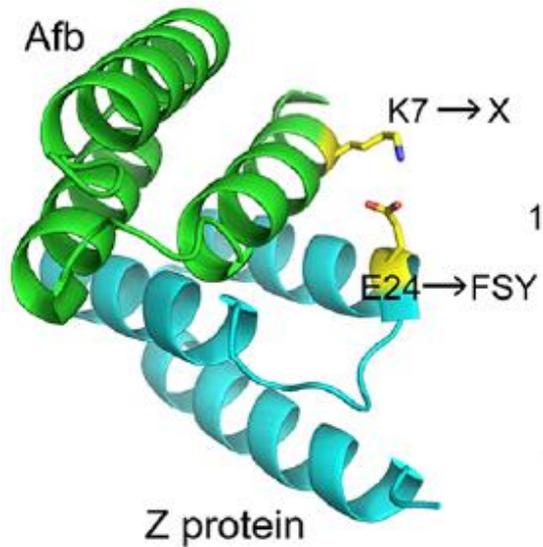
Coexpression of target protein gene
containing TAG codon with tRNA^{Pyl}_{CUA}/FSYRS

- FSY was successfully introduced in the protein of *E. coli* and HELA
- No toxicity to cells during the expression of FSY incorporated proteins.

SuFEx with Protein *in Vivo*

Afb-Z protein

MBP-Z24FSY, Afb-7X were coexpressed in *E. coli*.

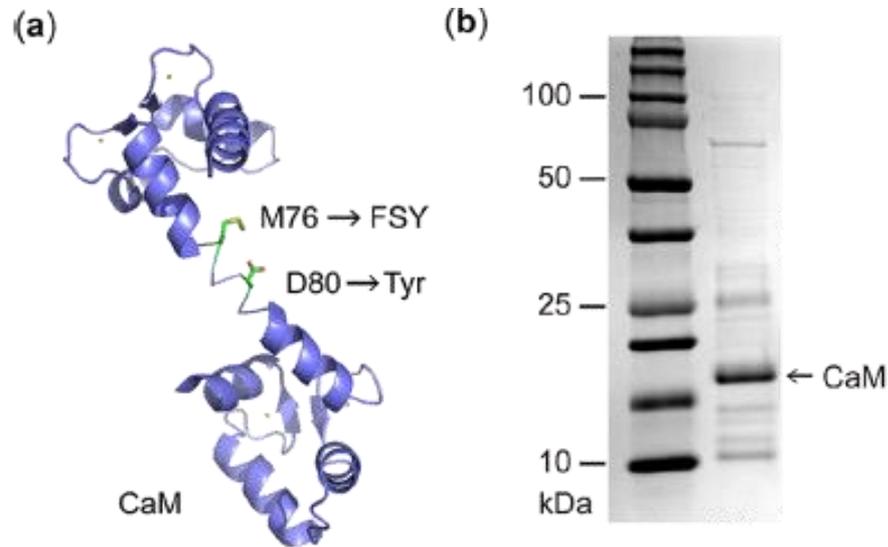


LC-MS/MS

Crosslinking between FSY and Lys/His/Thr

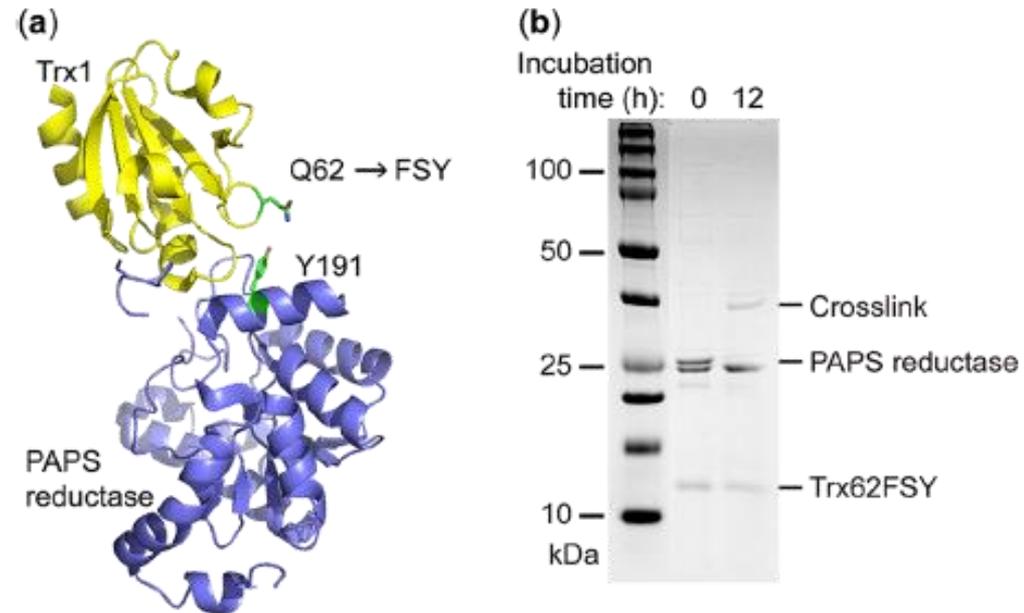
SuFEx with Protein *in Vivo*

Calmodulin



intramolecular crosslinking

Trx1-PAPS reductase



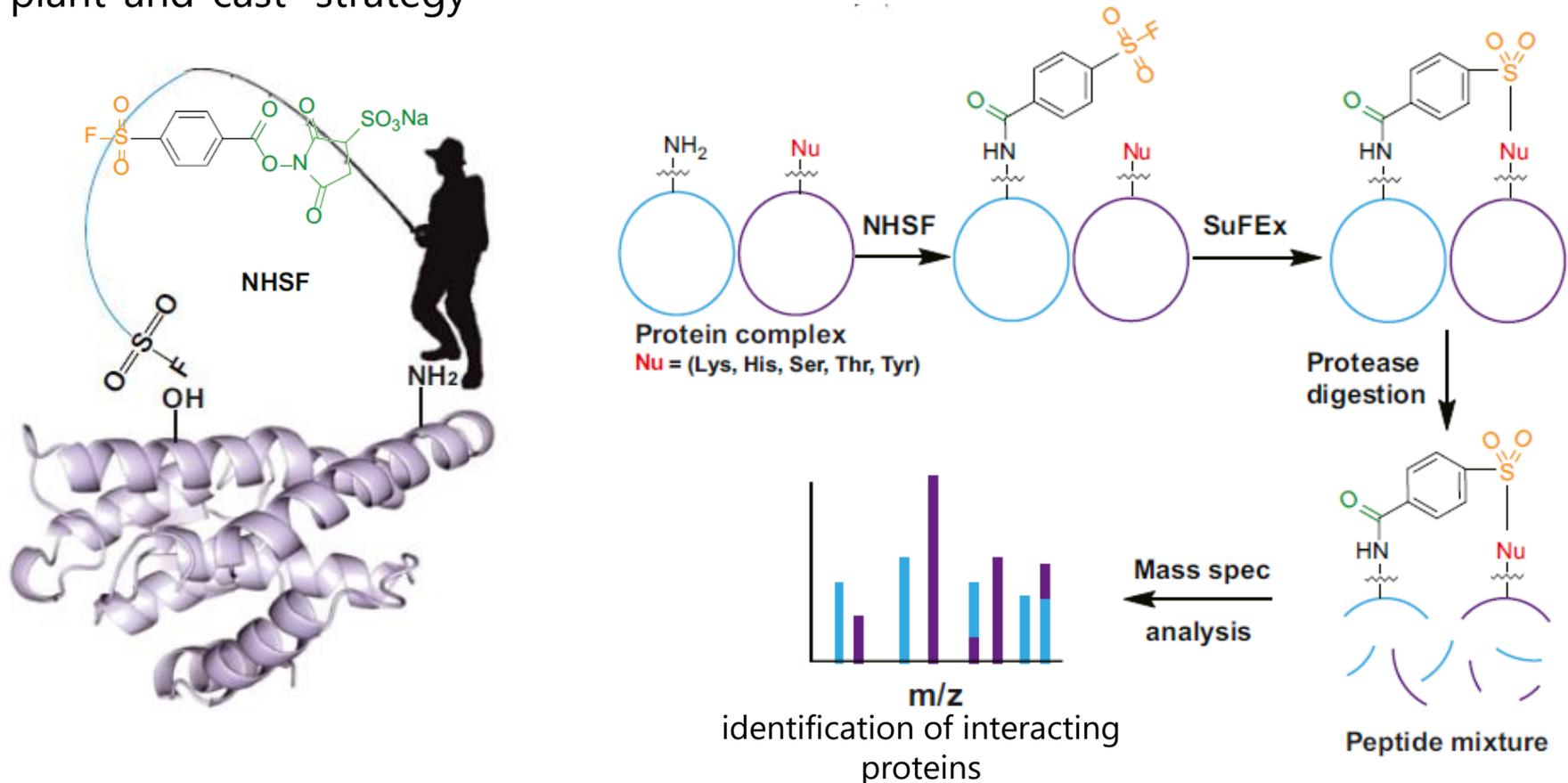
crosslinking with native Tyr

SuFEx Cross-linking for Mass Spectrometry

Cross-linking mass spectrometry (CXMS)

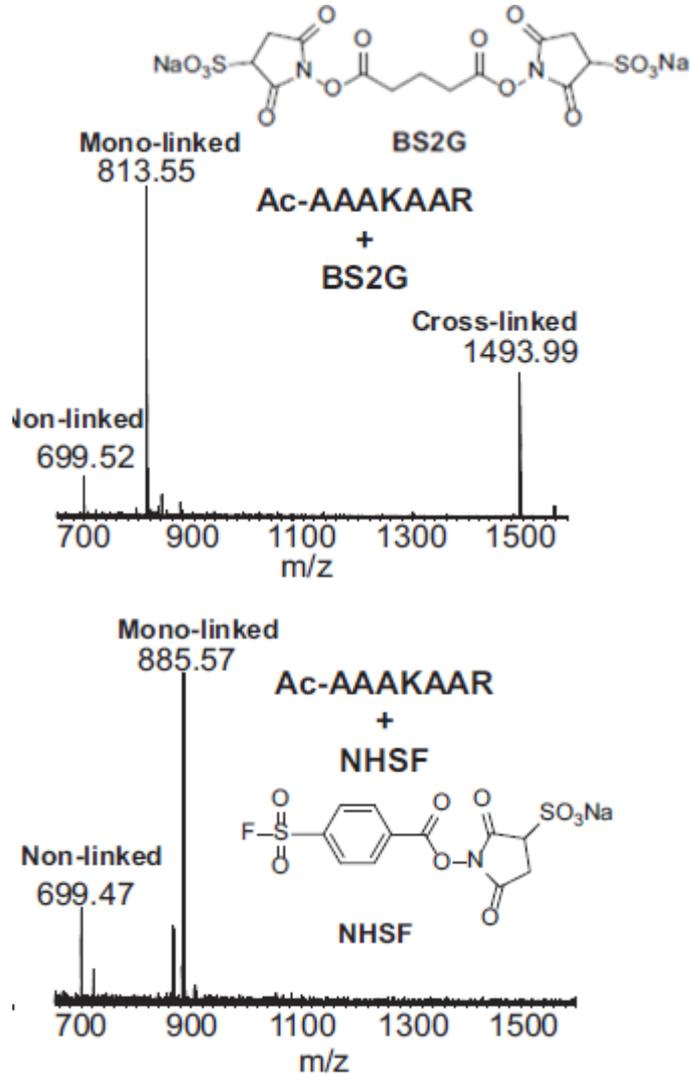
Identification of protein interaction partners

“plant-and-cast” strategy

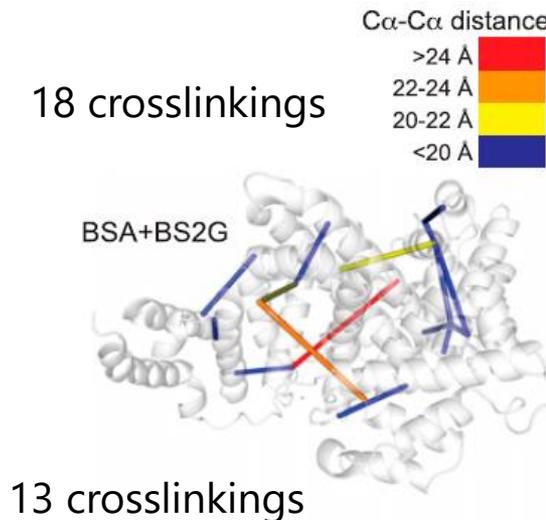


SuFEx Cross-linking for Mass Spectrometry

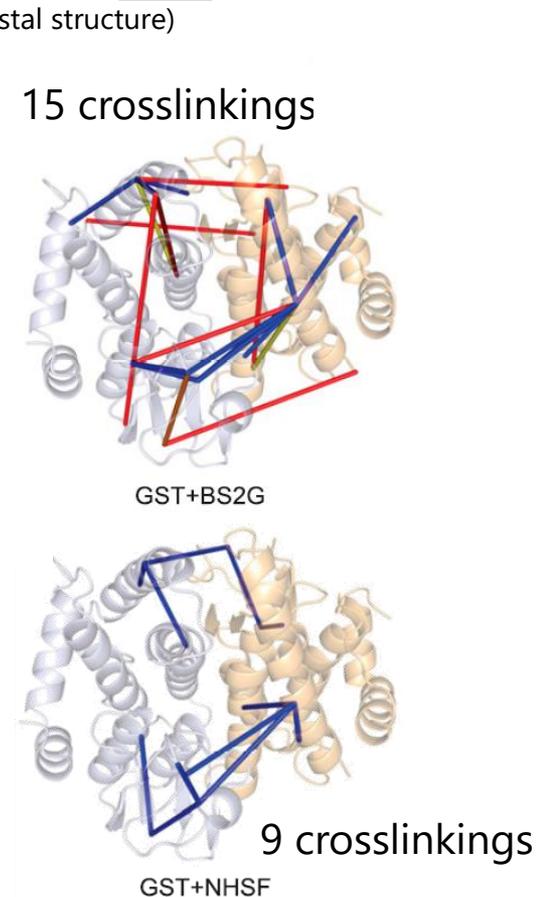
Reaction with peptide



CXMS analysis of BSA



GST



$\text{C}\alpha$ - $\text{C}\alpha$ distance (in the crystal structure)
 $>$ side chain length*2 + linker length ~ 20 Å
 \neq Cross-linking occurred at different structure from crystal structure.

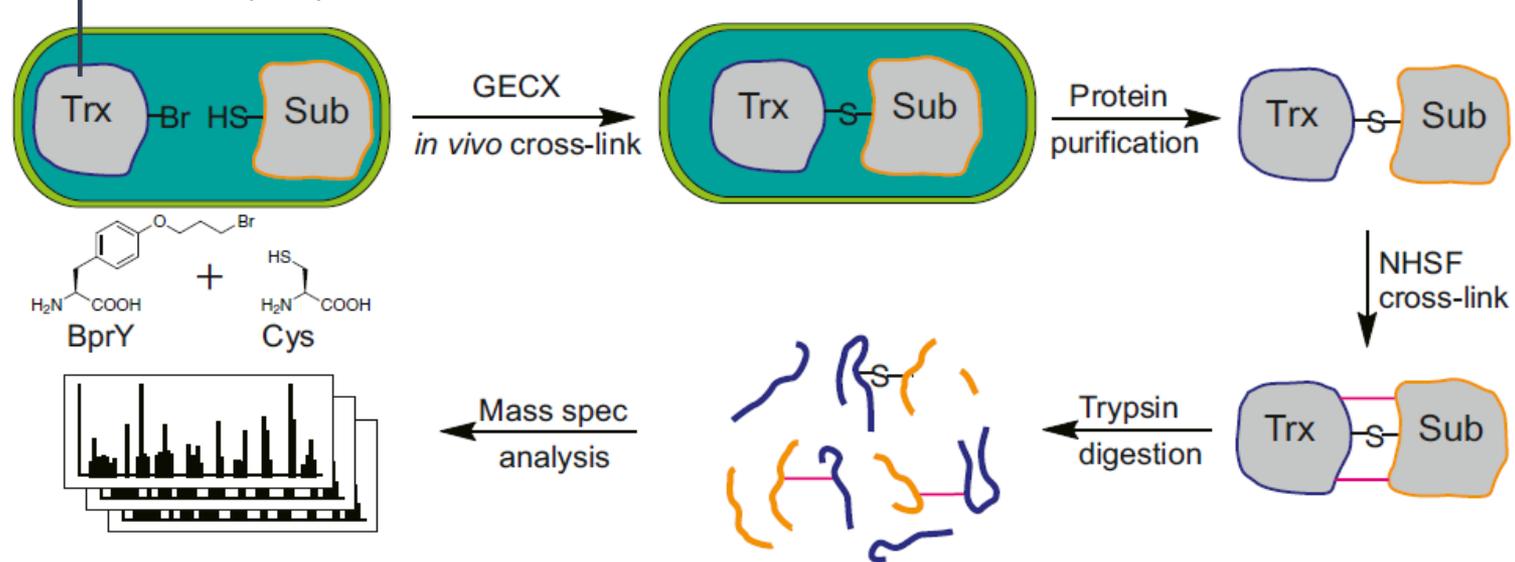
Non-specific cross-linking is observed with BS2G.

SuFEx Cross-linking for Mass Spectrometry

Analysis of Enzyme-Substrate Interaction

genetically encode chemical crosslinkers (GECX)

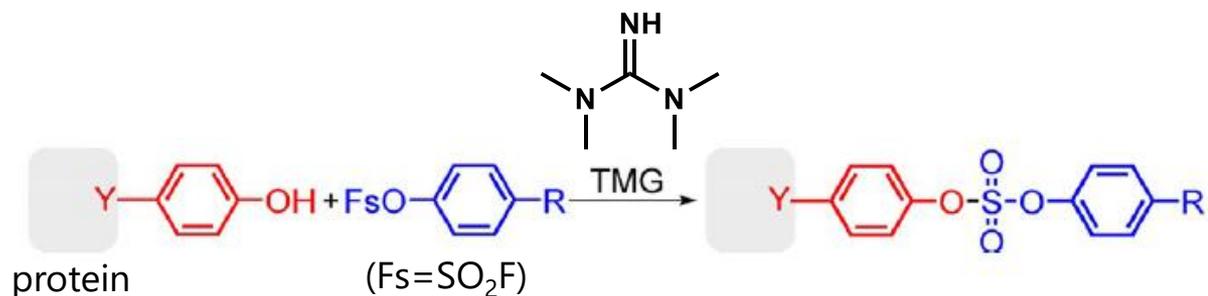
thioredoxin (Trx)



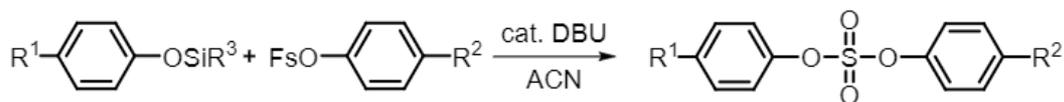
Additionally identified Trx-interacting peptides

Accession #	Gene	Description	Cross-linked peptides (cross-linked sites in red)
P43319	yraK	Uncharacterized fimbrial-like protein	MSDKIIHLTDDSFDTDVLK - GAWVDGMGSPSTGDFHDLVK
P36566	cmoM	tRNA 5-carboxymethoxyuridine methyltransferase	MSDKIIHLTDDSFDTDVLK - YCRQEPYITLGR
P16918	rhcC	May involved in natural ecology of cell	EAGEFSGEITGVTDGAGRHR - SDKIIHLTDDSFDTDVLK
C4ZUA4	menC	o-succinylbenzoate synthase	GIPTLLLFKNGEVAATK - EKVAK
P55138	ygcE	Uncharacterized sugar kinase YgcE	GIPTLLLFKNGEVAATK - GYGIR
P0AGA2	secY	Protein translocase subunit	GIPTLLLFKNGEVAATK - KANLK
P0AGC3	slt	Soluble lytic murein transglycosylase	NGEVAATK - DLFKR

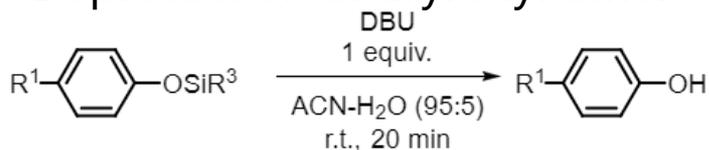
Tyr Bioconjugation



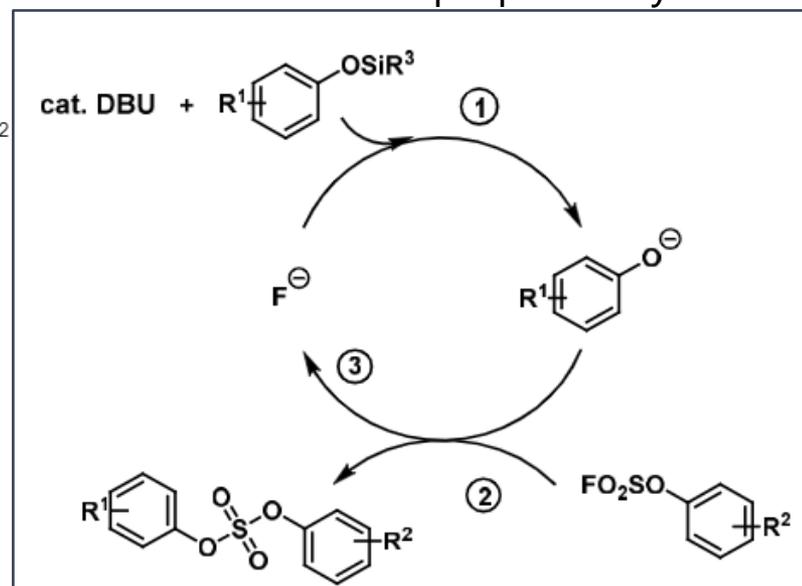
SuFEx



Deprotection of aryl silyl ether



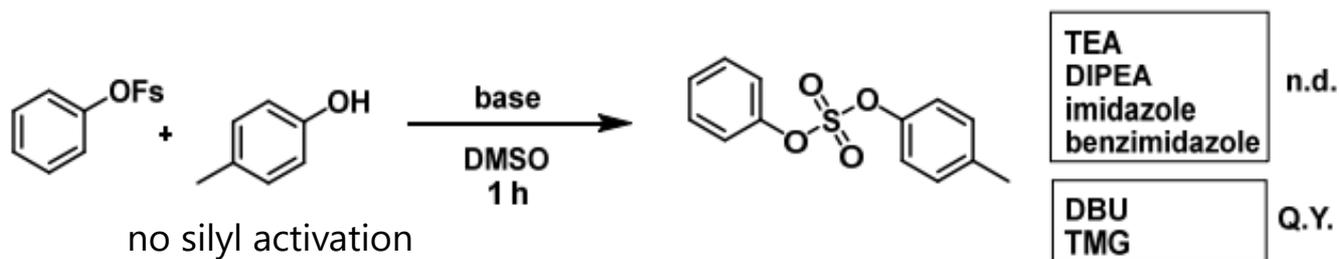
Mechanism of SuFEx proposed by authors



Use of stoichiometric amount of base avoids the silyl ether?

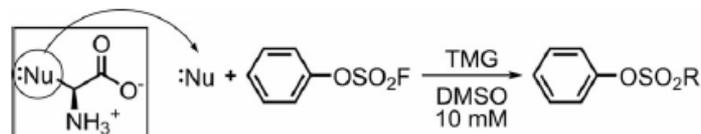
Tyr Bioconjugation

Screening of base



Residue selectivity

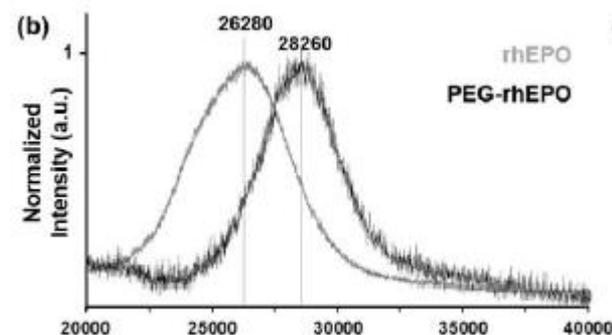
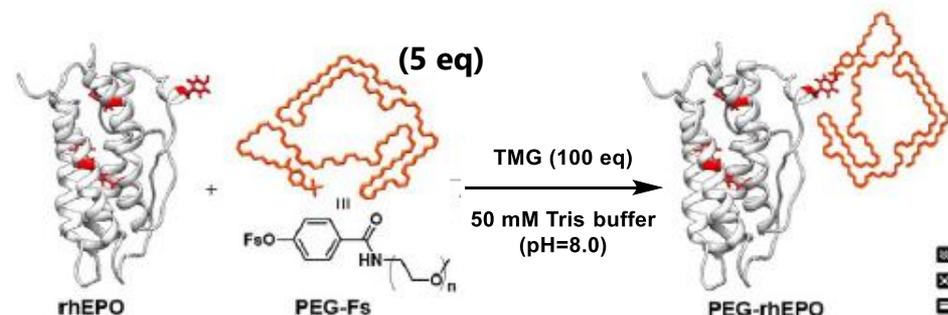
Table 1. Comparison of SuFEx reactivity in the reactions of various model nucleophiles representing amino acids with phenyl fosylate.



Entry	Model nucleophile	Amino acid	Time (h)	Yield ^[a] (%)
1	<i>p</i> -cresol	Y	1.5	93.5
2	<i>n</i> -butylamine	K	12	n.d.
3	Propanethiol	C	12	n.d.
4	Methanol	S	12	n.d.
5	<i>N</i> -propylguanidine	R	12	n.d.
6	3-methylindole	W	12	12.5
7 ^[b]	4-methylimidazole	H	12	trace

[a] Yield of isolated product. [b] 0.25 equiv of NiCl₂(H₂O)₆ was added.

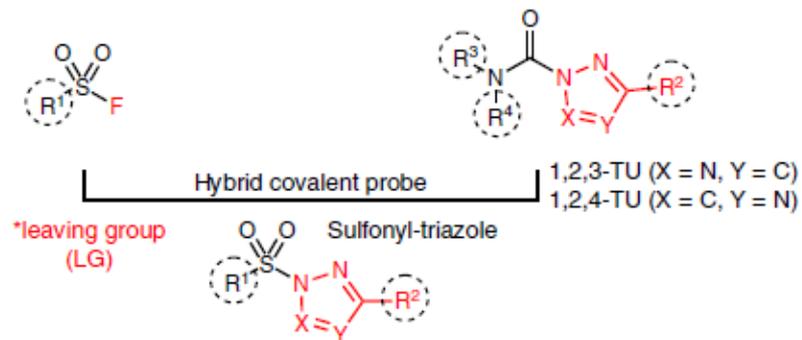
Recombinant human erythropoietin (rhEPO)



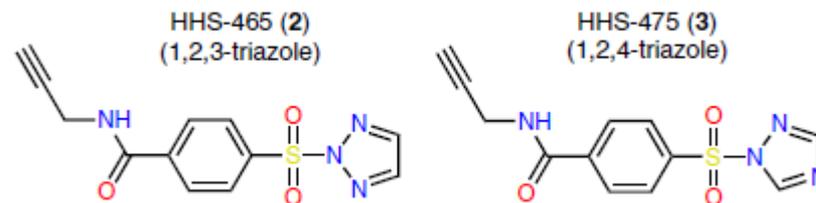
PEG
M_n=2372
Δm/z=1980

Tyr-Reactive Probe

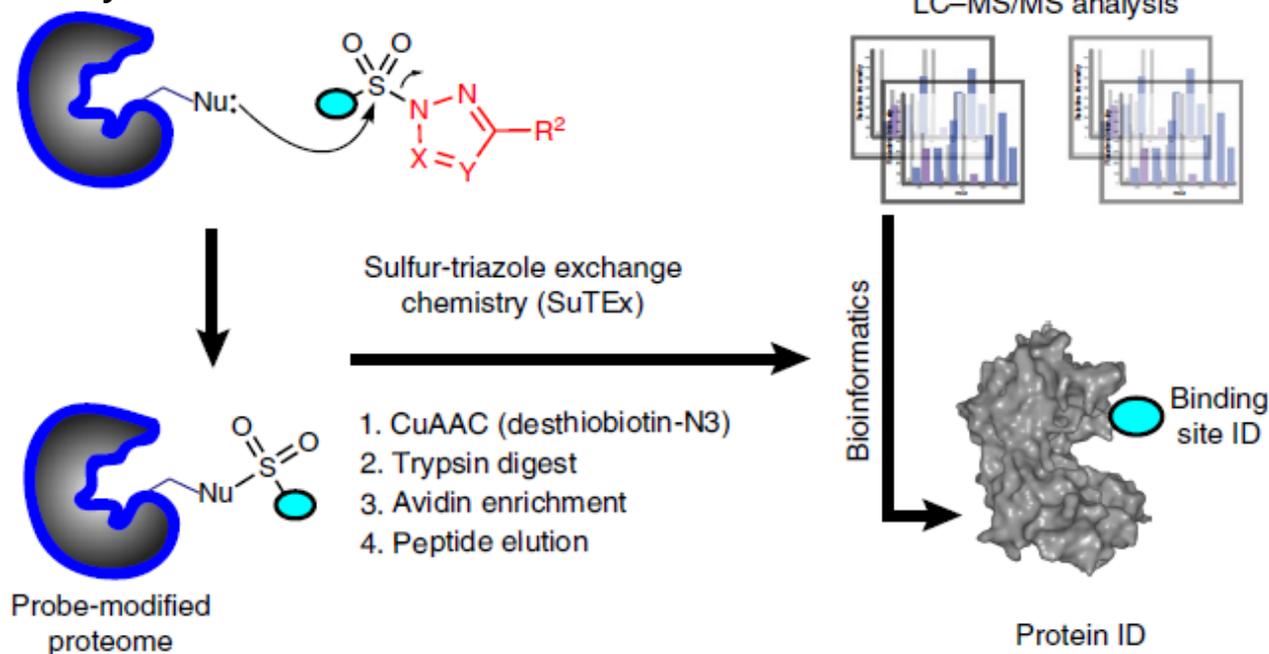
Sulfur-Triazole EXchange (SuTEx)



Probes

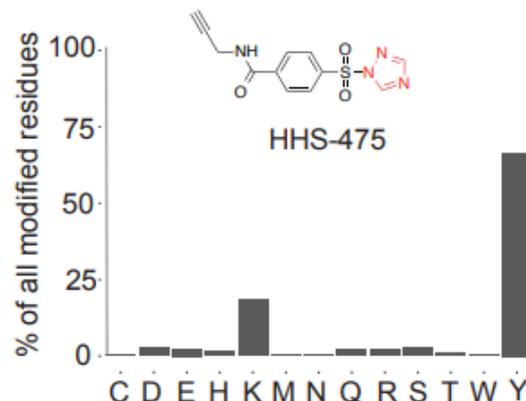
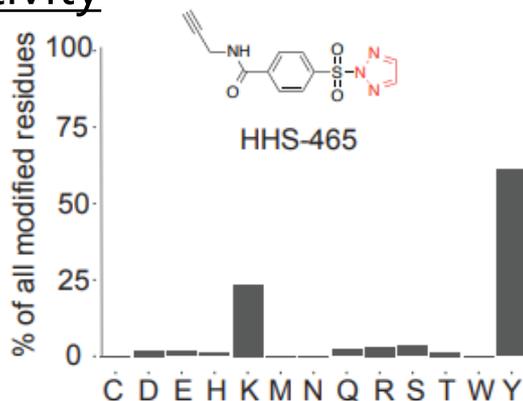


Proteome analysis

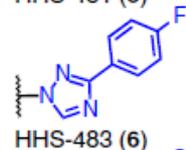
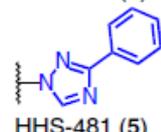
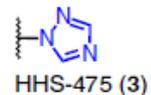
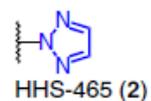
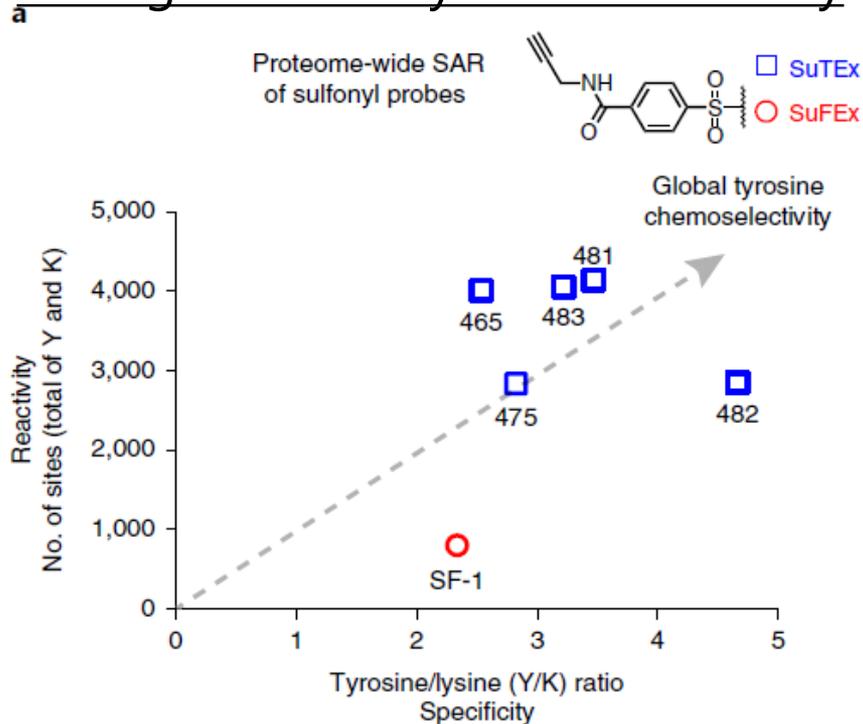


Tyr-Reactive Probe

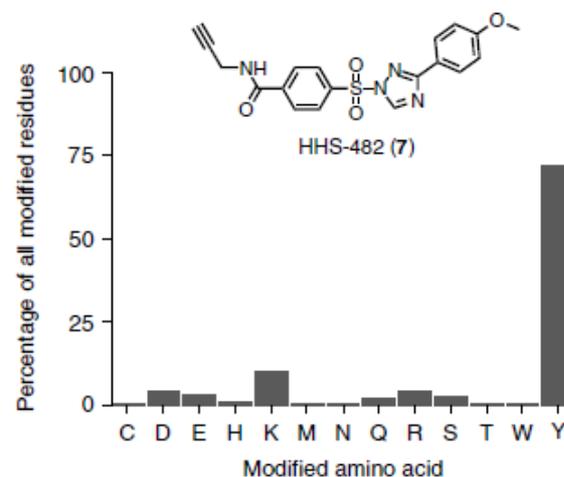
Residue selectivity



Tuning of LG for tyrosine selectivity



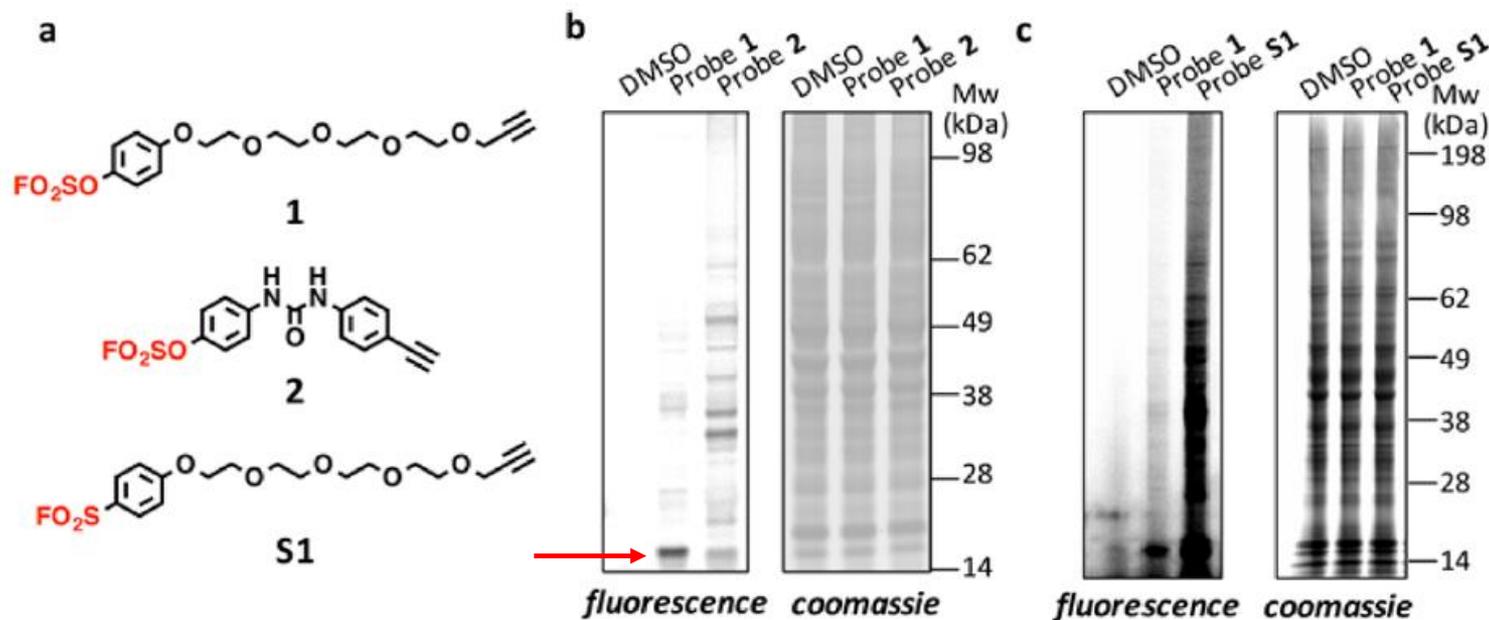
Tyrosine chemoselective probe



Covalent Inhibitor by SuFEx

Inhibitor of intracellular lipid binding protein (iLBP)

Evaluation of proteome reactivity of probes

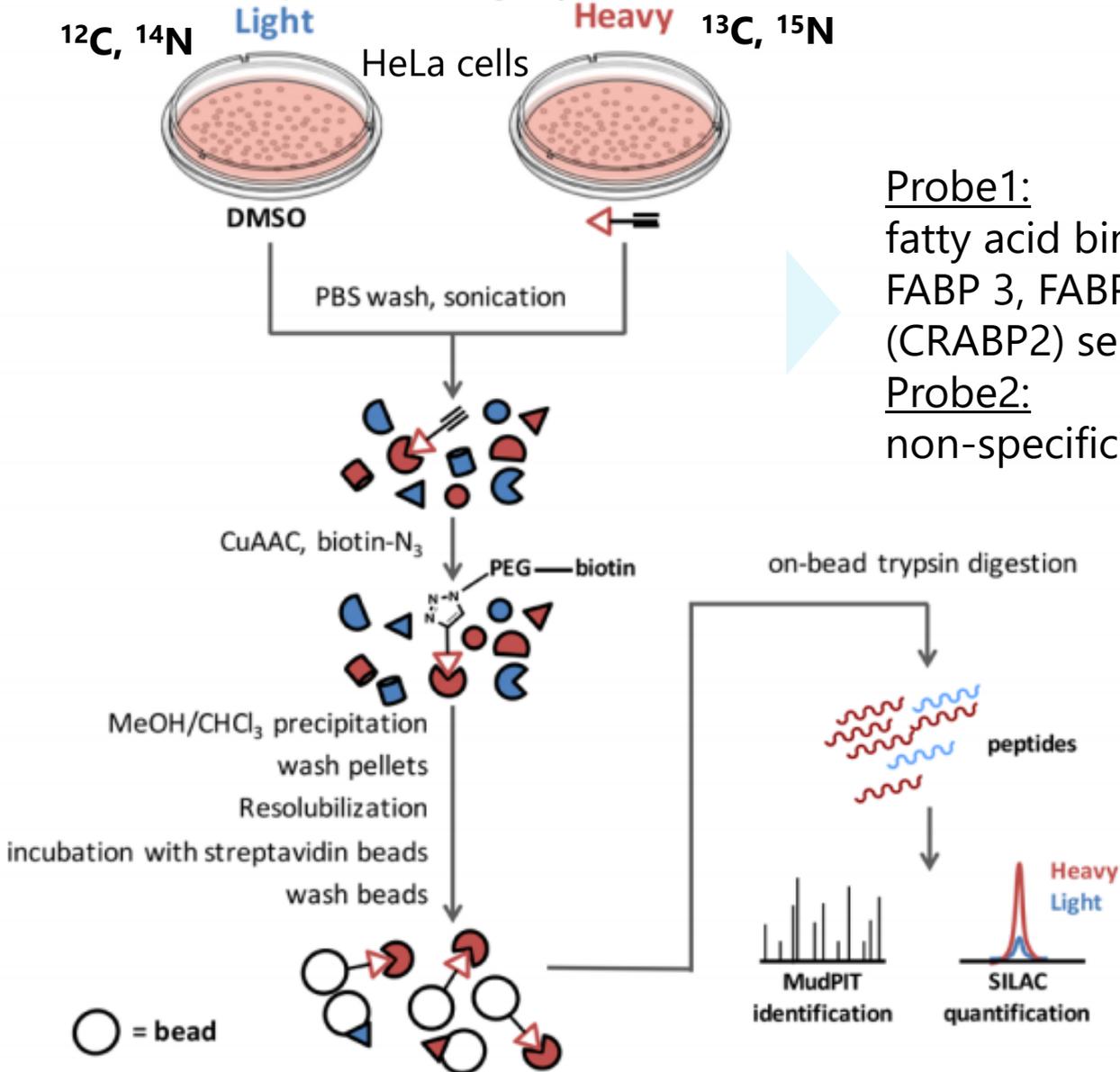


HeLa cell + blank/1/ 2 or S1 (40 μ M, 37 $^{\circ}$ C, 16 h),
CuAAC with cell lysate to attach fluolophore,
SDS-PAGE

Sulfonyl fluoride: highly reactive, resulting in non-specific reactivity

Covalent Inhibitor by SuFEx

Stable isotope labeling by amino acids in cell culture (SILAC)



Probe1:

fatty acid binding protein (a family of iLBPs)
FABP 3, FABP 5,
(CRABP2) selective labeling

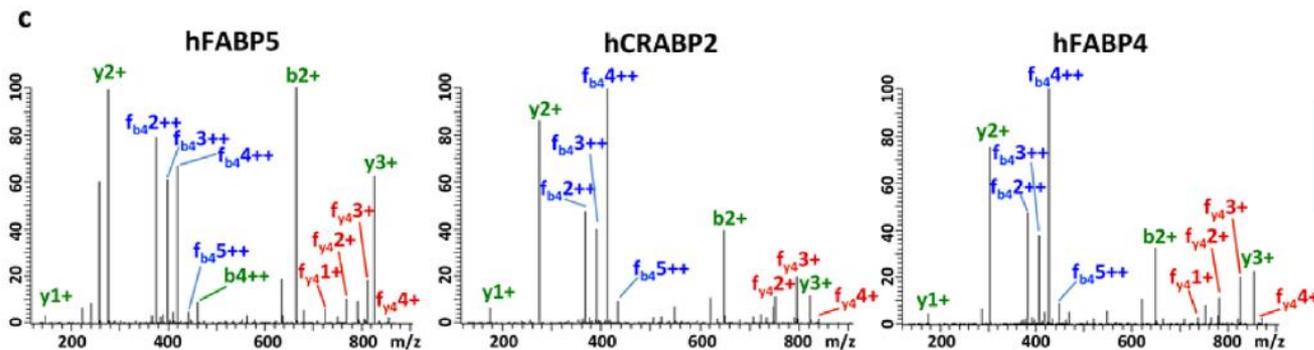
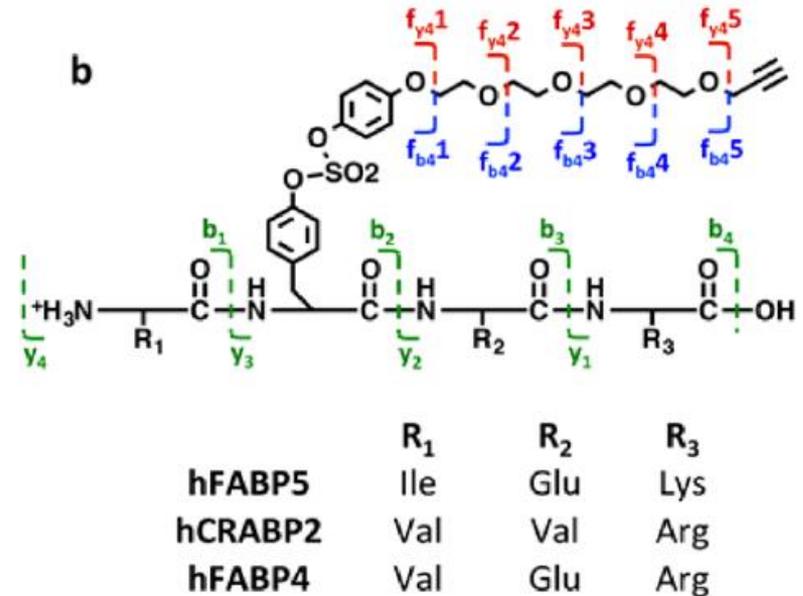
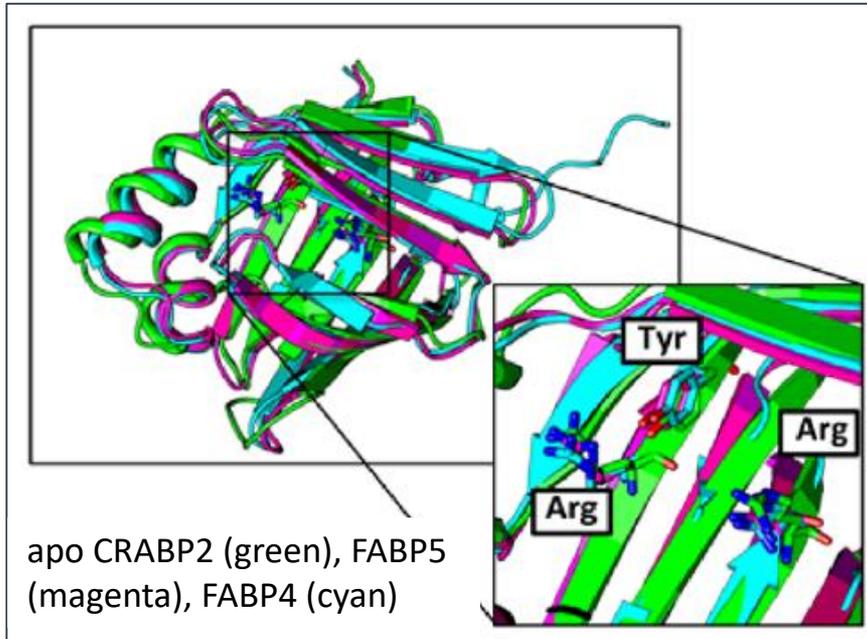
Probe2:

non-specific labeling

Covalent Inhibitor by SuFEx

Conserved Arg~Arg~Tyr module in iLBPs

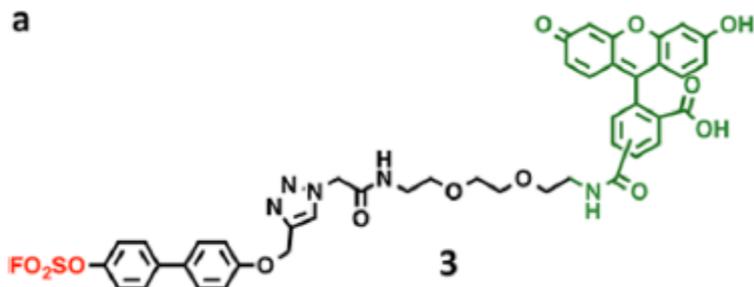
LC-MS/MS to determine reactive residue



All reacted at Tyr in Arg~Arg~Tyr module.

Covalent Inhibitor by SuFEx

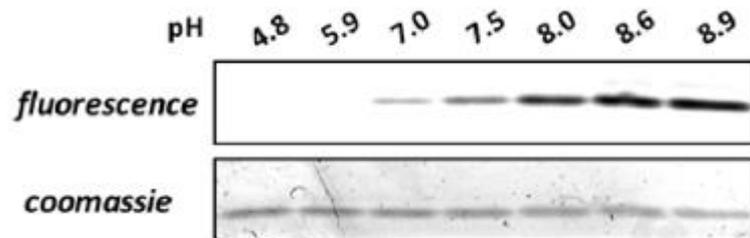
Introduction of biphenyl substructure



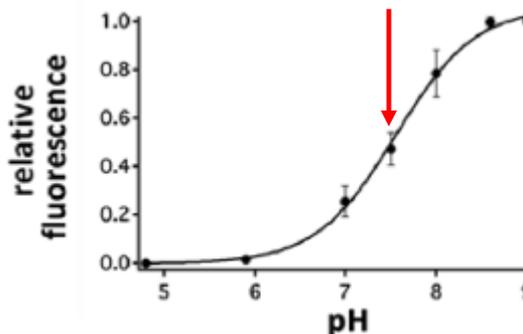
Improved reactivity for CRABP2
No reactivity with FABP3, 4, 5

Mechanism analysis of selectivity for CRABP2

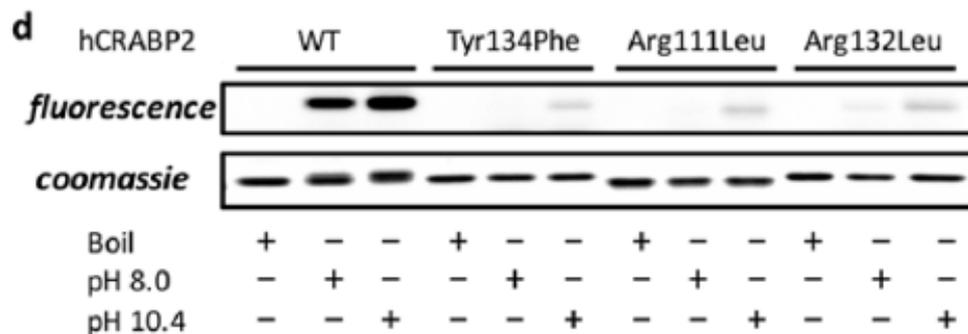
- Buffer pH dependence



Tyr134 pKa ~ 7.6
effect of Arg111, Arg132?



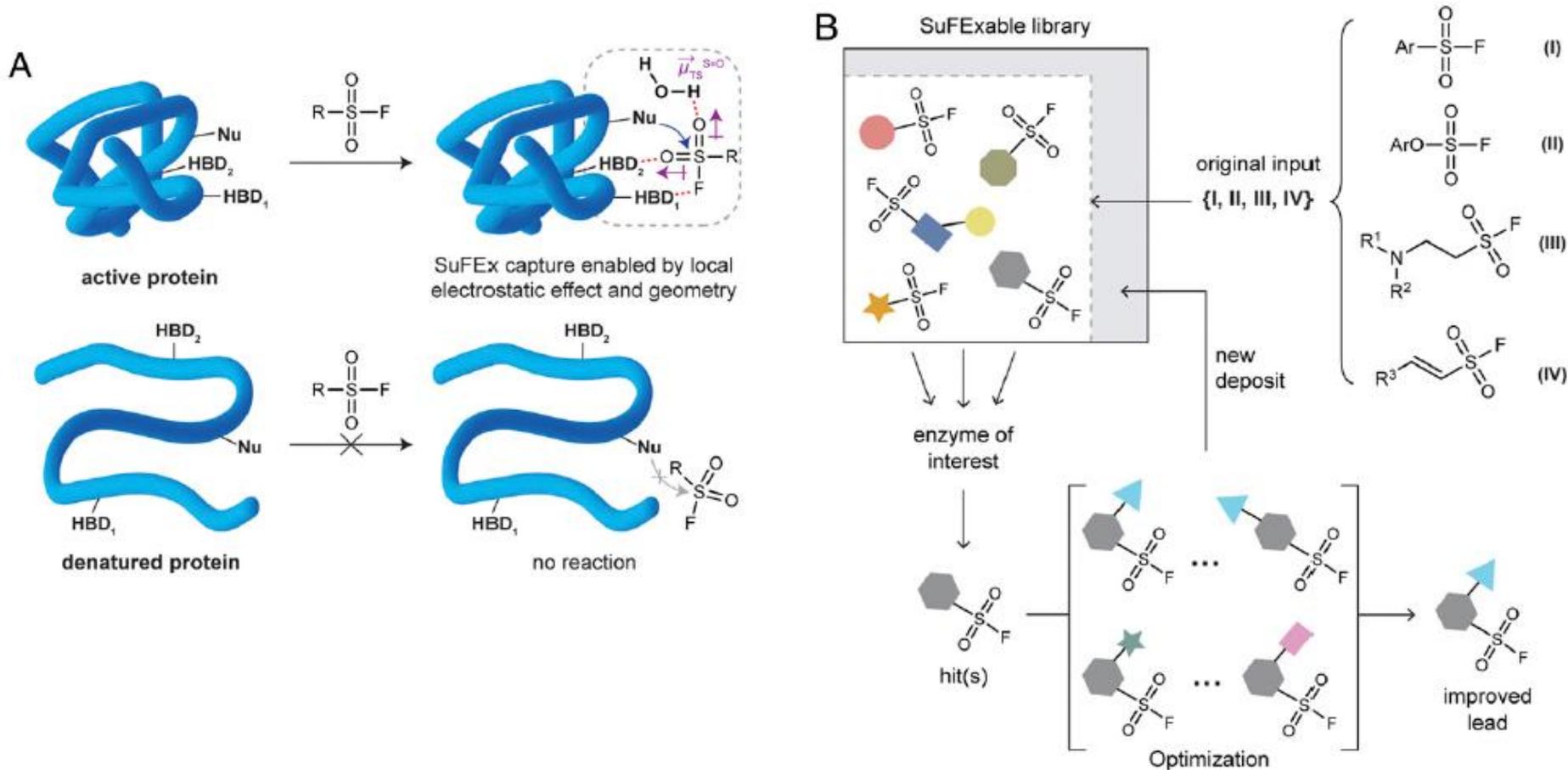
- Mutation experiment



Covalent Inhibitor by SuFEx

Covalent inhibitors of human neutrophil elastase (hNE).

PNAS 2019 116 18808

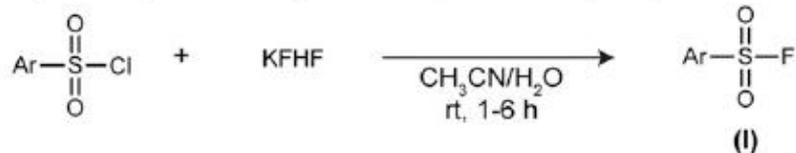


Covalent Inhibitor by SuFEx

Categorized SuFExable library

105 compounds

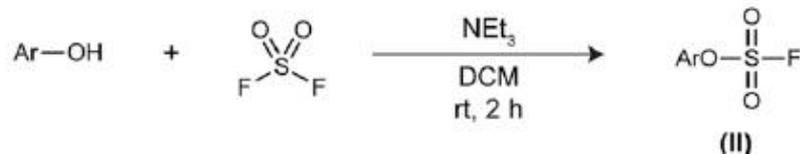
A aryl sulfonyl fluoride (33 entries, 90–100% yield)



relative rate
with ArOTBS

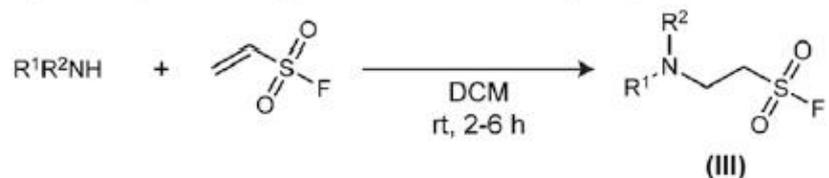
28.5

B aryl fluorosulfate (32 entries, 82–99% yield)



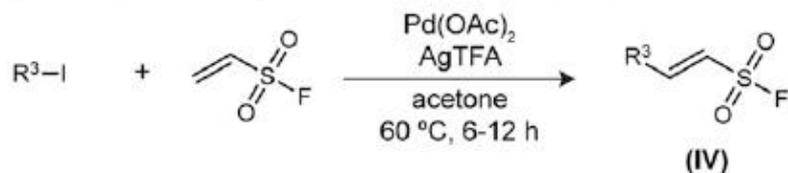
1

C alkyl sulfonyl fluoride (30 entries, 85–98% yield)



14.1

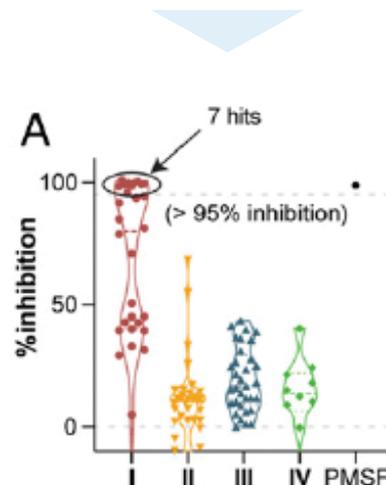
D vinyl sulfonyl fluoride (10 entries, 59–99% yield)



4.1

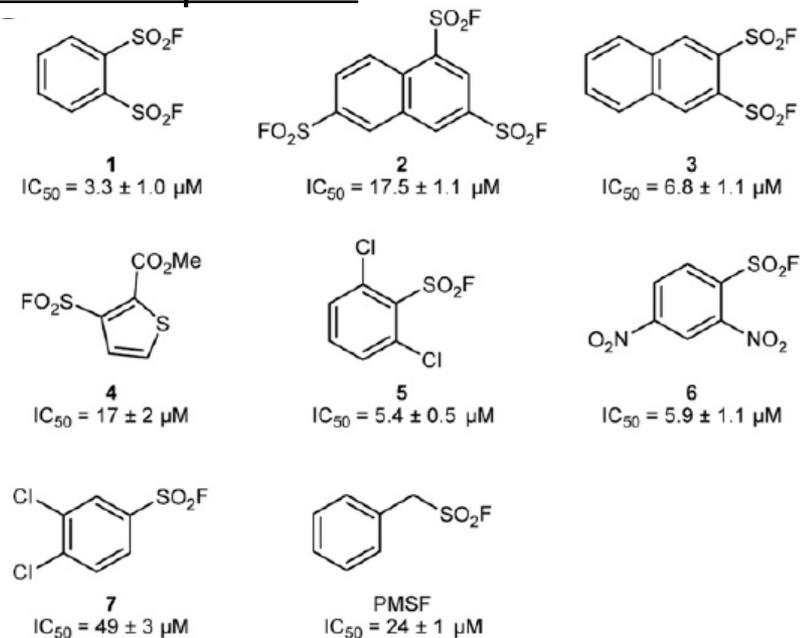
1st screen

Incubation with enzyme for 10 min
Addition of substrate peptide
(MeOSuc-AAPV-AMC)
Measure increase in fluorescence

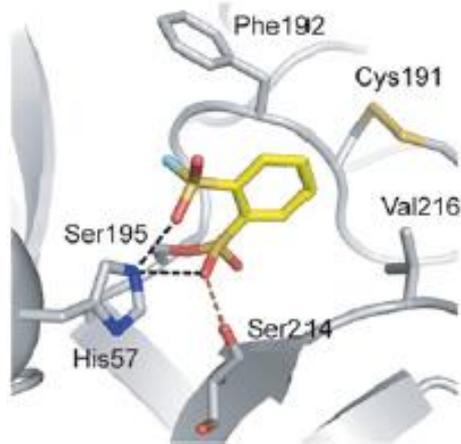


Covalent Inhibitor by SuFEx

Hit compounds



Crystal structure hNE+1

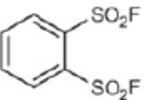
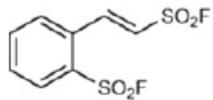
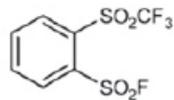
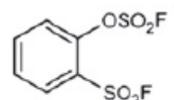
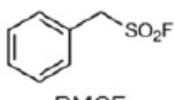


2nd screen

Compound	R =	IC_{50} (μM)*
1	SO ₂ F	3.3 ± 1.0
8	F	~120
9	Cl	82 ± 16
10	Br	20 ± 10
11	I	9.7 ± 1.2
12	Me	>200
13	OMe	73 ± 4
14	CN	13.3 ± 0.5
15	CF ₃	60 ± 8
16	NO ₂	20 ± 1
17	CO ₂ Me	37 ± 2
18	Ph	27 ± 1
19		2.2 ± 0.7
20		84.4 ± 0.6
21		>200
22	SO ₂ CF ₃	1.1 ± 0.1
23	SO ₂ (CF ₂) ₂ CF ₃	48 ± 2
24	OSO ₂ F	0.24 ± 0.02

Covalent Inhibitor by SuFEx

Selectivity between serine proteases

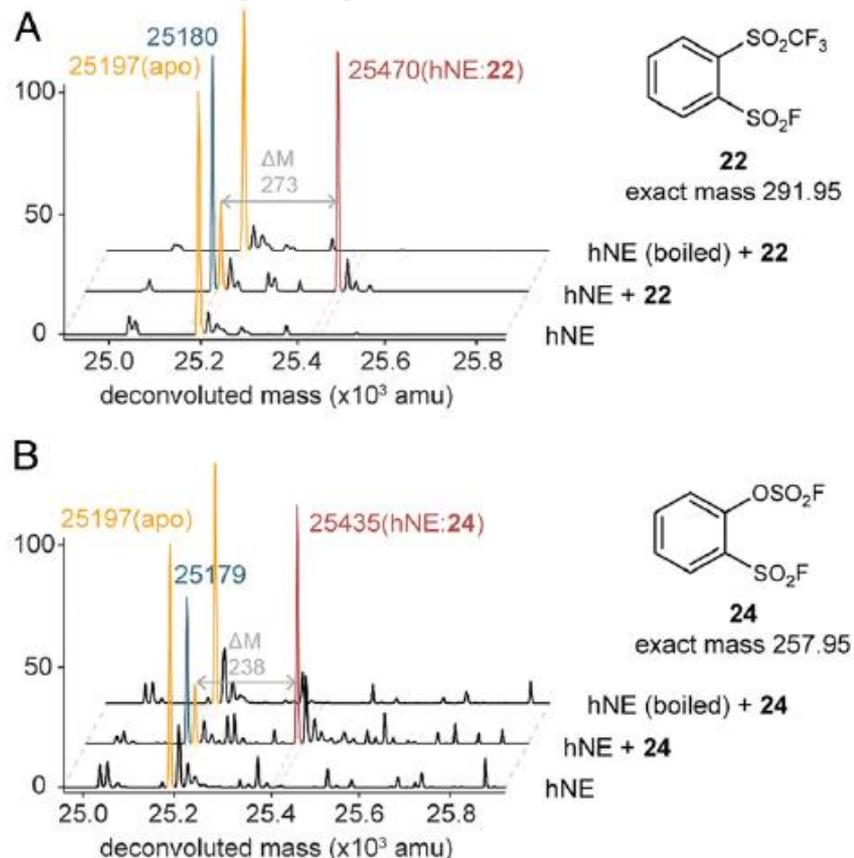
Compound	hNE IC ₅₀ (μM)*	hCG IC ₅₀ (μM)*	S [†]
 1	3.3 ± 1.0	190 ± 40	58
 19	2.2 ± 0.7	6.0 ± 0.7	2.7
 22	1.1 ± 0.1	>200	>182
 24	0.24 ± 0.02	>200	>833
 PMSF	24 ± 1	69 ± 6	2.5

*IC₅₀ values were measured based on 10-min incubation and are shown in mean ± SD (n ≥ 3).

[†]S value denotes the selectivity, defined by the ratio of IC₅₀ (hCG) over IC₅₀ (hNE).

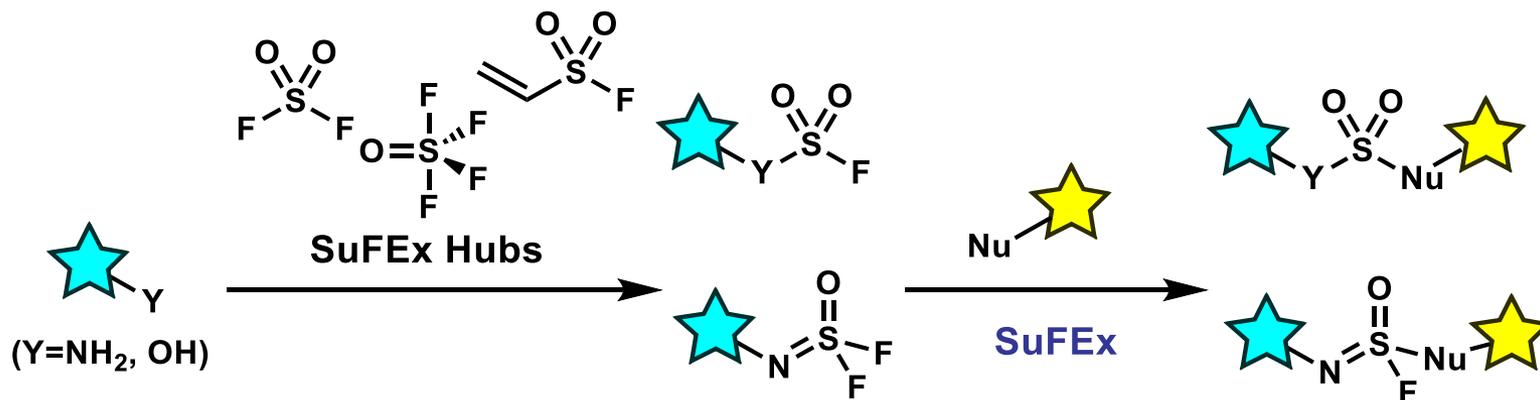
Effect of denaturation

MALDI-TOF-MS



Summary

SuFEx



- Advocated by Sharpless in 2014.
- Stable S(VI)-F bond
- Context dependent activation

New synthetic methodology
Proteome analysis
Analysis of PPI
Medicinal chemistry
Polymer synthesis

Challenges

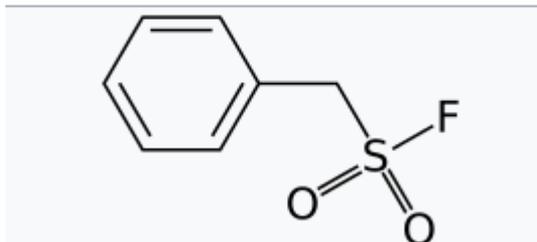
Mild/sharp activation of S-F bond
Tuning of leaving group instead of F-
Elucidation of mechanism

Appendix

Unstable Sulfonyl Fluoride

PMSF: serine protease inhibitor

Phenylmethanesulfonyl fluoride
(PMSF)



short half-life under aqueous conditions

pH 7: 110 min

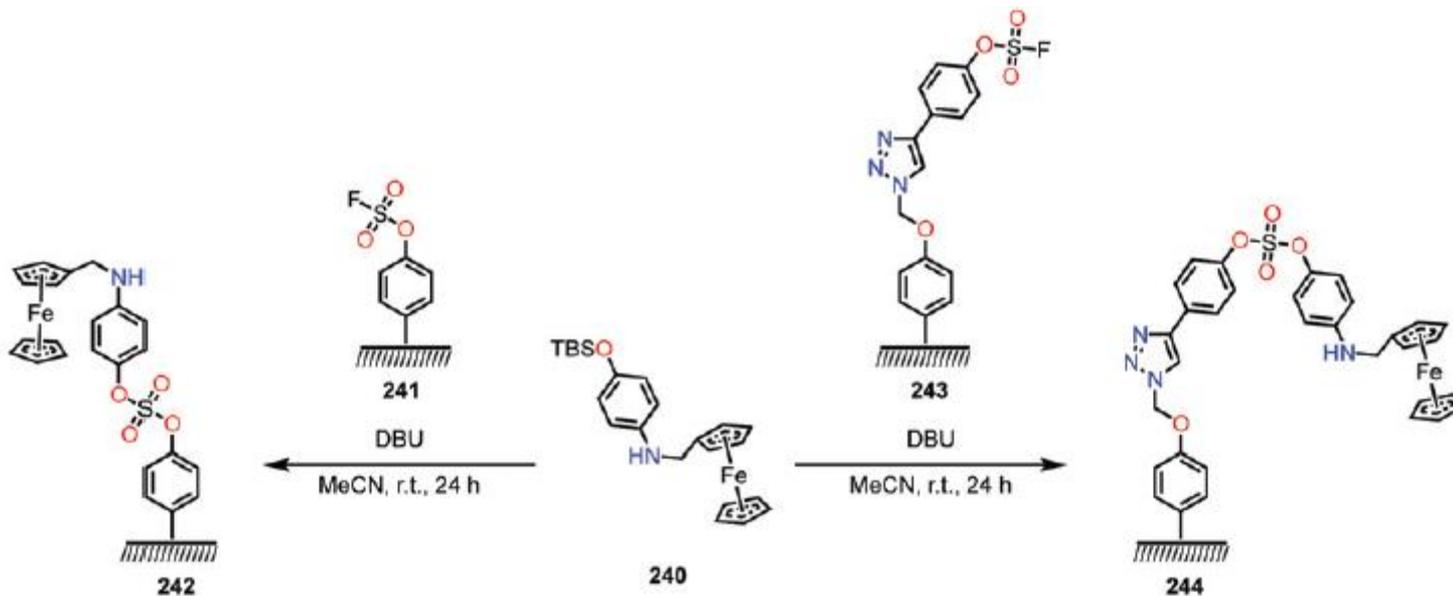
pH 7.5: 55 min

pH 8: 35 min

unstable under basic conditions
(forms sulfene via E₁cB elimination)

Aliphatic sulfonyl fluoride encounter the risk of hydrolysis under basic conditions.

Stability under Electrolytic Conditions

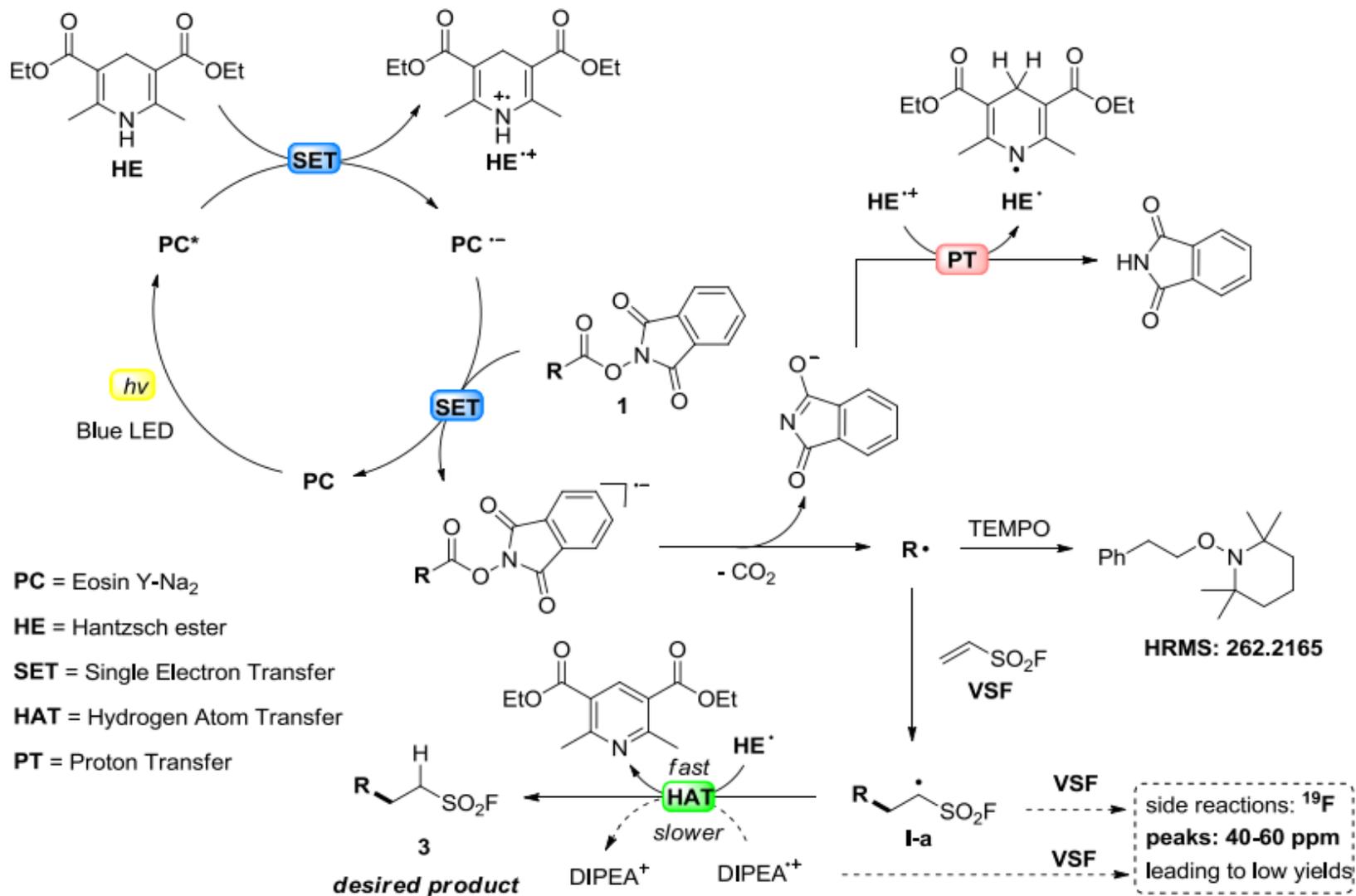


Scheme 48 Modification of carbon fibers by SuFEx click chemistry.

Electrochemical instability of 1,2,3-triazole is demonstrated with CV.

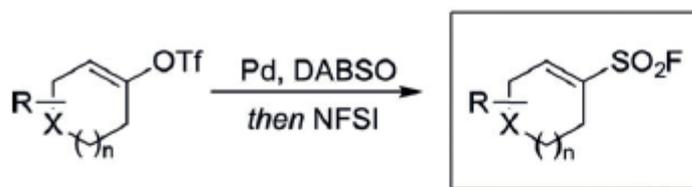
Radical Addition to ESF

Entire mechanism

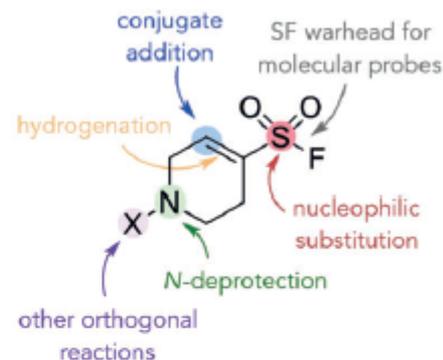


Sulfonylfluoride (RSO₂F)

Cyclic alkenylsulfonyl fluoride



- Readily available starting materials
- Compact molecular scaffold
- Multiple reactive sites



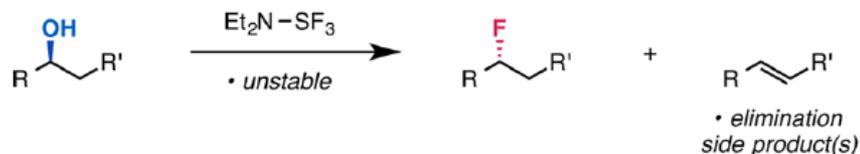
A family of compact low-molecular-weight cyclic alkenylsulfonyl fluorides can be readily prepared from the corresponding alkenyl triflates using palladium catalysis. These densely functionalized

reagents undergo a diverse range of derivatization reactions, including substitution at sulfur, conjugate addition, and N-functionalization.

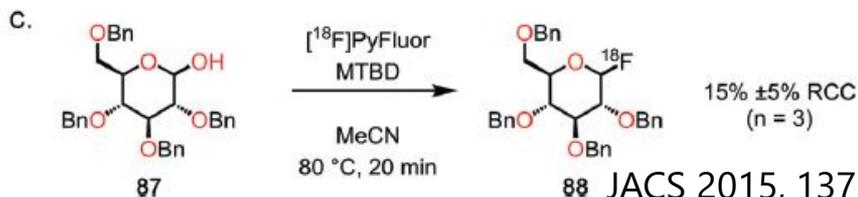
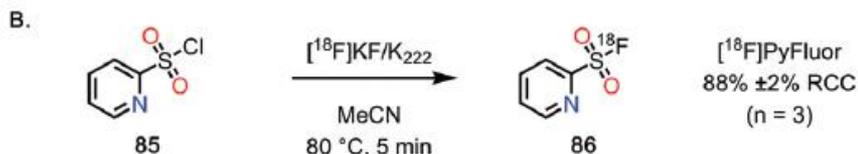
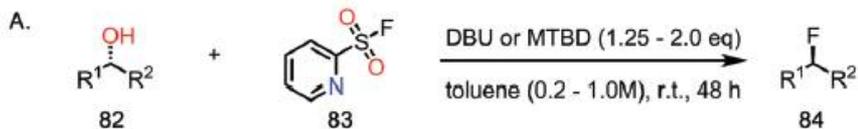
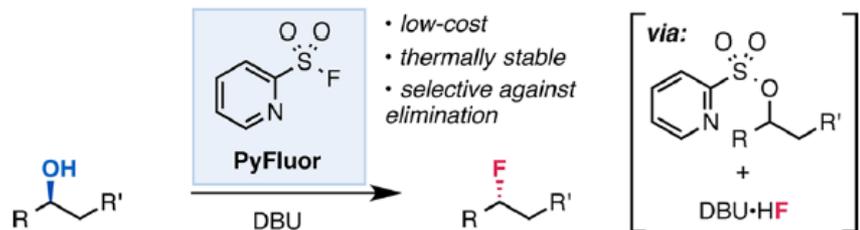
RSO₂F as Synthetic Reagent

PyFluor: deoxyfluorination

A. Deoxyfluorination with DAST

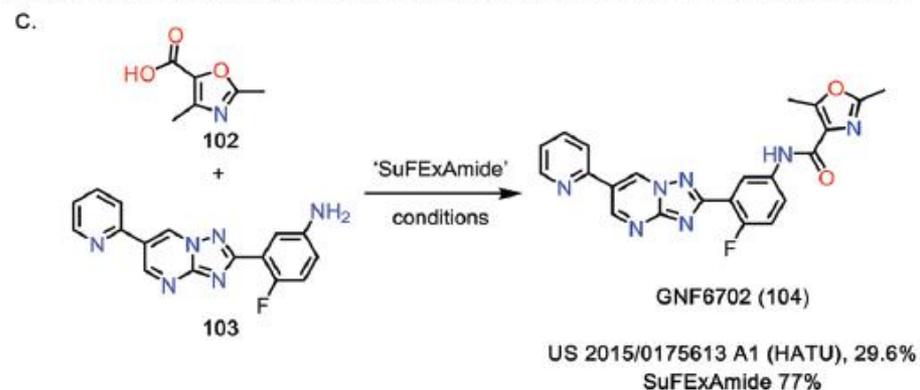
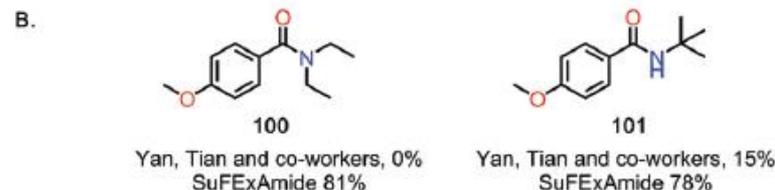


B. This work: Deoxyfluorination with PyFluor



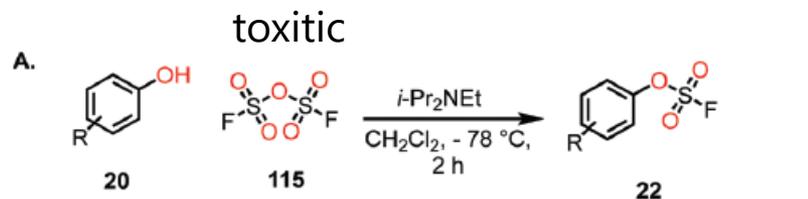
JACS 2015, 137, 9571

Coupling reagent

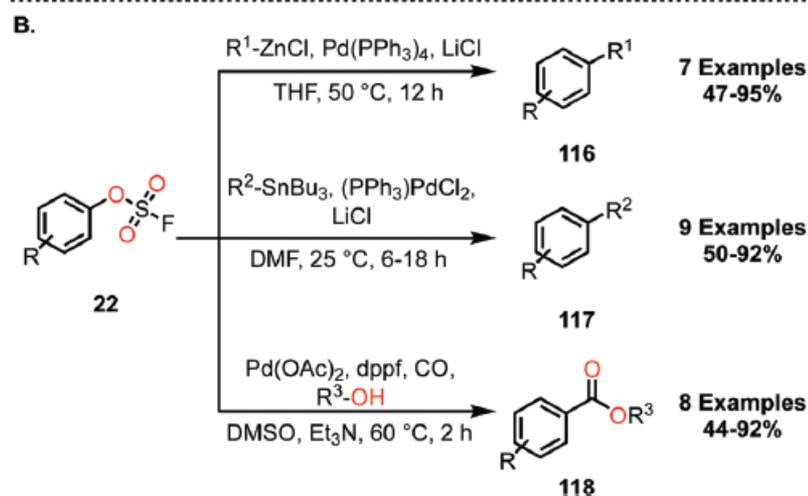


RSO₂F as Synthetic Reagent

ArOSO₂F as cross-coupling partner



SO₂F₂, AIF₃, imidazolium salt can be alternative.

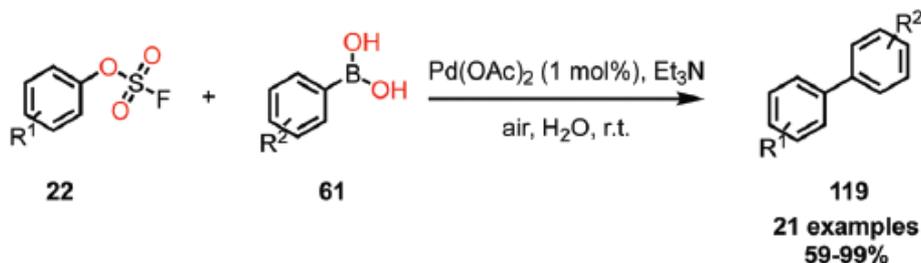


✓ high reactivity

-OSO₂F ~ -OTf, -Br > -OTs, -OMs
by Hanley

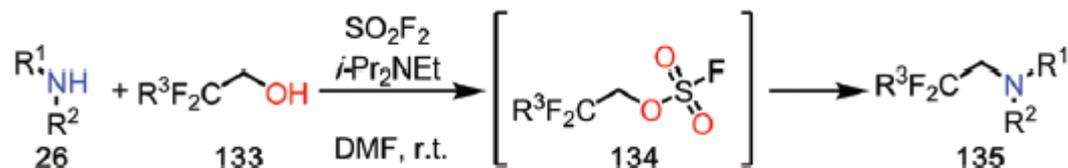
-Br ≥ -OTf > -OSO₂F > -Cl
by Sharpless

Additive free Suzuki-Miyaura coupling



RSO₂F as Synthetic Reagent

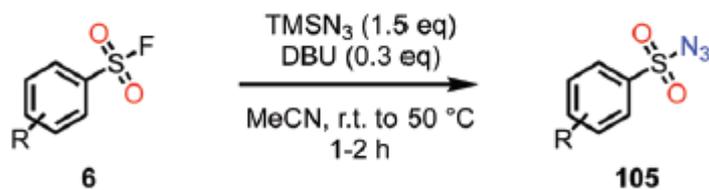
1,1-Dihydrofluoroalkylation of Amines



- no reaction with -I, -OTs, -OMs

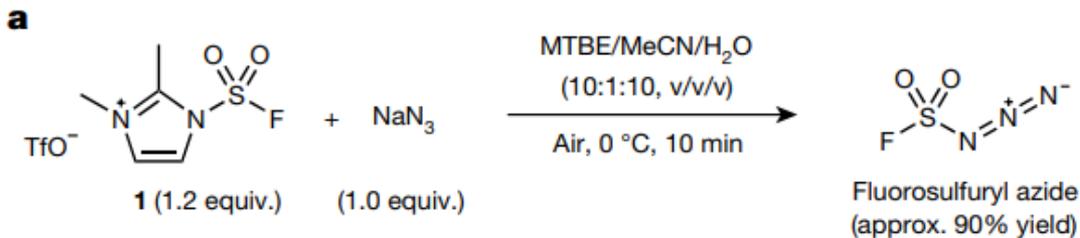
Synthesis of sulfonyl azide

Synlett 2016, 27, 1840

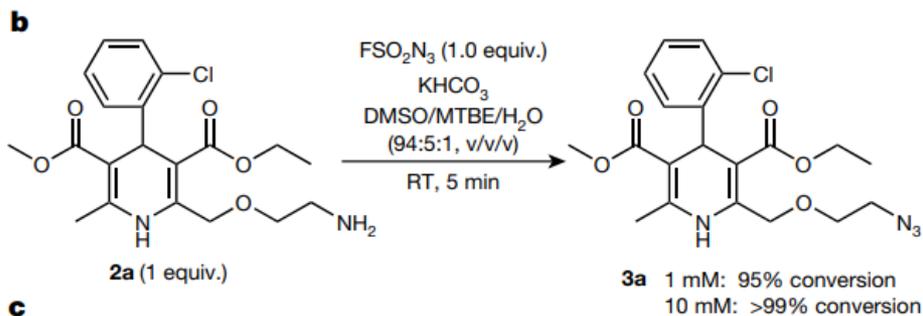
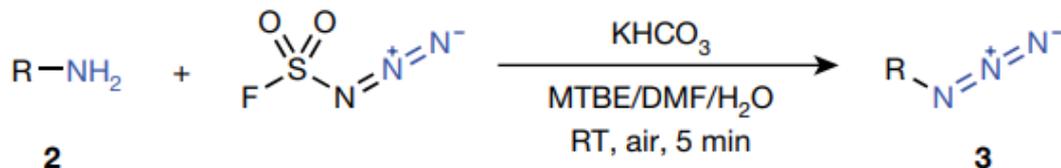


RSO₂F as Synthetic Reagent

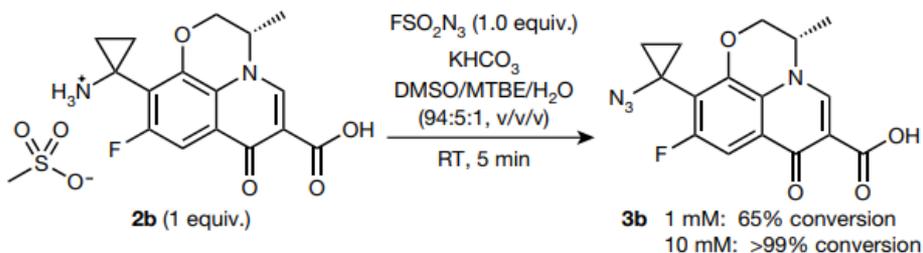
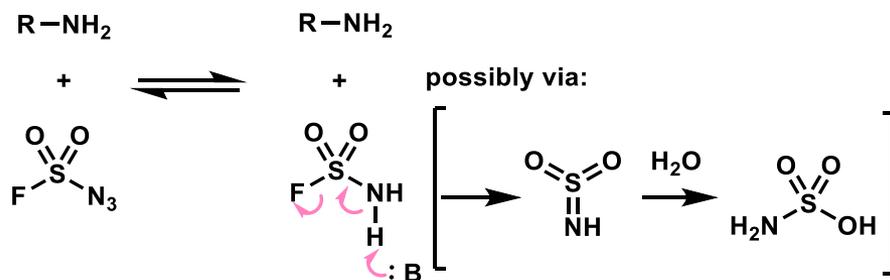
Diazotizing reagent: primary amine/sulfonamide to azide/sulfonylazide



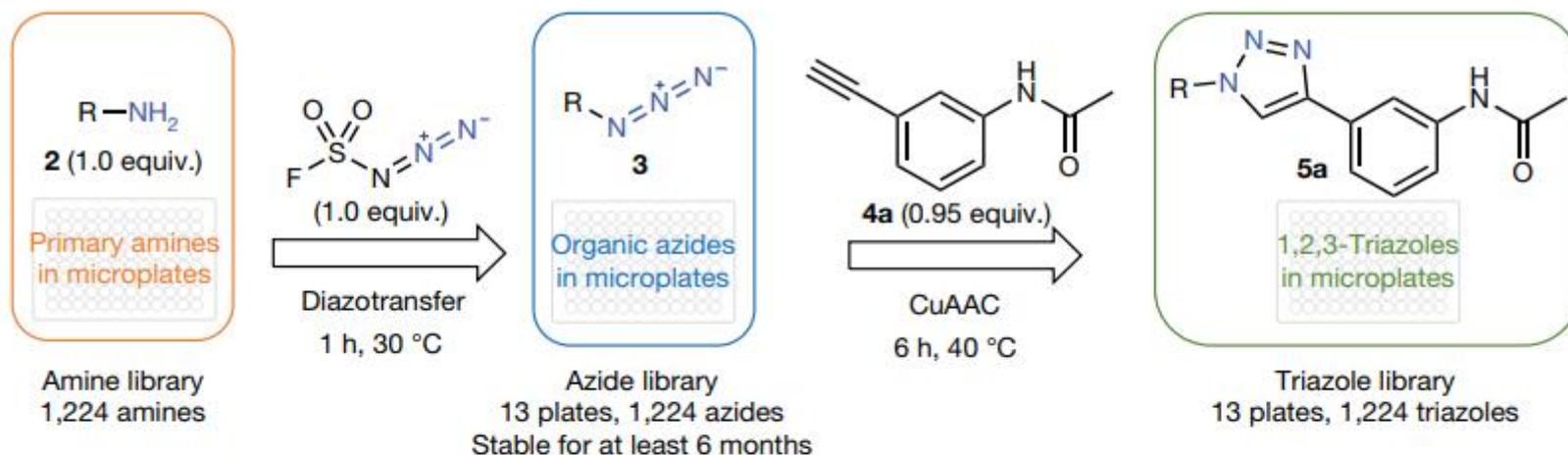
> 40 examples
orthogonal to most of function
(including 2°/3° amines)



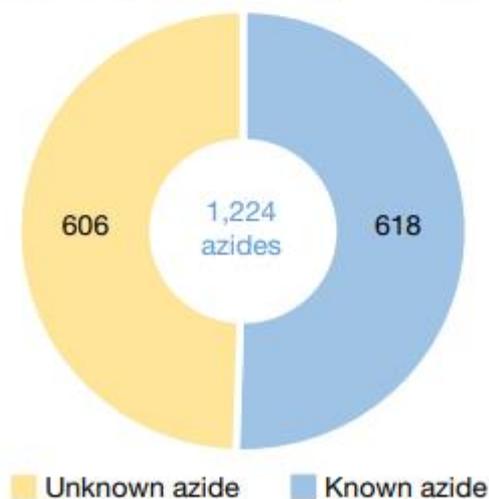
diazotransfer is reversible



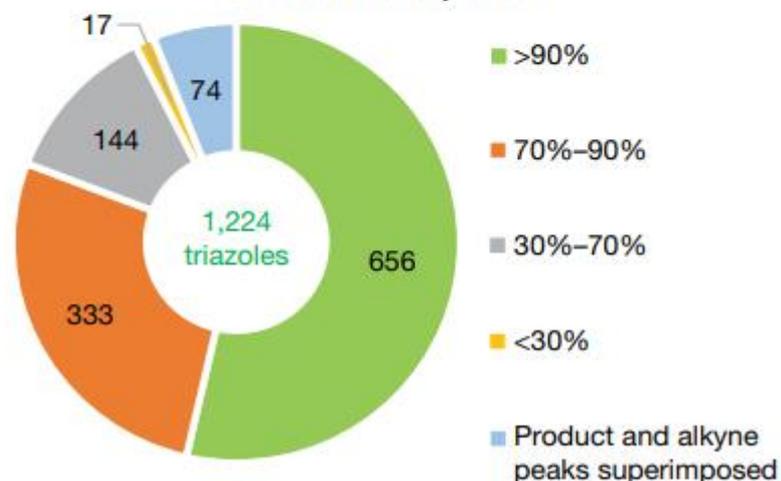
RSO₂F as Synthetic Reagent



Unknown compounds in the 1,224-azide library

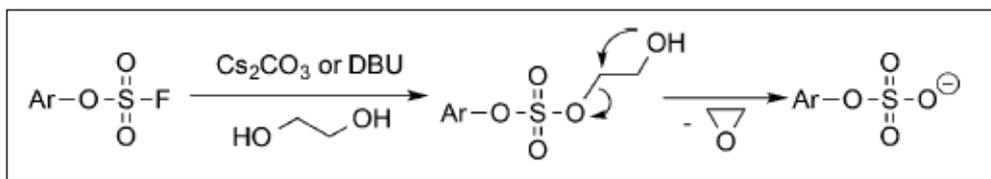
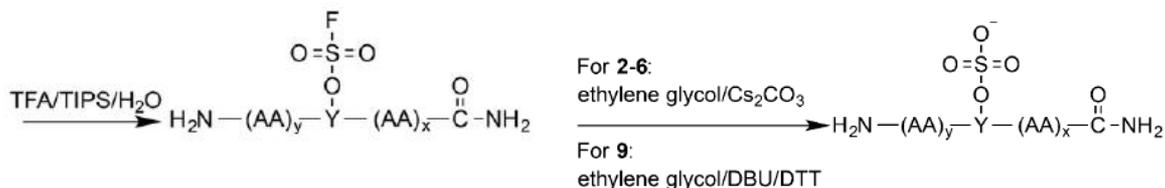
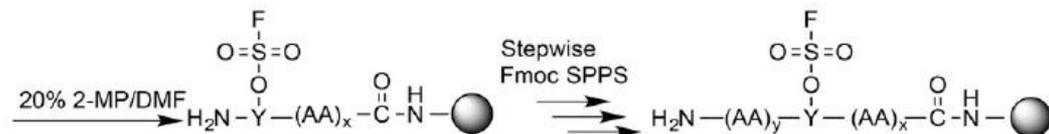
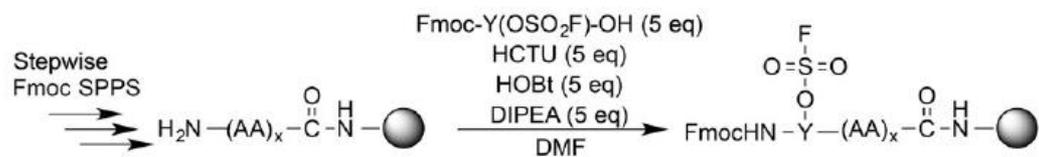
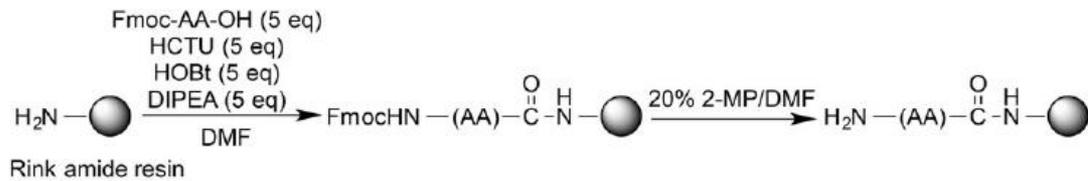
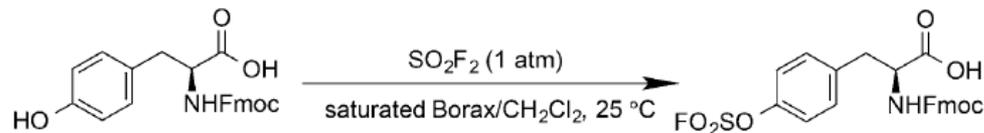


Conversion of alkyne 4a



Synthesis of Sulfotyrosine

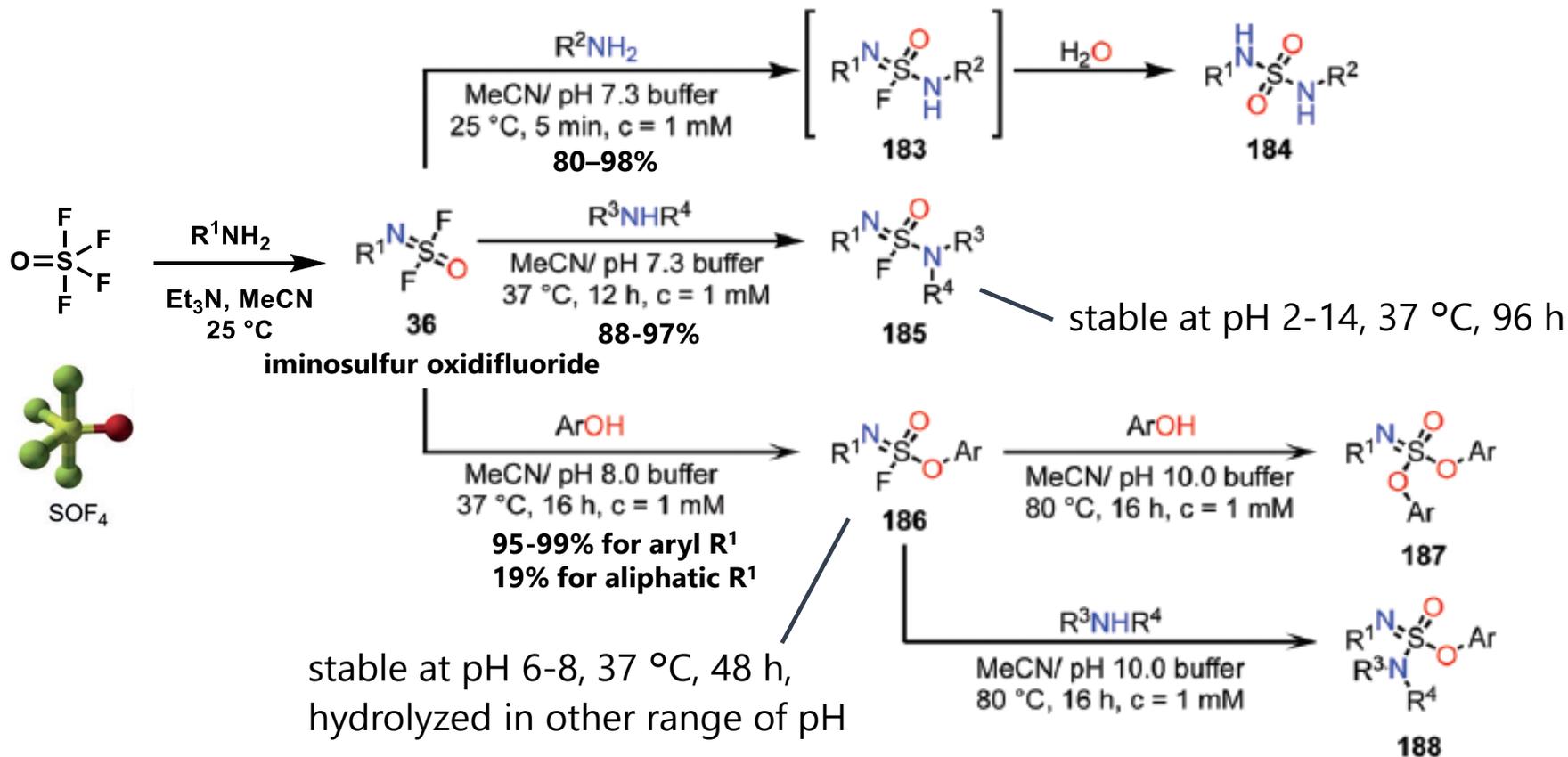
Tyr O-sulfation: seen in PTMs



Iminosulfur Oxidifluoride under Aqueous Conditions

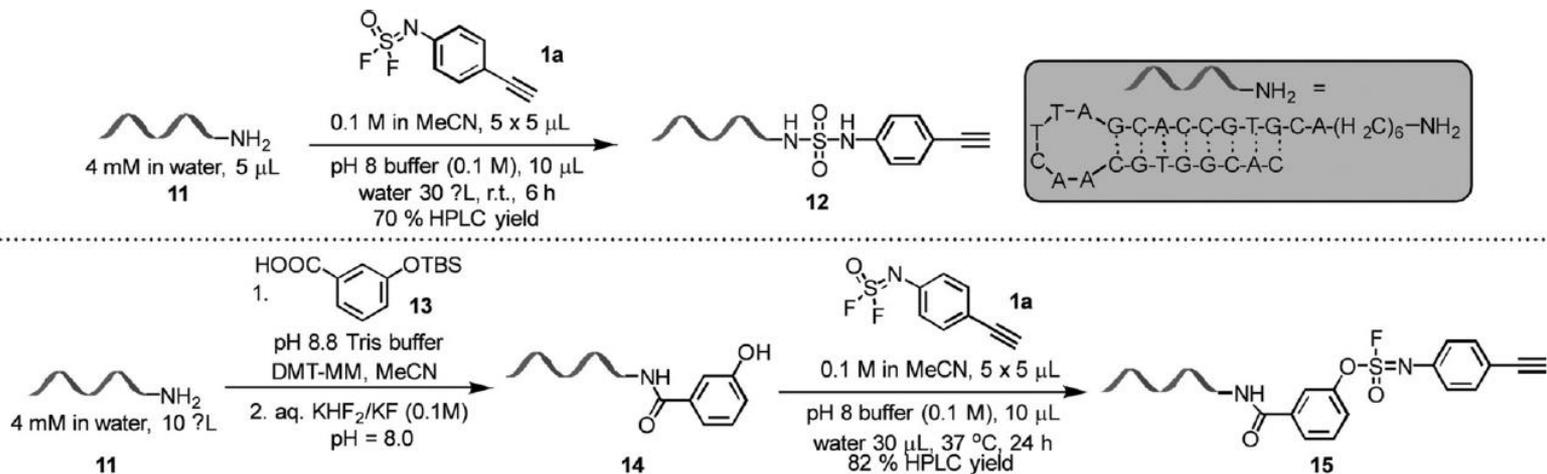
Derivatization under aqueous conditions

ACIE 2019, 58, 8



Iminosulfur Oxidifluoride under Aqueous Conditions

DNA modification



Protein modification

