

Lamps in Kanai laboratory



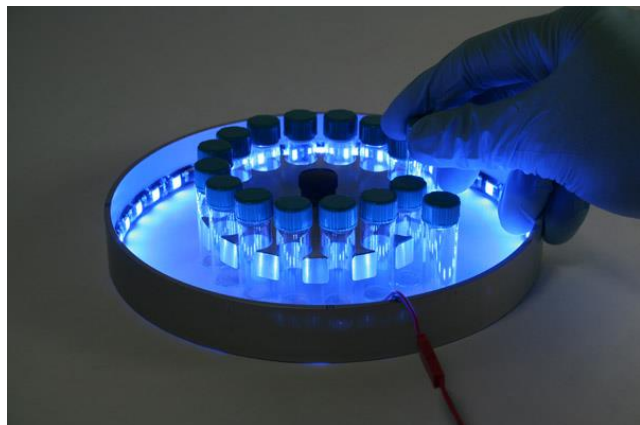
White LED (visible light)
株式会社ヴァロール製

http://www.valore.jp/led/vbl_s.html



LED of specific wavelength (e.g. purple or green)
株式会社ヴァロール製

http://www.valore.jp/led/vbl_s.html



Aldrich LED (435 or 400 nm)

<http://www.sigmaaldrich.com/japan/labware/micro-photochemical-reactor.html>



UV LED (365 nm) 大興製作所製
<http://www.daico.co.jp/products/products.php?id=21>

Setup of photoreactions (visible light LED)



Reaction vessel

LED lamp

- Put the reaction vessel ca. 2 cm from the lamp.
- White LED is stored in a box behind the old glove box (4F refresh room).
- Light intensity can be modulated with 2 dials.



Intensity dial

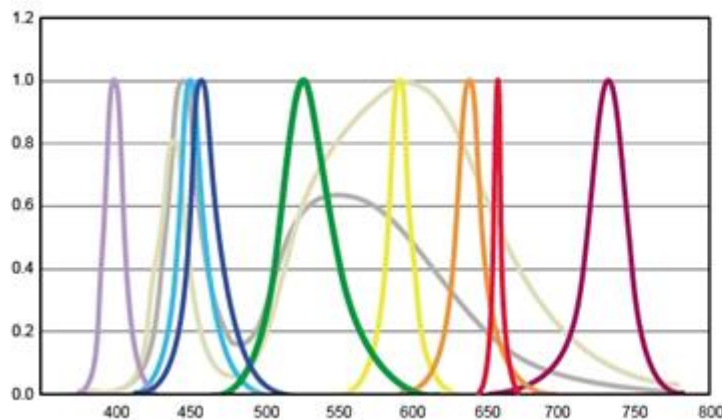
On/Off switch

Controller of Valore LEDs

VBL-SL150/SL300 series 搭載光

※ 400nmにレンズは装着できません。

Spectra of
Valore LEDs



I : 730nm*
R : 660nm*
R2 : 627nm
Y : 590nm
G : 530nm
B : 470nm
V : 400nm
W : White
L : WarmWhite

*はピーク波長、
その他はドミナント波長。

http://www.valore.jp/led/vbl_s.html

Setup of photoreactions (UV)



Run reactions in the UV-cut box.
(It is in the measurement room of 4th floor.)

[CAUTION!]

UV is harmful. Never use UV lamps out of the box.

37-degree experiment

500 nm LEDs are used in Sohma group for biological study.

They are stored in

- Incubator at 4F
- Eisai 1F

Tips

[In general]

- Photoreactions should be run in a test tube or a screw vial. Flasks do not give good results.
- Large scale reactions usually result in poor yield.

[Reaction in a screw vial]

- Vials can be put in a fraction collector (large), then in front of the lamp. (See appendix 3)
- Aldrich LED is suitable for a 4 mL screw vial.

[For visible light]

- White LED is recommended for initial trials.
- Light is too shiny for people around you. Put cartons or aluminum foils around the reaction.

[For UV]

- UV light is harmful, so be sure to use it inside the UV-cut box.
- Turn on the fan attached on the UV-cut box (in order to release the heat inside the box).

[Recommended reviews of photoredox catalysis]

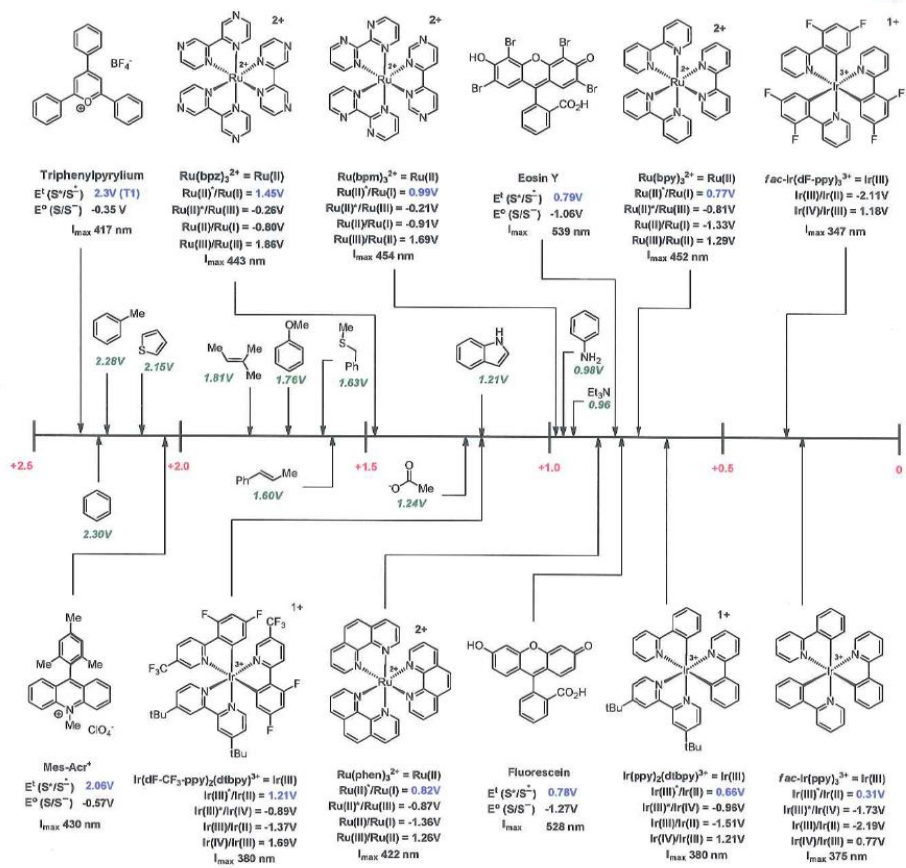
- 有機合成化学協会誌 2014, 72, 538.
- *Chem. Rev.* **2013**, 113, 5322. (with λ_{max} and $E_{1/2}$ of representative photoredox catalysts)

Appendix 1: Redox potential of common photocatalysts

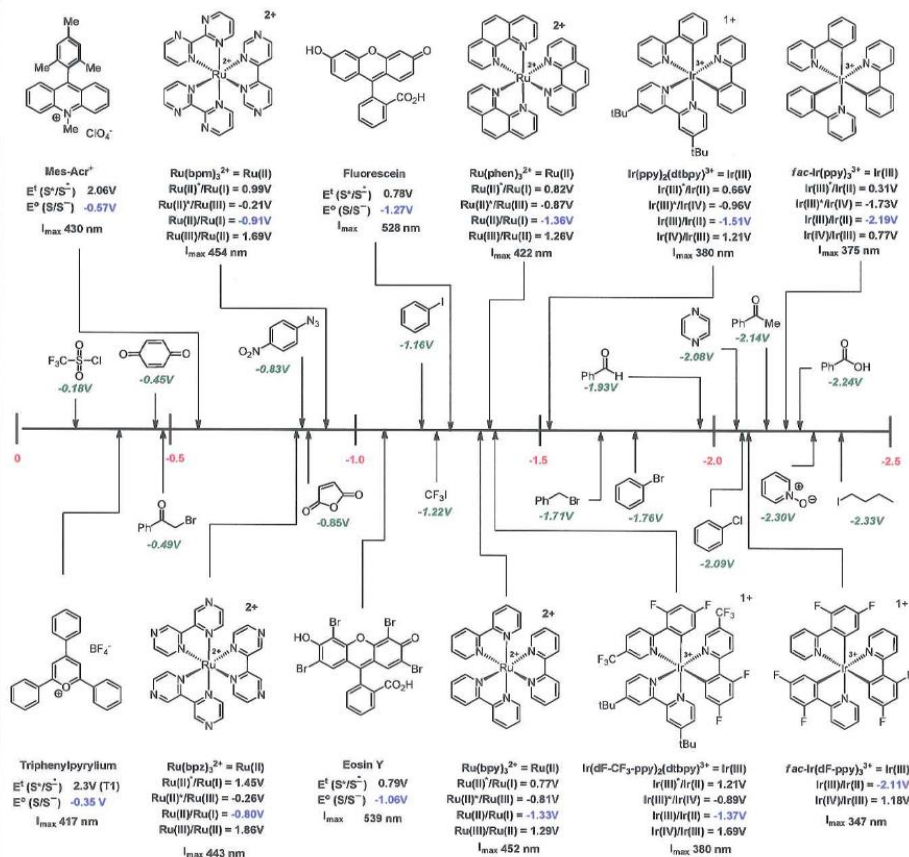
Electrochemical Series of Photocatalysts and Common Organic Compounds

Excited State Oxidations

E° (V vs SCE)

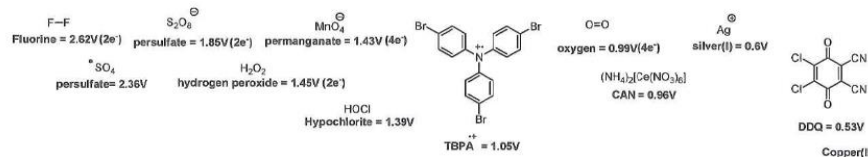


Reduced State Reductions

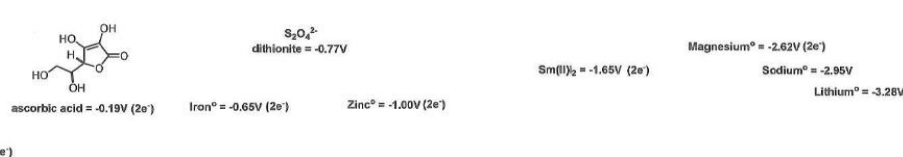


Common Oxidizing Agents

E° (V vs SCE)



Common Reducing Agents



compiled by D. DiRocco 2014

Appendix 2: Absorption and excited state lifetime of common photocatalysts

entry	photocatalyst	excited-state lifetime, τ (ns)	excitation λ_{max} (nm)	emission λ_{max} (nm)	ref
1	$\text{Ru}(\text{bpm})_3^{2+}$	131 ^b	454	639 ^b	161
2	$\text{Ru}(\text{bpz})_3^{2+}$	740	443	591	55
3	$\text{Ru}(\text{bpy})_3^{2+}$	1100	452	615	1, 3
4	$\text{Ru}(\text{phen})_3^{2+}$	500	422	610 ^c	1, 129
5	$\text{Ir}[\text{dF}(\text{CF}_3)\text{ppy}]_2(\text{dtbbpy})^+$	2300	380	470	77
6	$\text{Ir}(\text{ppy})_2(\text{dtbbpy})^+$	557		581	58, 77
7	$\text{Cu}(\text{dap})_2^+$	270		670 ^d	33
8	<i>fac</i> - $\text{Ir}(\text{ppy})_3$	1900	375	494 ^e	38

Modified from *Chem. Rev.* **2013**, 113, 5322.

Appendix 3: Photoreactions with many vials



3 vials can be piled up in
one fraction collector.