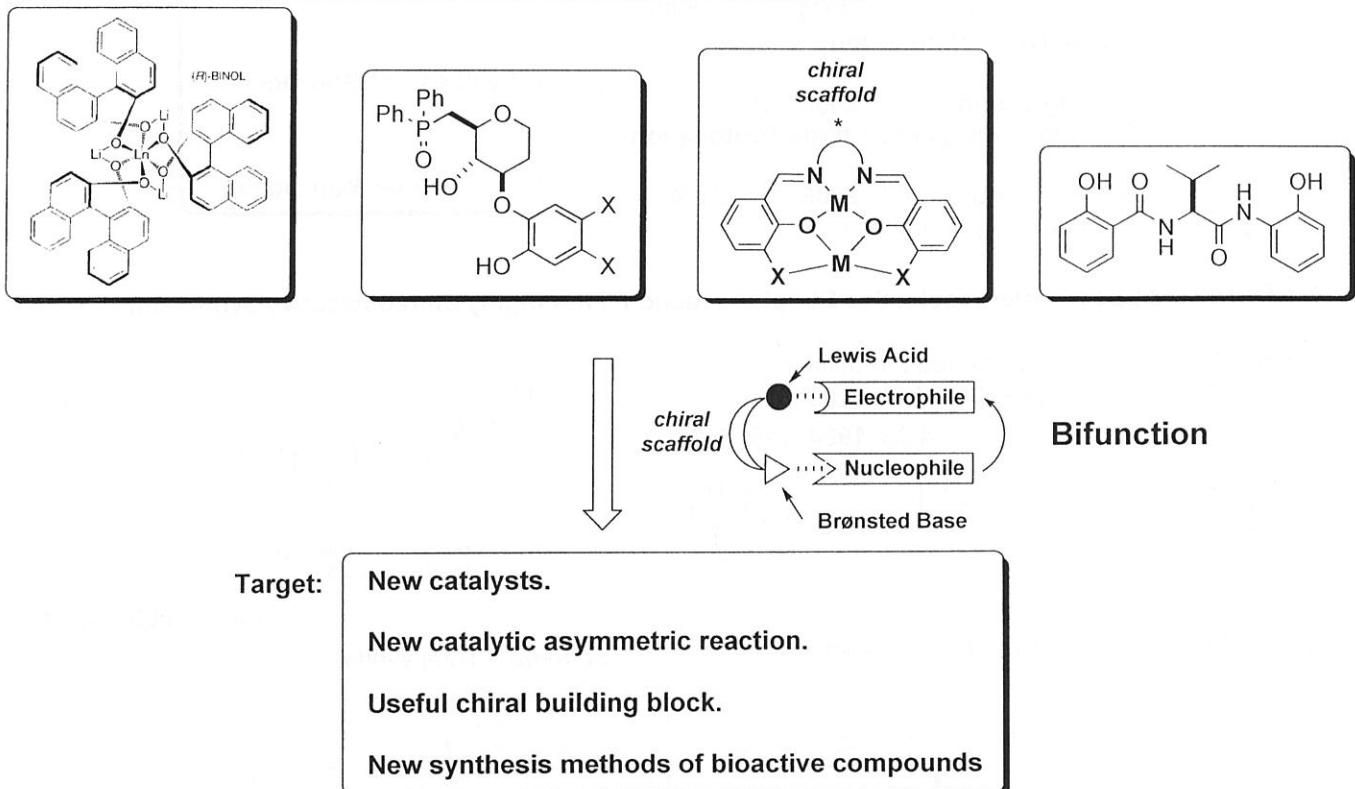


“研究の流れを創ろう”

Why this topic?

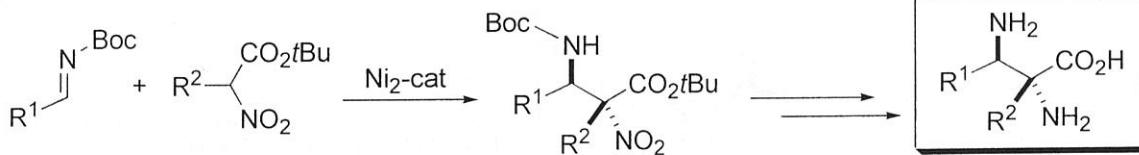
Chen Zhihua (D2)
2009/4/4

Original concept of our lab: Multi-metal center catalysts

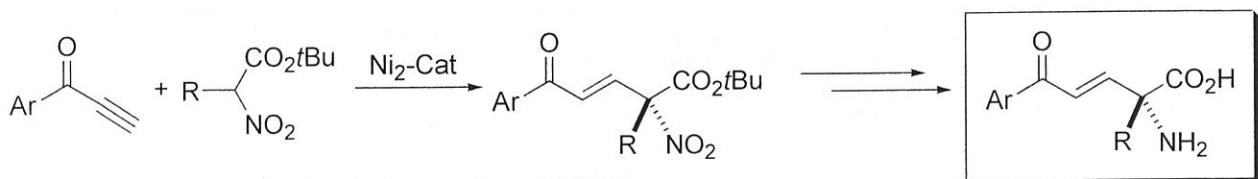


Why Manzacidin: To apply my research results:

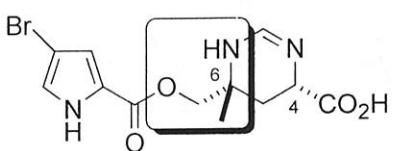
i α,β -diamino acid:



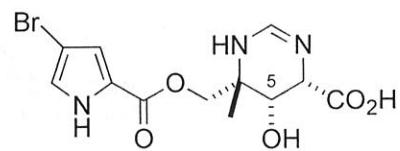
ii α,α -disubstituted amino acid



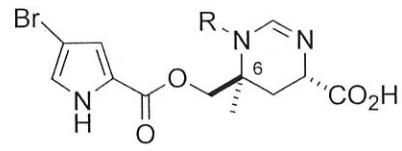
One of Nature compounds contain α,α -disubstituted amino acid: Manzacidin



Manzacidin A



Manzacidin B



Manzacidin C: R = H
Manzacidin D: R = Me

Isolation: Isolated from the Okinawan sponge *Hymeniacidon* sp. (Kobayashi et al. JOC. 1991, 56, 4574.)
Bioactivity: α -adrenoceptor blockers, actomyosin ATPase activators
Structure: C4, 6-tertiary-quarternary carbon center

Typical total synthesis of Manzacidin:

1 Ohfune Group: JACS. 2000, 122, 10708.

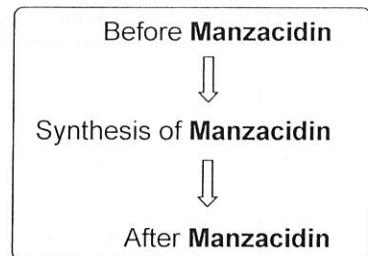
(Diastereoselective Strecker reaction)

First highly stereoselective synthesis)

2 Du Bois Group: JACS. 2002, 124, 12950.
(Oxidative C-H insertion)

3 Li Deng Group: JACS. 2006, 128, 3928.
(Tandem conjugate addition-Protonation)

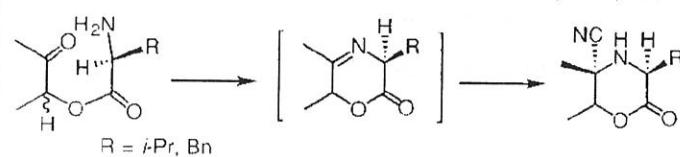
4 Maruoka Group: JACS. 2006, 128, 2174.
(1,3 Dipolar cycloaddition)



1 Ohfune Group: (Diastereoselective Strecker reaction, First highly stereoselective synthesis)

Concept: Synthesis of α -substituted threonine:
Diastereoselective Strecker reaction

JACS. 1994, 116, 7405.

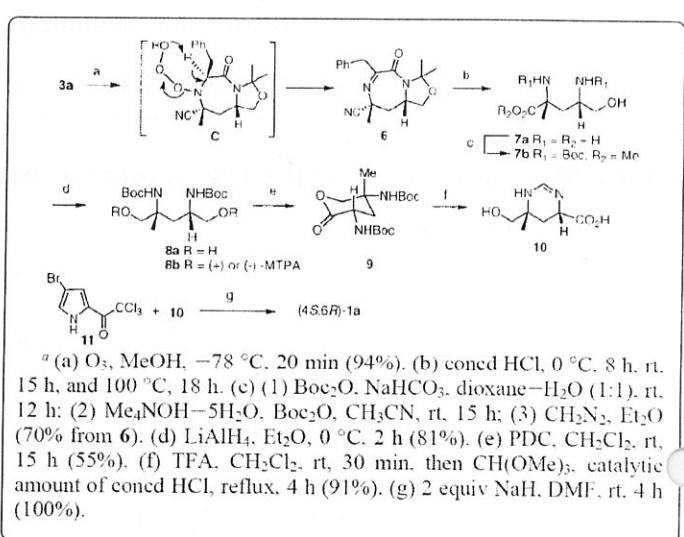
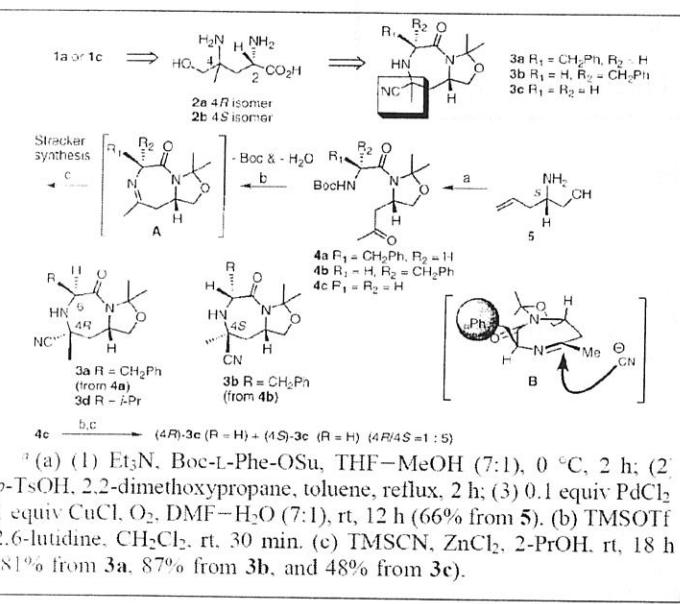


Scheme 1 Diastereoselective Strecker reaction

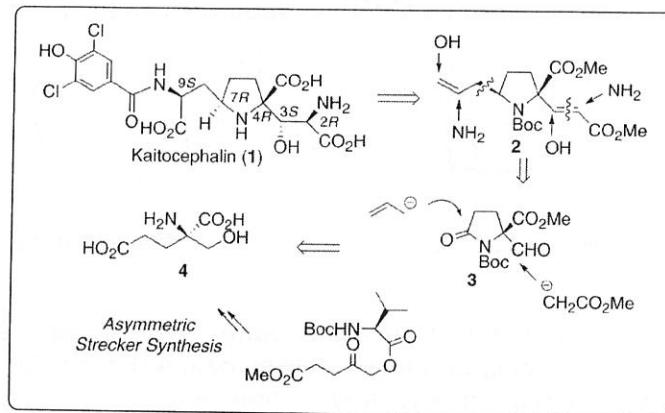
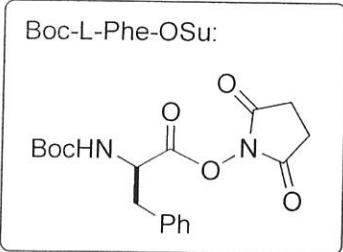


JACS. 2000, 122, 10708.

Scheme 2 Total synthesis



Scheme 3 Total synthesis of Kaitocephalin
JOC. 2005, 7, 4165.



2 Du Bois Group

i Before Manzacidin

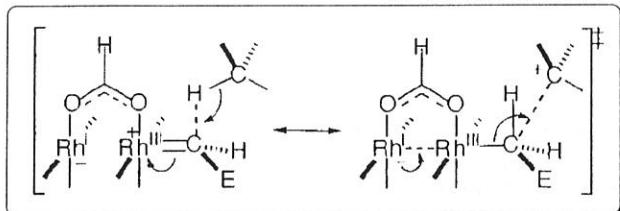
a Background of C-H activation around 2001:

Chem. Rev. **1998**, *98*, 911.

Journal of Organo Metallic Chemistry **2001**, *618*, 47.

Scheme 5 Carbene C-H insertion mechanism speculated by Nakamura group

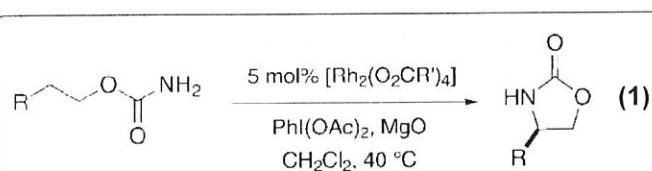
JACS **2002**, *124*, 7181.



Also see literature Seminar of Yamaguchi(2006) and Tanaka (2007)

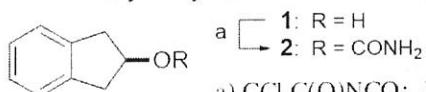
b A Rh-Catalyzed C-H Insertion Reaction for the Oxidative Conversion of Carbamates to Oxazolidinones^{**}

Angew **2001**, *40*, 598.



Benefit:

1 SM is easy to synthesis.

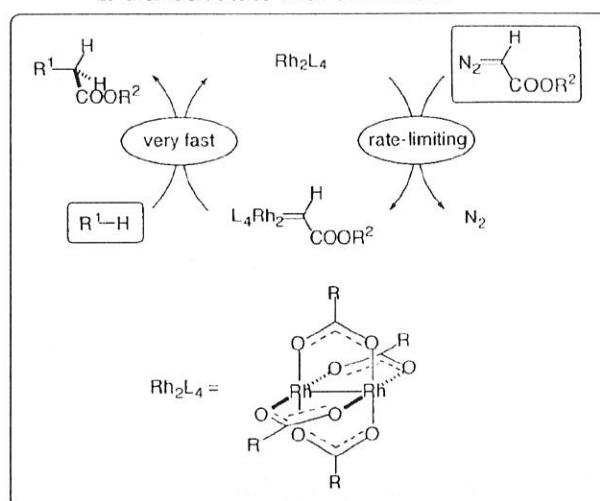


a) $\text{CCl}_3\text{C}(\text{O})\text{NCO}$; $\text{K}_2\text{CO}_3/\text{MeOH}$;

2 Commercial oxidant: $\text{PhI}(\text{OAc})_2$

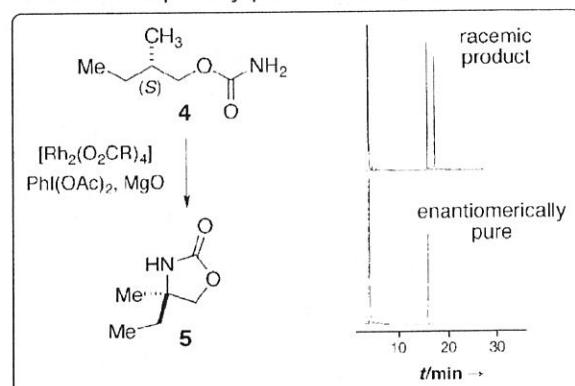
3 Nitrogen can be introduce to unactivated C-H bond.

Scheme 4 Rhodium Tetracarboxylate-catalyzed C-H bond activation/C-C bond forming reaction of an α -diazoacetate with an Alkane



Concept:
New convenient way: in-situ generate $\text{L}_4\text{Rh}_2=\text{NHR}$

Scheme 6 Optically pure carbamate



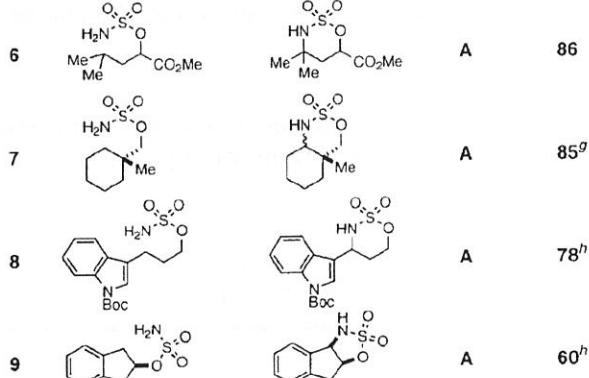
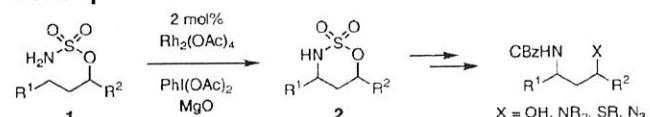
c Synthesis of 1,3-Difunctionalized Amine Derivatives through Selective C-H Bond Oxidation

JACS **2001**, *123*, 6935.

Table 1

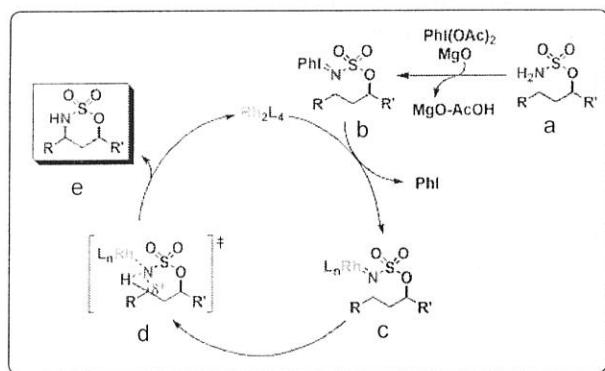
Entry	Substrate	Product	Catalyst ^a	Yield ^b
1			A	90 ^c
2			B	75 ^d
3			A	80 ^c
4			B	91 ^e
5			B	78 ^f

Concept:

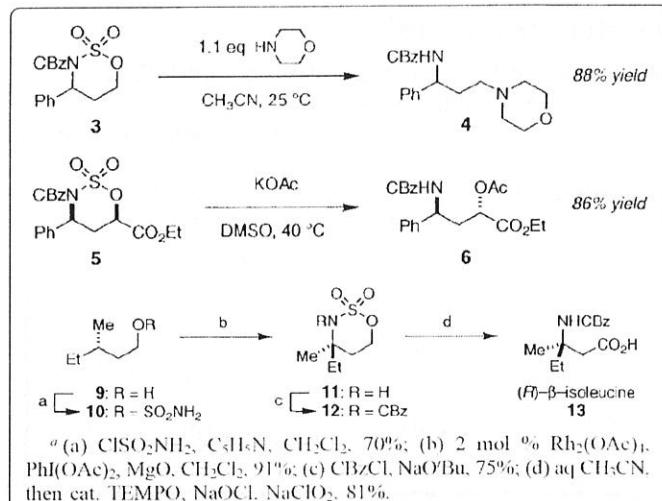


^a (a) Catalyst: A = $\text{Rh}_2(\text{OAc})_4$, B = $\text{Rh}_2(\text{oct})_4$. (b) Reactions (Rhodium octanoate)

Scheme 7 Mechanism



Scheme 8 Conversion reaction



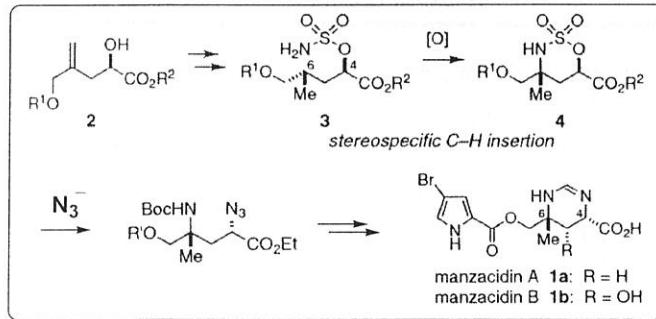
^a (a) CISO_2NH_2 , $\text{C}_5\text{H}_5\text{N}$, CH_2Cl_2 , 70%; (b) 2 mol % $\text{Rh}_2(\text{OAc})_4$, $\text{PhI}(\text{OAc})_2$, MgO , CH_2Cl_2 , 91%; (c) CBzCl , NaO^+Bu , 75%; (d) aq CH_3CN , then cat. TEMPO, NaOCl , NaClO_2 , 81%.

ii Synthesis of Manzacidin

Enantioselective Synthesis of the Bromopyrrole Alkaloids Manzacidin A and C by Stereospecific C–H Bond Oxidation

JACS. 2002, 124, 12950.

Scheme 9 Retro-synthesis



Scheme 10 Total synthesis

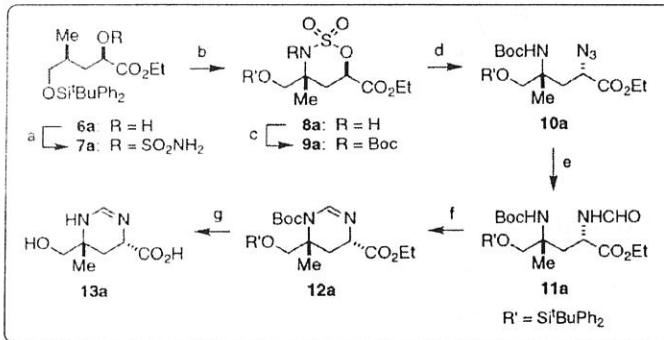
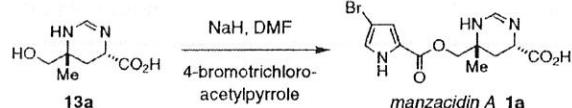


Table 2 Hydrogenation Reaction

entry	catalyst ^a	ligand	6a/6c ^b
1	5% Pt-C	none	50:50
2	(Ph ₃ P) ₂ RhCl	none	50:50
3	Ir(cod)(pyr)(PCy ₃) ₂ PF ₆	none	65:35
4	Rh(cod) ₂ OTf	(R)-PHANEPHOS ^c	75:25
5	Rh(nbd) ₂ BF ₄	dppb	40:60
6	Rh(nbd) ₂ BF ₄	(R)-BINAP	20:80
7	Rh[((S,S)-Et-DUPHOS)(cod)]OTf	none	>5:95



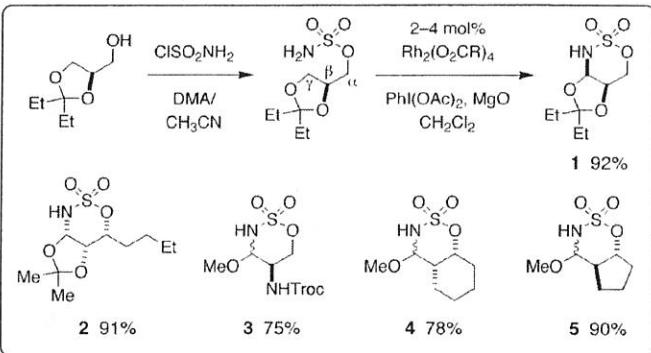
^a Conditions: (a) CISO_2NCO , HCO_2H , 87% (3:1 mixture of C6 epimers); (b) 2 mol % $\text{Rh}_2(\text{OAc})_4$, $\text{PhI}(\text{OAc})_2$, MgO , CH_2Cl_2 , 85%; (c) Boc_2O , $\text{C}_5\text{H}_5\text{N}$; (d) NaN_3 , DMF , 92%, two steps; (e) H_2 , $\text{Pd}-\text{C}$, then N -formylbenzotriazole; (f) POCl_3 , $2,6\text{-Bu}_2\text{-4-MeC}_6\text{H}_2\text{N}$, 73%, two steps; (g) 8 M HCl , DME , 60 °C, NaHCO_3 , 60 °C, 99%.

iii After Manzacidin

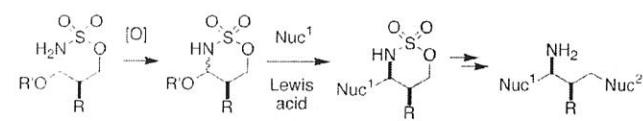
a Novel Iminium Ion Equivalents Prepared through C–H Oxidation for the Stereocontrolled Synthesis of Functionalized Propargylic Amine Derivatives

JACS. 2003, 125, 2028

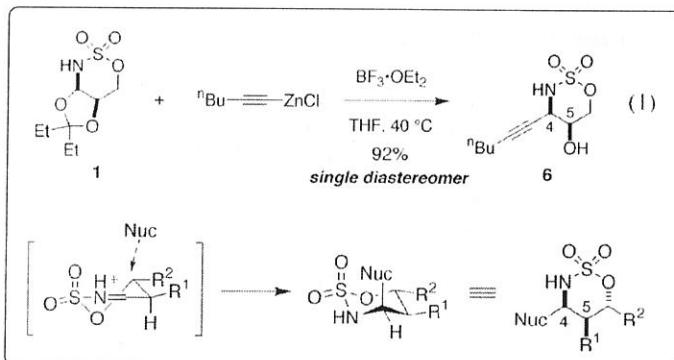
Scheme 11 Rh-catalyzed C - H insertion furnishes *N*,*O*-acetal substrates



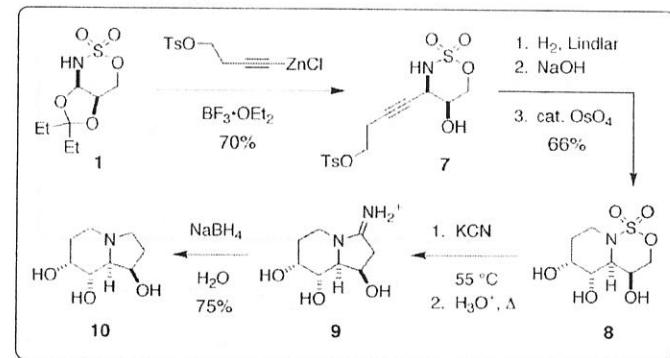
Concept: N,O-acetal oxathiazinane heterocycles as reactive iminium ion equivalents



Scheme 12

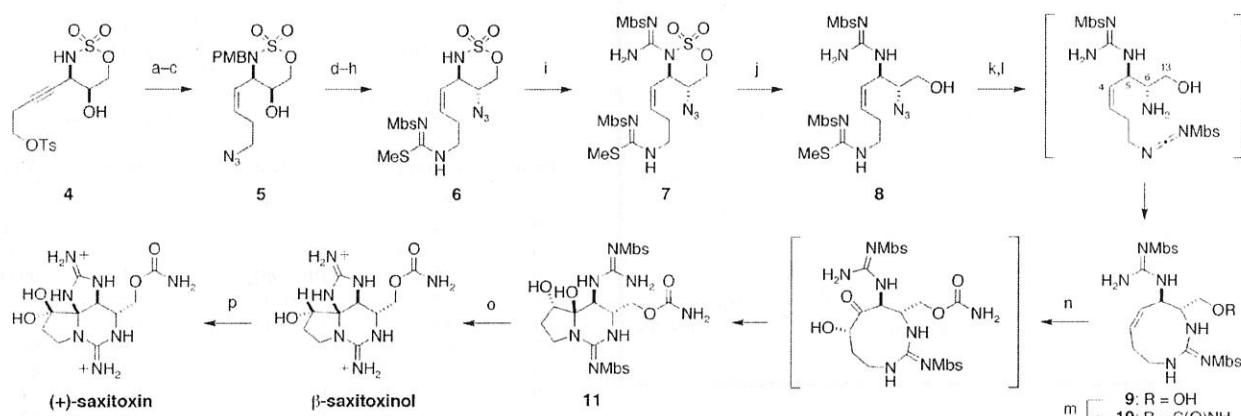
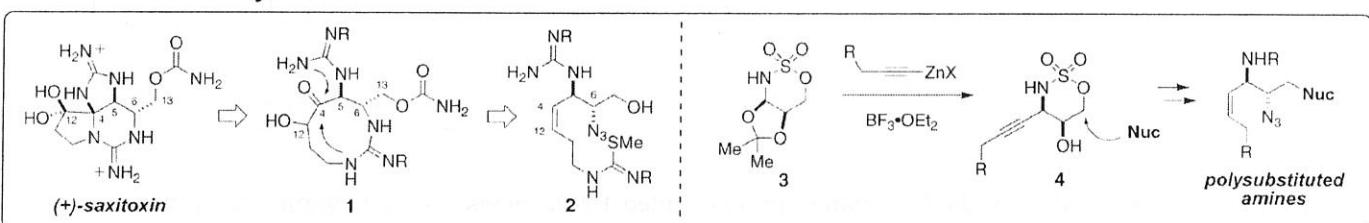


Scheme 13 Application



b A Synthesis of (+)-Saxitoxin JACS. 2006, 128, 3926.

Scheme 14 Retro-synthesis



Conditions: (a) H_2 , $\text{Pd/CaCO}_3/\text{Pb}$, THF ; (b) NaN_3 , ${}^\circ\text{Bu}_4\text{NI}$, DMF , 90% (2 steps); (c) $p\text{-MeOC}_6\text{H}_4\text{CH}_2\text{Cl}$, ${}^\circ\text{Bu}_4\text{NI}$, K_2CO_3 , CH_3CN , 85%; (d) Me_3P , $\text{THF-H}_2\text{O}$; (e) $\text{Me}_2(\text{Cl})\text{C}\equiv\text{NMbs}$, ${}^\circ\text{Pr}_2\text{NEt}$, CH_3CN , 72% (2 steps); (f) Tl_2O , $\text{C}_5\text{H}_5\text{N}$, DMAP , CH_2Cl_2 ; (g) NaN_3 , DMF , -15°C , 70% (2 steps); (h) $(\text{NH}_4)_2\text{Ce}(\text{NO}_3)_6$, ${}^\circ\text{BuOH}/\text{CH}_2\text{Cl}_2$, 74%; (i) KO°Bu , $\text{Cl}_2\text{C}\equiv\text{NMbs}$; then $(\text{Me}_2\text{Si})_2\text{NH}$, 70% (+20% of 6); (j) aq. CH_3CN , 70°C , 95%; (k) Me_3P , $\text{THF-H}_2\text{O}$; (l) Ag_2NO_3 , Et_3N , CH_3CN , 65% (2 steps); (m) $\text{Cl}_3\text{CC(O)NCO}$, $\text{THF/CH}_3\text{CN}$, -78°C ; then K_2CO_3 , MeOH , 82%; (n) 10 mol % of OsCl_3 , Oxone , Na_2CO_3 , $\text{EtOAc/CH}_3\text{CN/H}_2\text{O}$, 57%; (o) $\text{B}(\text{O}_2\text{CCF}_3)_3$, $\text{CF}_3\text{CO}_2\text{H}$, 82%; (p) DCC , $\text{C}_4\text{H}_9\text{N}\cdot\text{HO}_2\text{CCF}_3$, DMSO , 70%. Mbs = $p\text{-MeOC}_6\text{H}_4\text{SO}_2$.

c Catalytic Intermolecular Amination of C–H Bonds JACS. 2007, 129, 562

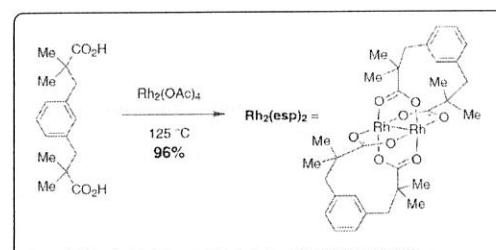
Table 3 Varying the influencece of the nitren precursor

Entry	$\text{Ph}-\text{CH}_2-\text{Me}$	H_2NR	Yield ^{a,b}	2 mol% $\text{Rh}_2(\text{esp})_2$	
				Phl($\text{O}_2\text{C}^\circ\text{Bu}$) ₂	$\text{Phl}(\text{O}_2\text{C}^\circ\text{Bu})_2$
1	$\text{H}_2\text{N}-\text{S}(\text{O})_2-\text{CH}_2\text{CCl}_3$	72	5	$\text{H}_2\text{N}-\text{S}(\text{O})_2-\text{NO}_2$	29
2	$\text{H}_2\text{N}-\text{S}(\text{O})_2-\text{C}^\circ\text{Bu}$	47	6	$\text{H}_2\text{NSO}_2\text{CH}_3$	20
3	$\text{H}_2\text{N}-\text{S}(\text{O})_2-\text{C}_6\text{H}_4-\text{CH}_3$	20	7	$\text{H}_2\text{NSO}_2\text{CF}_3$	35
4	$\text{H}_2\text{N}-\text{S}(\text{O})_2-\text{C}_6\text{H}_4\text{CH}_3$	< 5	8	H_2NCOCF_3	< 5

Temp:rt, Sol: C_6H_6

Table 4 Comparison of rhodium catalyst

catalyst	%conv. ^a
$\text{Rh}_2(\text{O}_2\text{C}^\circ\text{Bu})_4$	< 5
$\text{Rh}_2(\text{O}_2\text{CPh}_3)_4$	10
$\text{Rh}_2(\text{NHCOCF}_3)_4$	35
$\text{Rh}_2(\text{S-TCPTAD})_4$	15
$\text{Rh}_2(\text{S-biTISP})_2$	0
$\text{Rh}_2(\text{esp})_2$	75 (70) ^b



$\text{Rh}_2(\text{esp})_2$: Aldrich: ¥7000/100mg

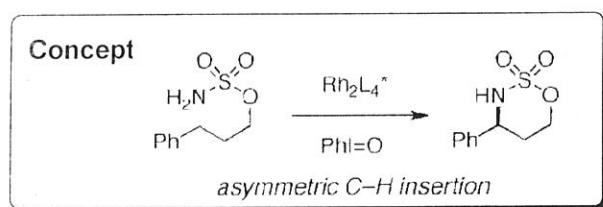


Table 5 Chiral rhodium carboxamidate catalyst

Rh_2L_4	2 mol% Rh_2L_4 PhI=O 3 Å MS	Rh_2L_4	$E_{\text{ox}}^{(\text{VII} \rightarrow \text{VIII})}$	% yield	% ee
1		1	11 mV	< 5	—
2		2	120 mV	22	54
3		3	242 mV	< 5	nd
4		4	330 mV	85	92
5		5	742 mV	< 10	nd

$\text{Rh}_2\text{L}_4 =$

Table 6 Substrat scope

1. R = H 2. R = OMe 3. R = CF ₃	1. R = H 2. R = OMe 3. R = CF ₃	1. R = H 2. R = OMe 3. R = CF ₃
4. 45%, 85% ee 5. 98%, 92% ee	4. 45%, 85% ee 5. 98%, 92% ee	4. 45%, 85% ee 5. 98%, 92% ee
6. 72%, 63% ee	7. 87%, 99% ee	8. 55%, 94% ee

e Catalytic C–H Amination for the Preparation of Substituted 1,2-Diamines JACS. 2008, 130, 11248

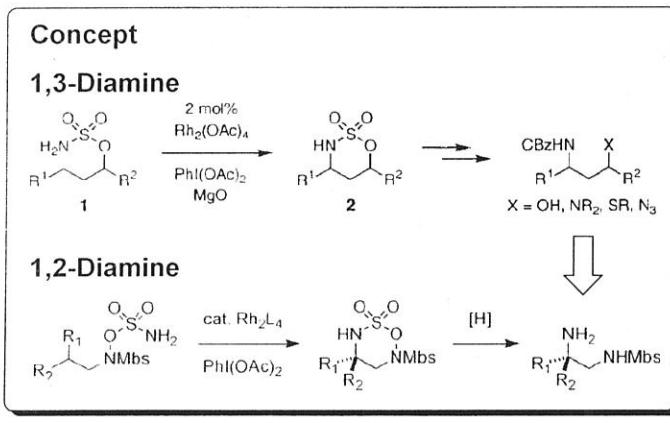


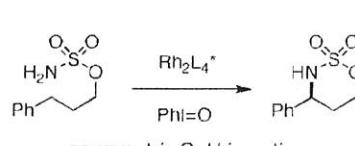
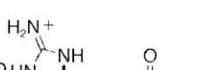
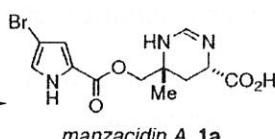
Table 7 Modification

Entry ^a	R Group	Catalyst	Solvent	Yield ^b	
				Yield (%)	Conversion (%)
1	Boc	$\text{Rh}_2(\text{oct})_4$	C ₆ H ₆	— (29)	—
2	CHO	$\text{Rh}_2(\text{oct})_4$	C ₆ H ₆	— (23)	—
3	MeSO ₂	$\text{Rh}_2(\text{oct})_4$	C ₆ H ₆	— (100)	—
4	Mbs	$\text{Rh}_2(\text{oct})_4$	C ₆ H ₆	99 (100)	99
5	Mbs	$\text{Rh}_2(\text{oct})_4$	toluene	77	—
6	Mbs	$\text{Rh}_2(\text{oct})_4$	CH ₂ Cl ₂	74	—
7	Mbs	$\text{Rh}_2(\text{oct})_4$	EtOAc	90	—
8	Mbs	$\text{Rh}_2(\text{OAc})_4$	C ₆ H ₆	87	—
9	Mbs	$\text{Rh}_2(\text{O}_2\text{CCPh}_3)_4$	C ₆ H ₆	43	—
10	Mbs	$\text{Rh}_2(\text{esp})_2$	C ₆ H ₆	98	—

^a Reactions were performed with 2 mol % of catalyst, 2.3 equiv of MgO, and 1.1 equiv of PhI(OAc)₂. ^b Isolated yields; values in parentheses denote percent conversions based on integration of the unpurified ¹H NMR spectrum.

Summary of "Rh₂L₄, PhI(OR)₂, MgO" system

- 1)
- 2)
- 3)
- 4)



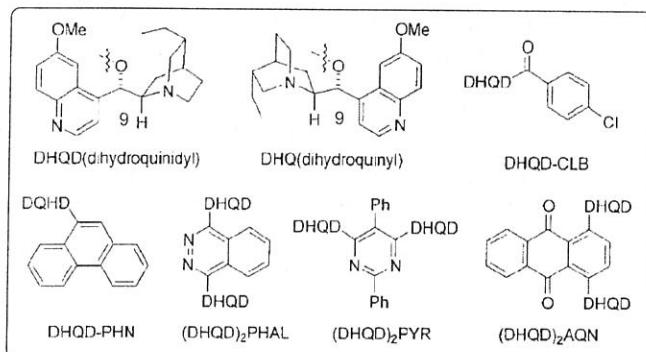
One catalytic system, but different SM to get different useful building block!

3 Li Deng group:

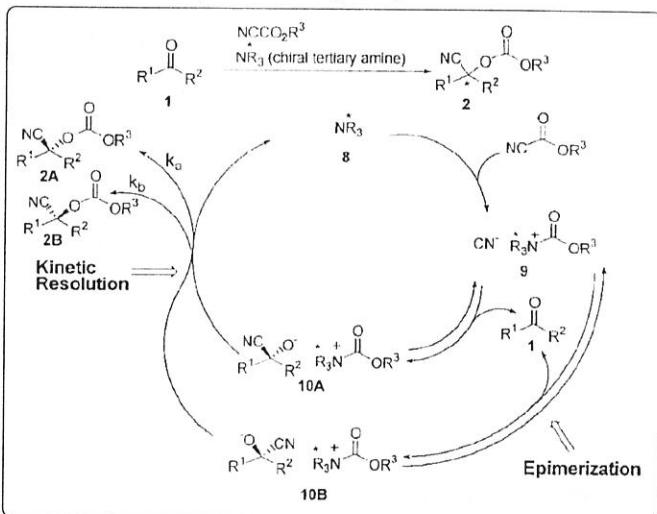
i Before Manzacidin

a A Highly Enantioselective Chiral Lewis Base-Catalyzed Asymmetric Cyanation of Ketones
JACS 2001, 123, 6195.

Fig 1 Structures of modified cinchona alkaloids

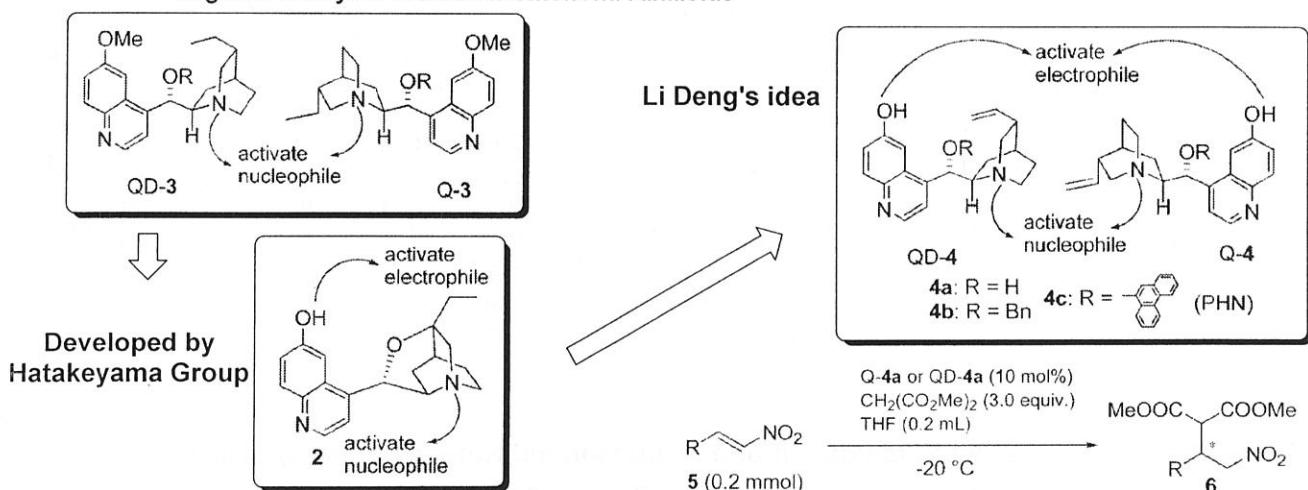


Scheme 15 Proposed catalytic cycle



b Highly Enantioselective Conjugate Addition of Malonate and β -Ketoester to Nitroalkenes: Asymmetric C–C Bond Formation with New Bifunctional Organic Catalysts Based on Cinchona Alkaloids

JACS 2004, 126, 9906.

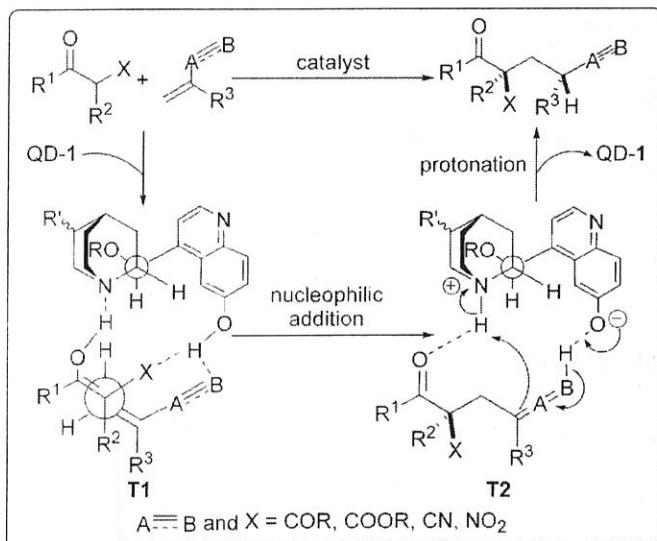


ii Synthesis of Manzacidin

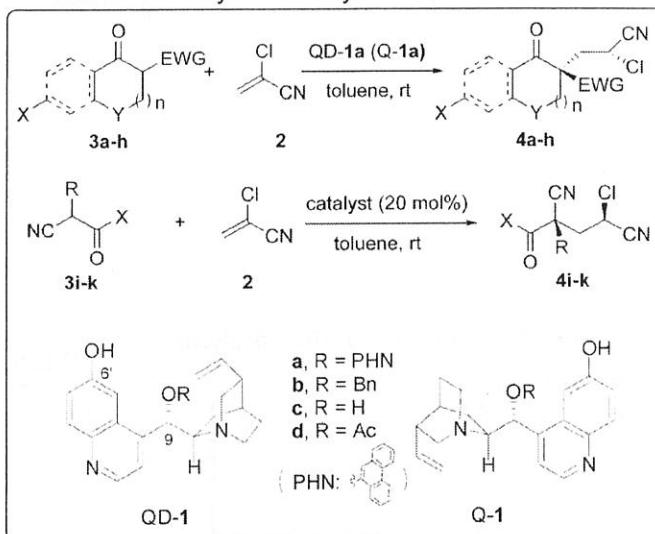
Dual-Function Cinchona Alkaloid Catalysis: Catalytic Asymmetric Tandem Conjugate Addition–Protonation for the Direct Creation of Nonadjacent Stereocenters

JACS 2006, 128, 3928.

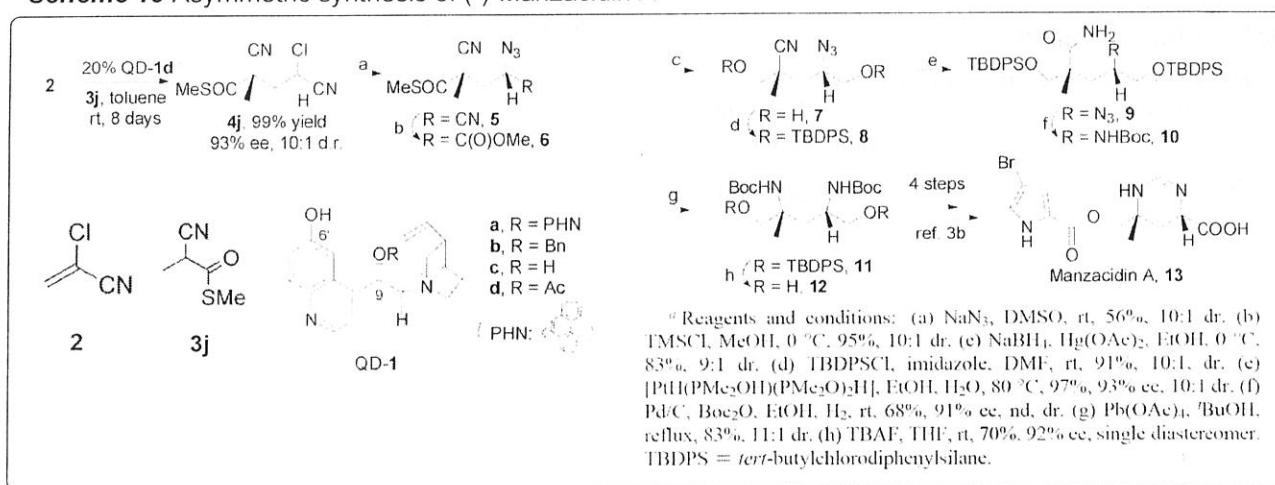
Scheme 16 Proposed model for Tandem conjugate addition-Protonation



Scheme 17 Cycl and acyclic Michael Donors



Scheme 18 Asymmetric synthesis of (-)-Manzacidin A

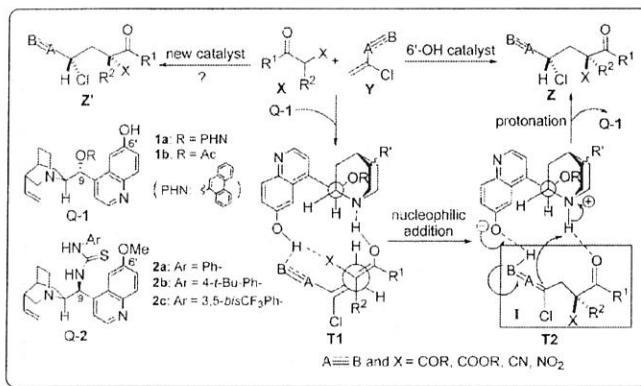


iii After Manzacidin

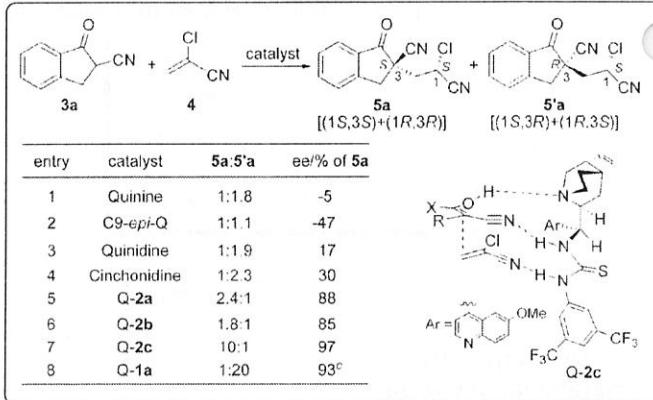
- a Control of Diastereoselectivity in Tandem Asymmetric Reactions Generating Nonadjacent Stereocenters with Bifunctional Catalysts by Cinchona Alkaloids

JACS 2007, 129, 768.

Scheme 19 Proposed model

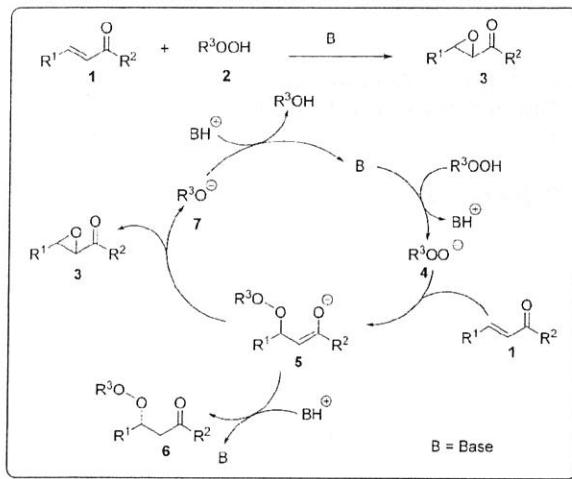


Scheme 20 Reaction using Q-2

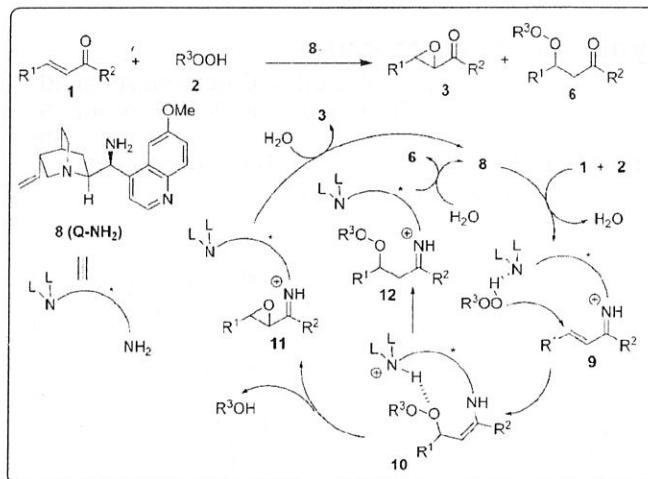


- b Catalytic Enantioselective Peroxidation of α,β -Unsaturated Ketones JACS 2008, 130, 8134.

Scheme 21 Mechanism of epoxidation



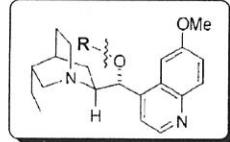
Scheme 22 Proposed catalytic cycle



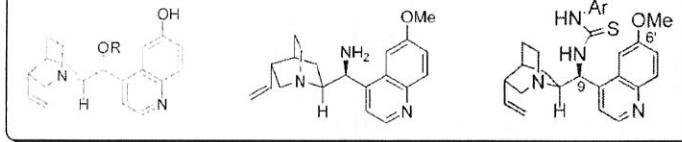
Summary: Development of the catalysts

Chiral Lewis

Base Catalysts:



Bifunctional Catalysts



4 Maruoka group

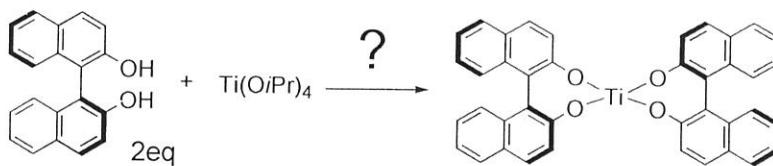
i Before Manzacidin

a Bis(((S)-binaphthoxy)(isopropoxy)titanium) Oxide as a μ -Oxo-Type Chiral Lewis Acid: Application to Catalytic Asymmetric Allylation of Aldehydes

JACS 2003, 125, 1708.

Background: Keck et al.

JACS 1993, 115, 8467.



Maruoka group: Ti-O-Ti

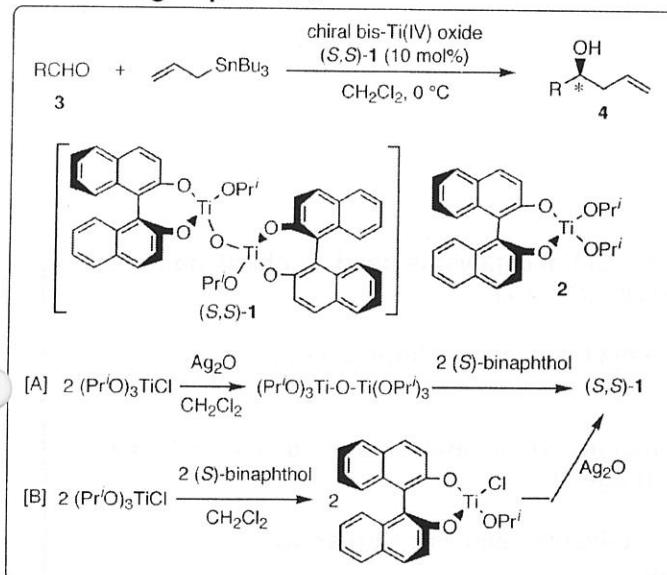


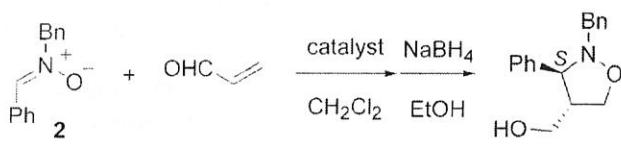
Table 8

entry	aldehyde	Ti catalyst (mol %)	time (h)	% yield ^b	% ee ^c (config) ^d
1	PhCH ₂ CH ₂ CHO	1 (10)	4	84	99 (<i>R</i>)
2		1 (10) ^e	16	82	98 (<i>R</i>)
3		1 (5)	12 [^f]	77 [95]	98 [98] (<i>R</i>)
4	CH ₃ (CH ₂) ₆ CHO	1 (10)	12	85	99 (<i>R</i>) ^g
5		2 (20)	12	14	81 (<i>R</i>) ^g
6		1 (5)	24 [^f]	86 [92]	99 [98] (<i>R</i>) ^g
7	(CH ₃) ₂ CHCHO	1 (10)	28 [^f]	71 [91]	>99 [99] (<i>S</i>) ^h
8		2 (20)	28	7	85 (<i>S</i>) ^h
9	PhCH=CHCHO	1 (10)	15	70	95 (<i>S</i>)
10	PhCHO	1 (10)	7	90	96 (<i>S</i>)
11		1 (10) ^e	24	81	96 (<i>S</i>)
12		1 (5)	[9] ^f	[94]	[97] (<i>S</i>)
13	<i>p</i> -bromobenzaldehyde	1 (10)	15	85	98 (<i>S</i>)
14	furfural	1 (10)	18	96	97 (<i>S</i>)

b Asymmetric 1,3-Dipolar Cycloaddition Reaction of Nitrones and Acrolein with a Bis-Titanium Catalyst as Chiral Lewis Acid

JACS 2005, 127, 11927.

Table 9 Ti catalyst for 1,3 dipolar cycloaddition



entry	catalyst	(mol %)	conditions ($^\circ C$, h)	yield (%) ^{b,c}	ee (%) ^d [config] ^e
1	(<i>S,S</i>)-1	10	0, 2	78	89 [<i>S</i>]
2	$Ti(Oi-Pr)_4$	20	0, 2	40	60 [<i>S</i>]
3	CITi(<i>Oi-Pr</i>) ₃	20	0, 2	36	60 [<i>S</i>]
4	(<i>S,S</i>)-1	10	-20, 17	90	91 [<i>S</i>]
5	(<i>S,S</i>)-1	10	-40, 24	94	93 [<i>S</i>]

ii Synthesis of Manzacidin A: The most efficient route.

JACS 2006, 128, 2174.

Enantioselective 1,3-Dipolar Cycloaddition Reaction between Diazoacetates and α -Substituted Acroleins: Total Synthesis of Manzacidin A

Scheme 23 Retro-synthesis

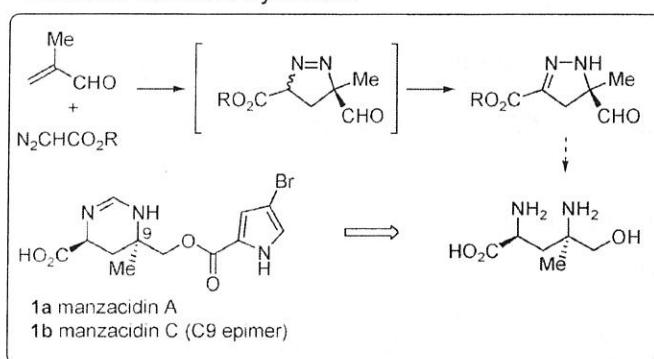


Table 10 Titanium BINOLates

entry	R ¹	catalyst (mol %)	conditions ($^\circ C$, h)	yield (%) ^b	ee (%) ^c
1	Et	—	rt, 40	16	—
2	Et	2a (10)	0, 1	—	—
3	Et	2a (10)	-40, 4	42	88
4	Et	2b (10)	-40, 2	54	90
5	Et	2c (5)	-40, 3	52	95
6	t-Bu	2b (10)	-40, 1	52	91
7	t-Bu	2c (5)	-40, 1	43	94

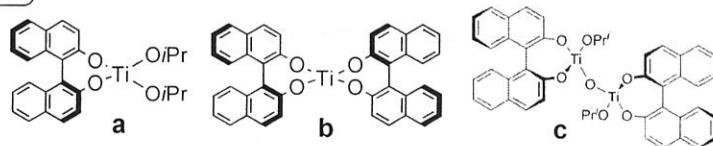
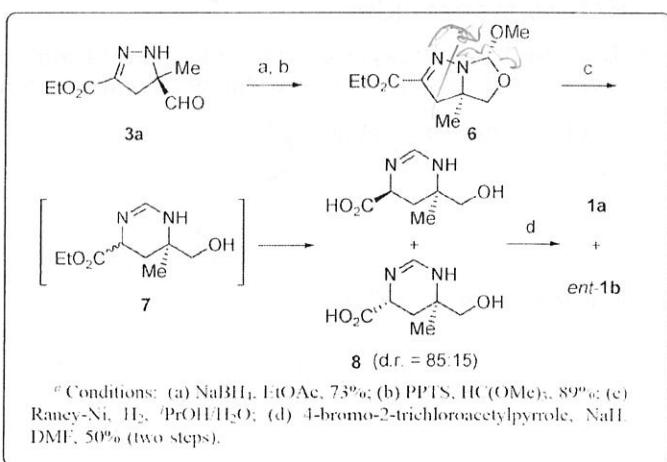


Table 11 Substrat scope

entry	R ²	catalyst (mol %)	time (h)	yield (%) ^a	ee (%) ^c	titanium BINOLate 2b-c
						CH ₂ Cl ₂ , -40 °C
1	Me	2b (10)	1	52	91 ^d	Bu' ² O ₂ C / N-HH-C(=O)-CH ₂ -R ²
2	Me	2c (5)	1	43	94 ^d	
3	Et	2b (10)	3	63	83	
4	Et	2c (5)	3	48	84	
5	BnOCH ₂ CH ₂	2b (10)	1	81	80	
6	PhCH ₂ CH ₂	2b (10)	4	63	82	
7	i-Pr	2b (10)	3	82	92	
8	Cy	2b (10)	5	77	94	
9	Cy	2c (5)	5	75	94	

Scheme 24 Total synthesis



Summary of total synthesis of Manzacidin A

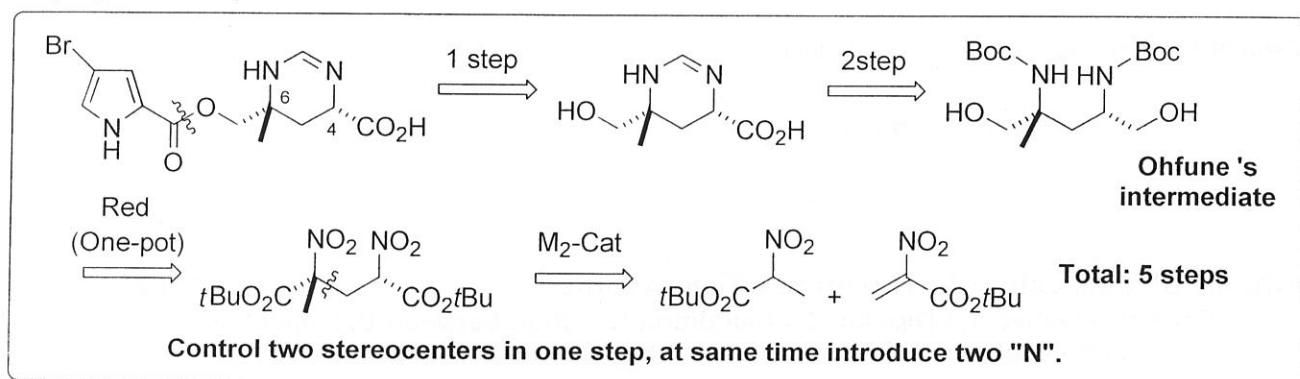
1 Ohfune Group: Diastereoselective Strecker reaction, Chiral catalyst is used as chiral source, First highly stereoselective synthesis. (Page 2)

2 Du Bois Group: Oxidative C-H insertion, Chiral catalyst is used as chiral source, New synthesis method. (Page 4)

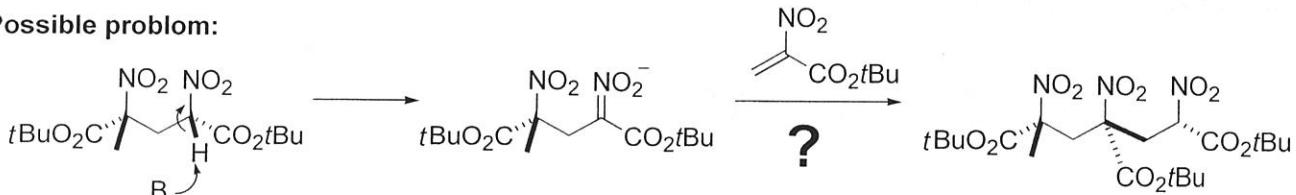
3 Li Deng Group: Tandem conjugate addition-Protonation, Chiral catalyst is used as chiral source, Stereoselective efficient method. (Page 8)

4 Maruoka Group: 1,3 Dipolar cycloaddition, Chiral catalyst is used as chiral source, The most efficient route. (Page 10)

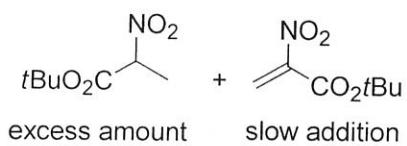
My synthesis plan:



Possible problem:



Solution a



Solution b: Other Electrophiles

