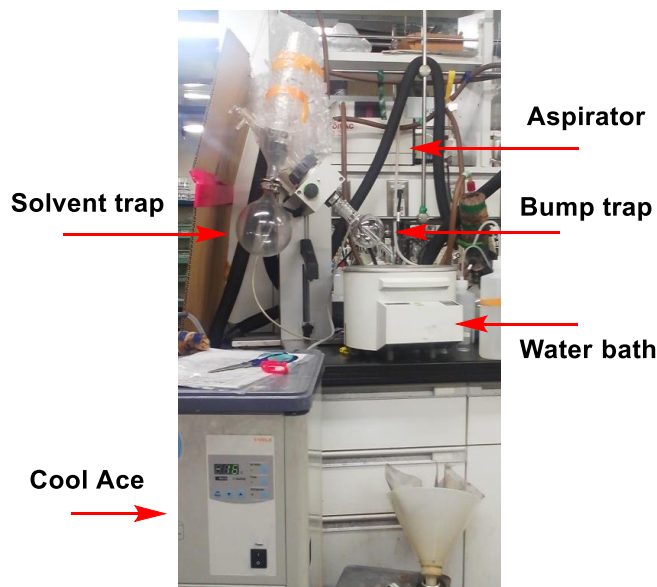


# Rotary Evaporator



## Operation

1. Turn on Cool Ace (all the day).
2. Remove remaining solvents in the solvent trap before use.
3. Attach a flask containing the solution to the bump trap with a joint clip.  
(Stick the stirrer bar to the upper side of the flask with a magnet in order to avoid bumping.)  
**Bad-smelling solutions should be evaporated in the fume hood.**  
**KOH trap should be used when the solution contains acid.**
4. Start the rotation and turn on the aspirator.
5. Heat the water bath if necessary.
6. When all solvent has been removed, turn off the vacuum and stop the rotation.  
**Don't leave the flask for the long time when finished.**

## After use

1. Turn off the water bath after use.
2. **Clean up bump trap, solvent trap and water bath if it gets dirty.**

## Glass vial

The vials can be connected to the bump trap with silicon adapter.  
(Be careful not to drop the sample!)



## KOH trap

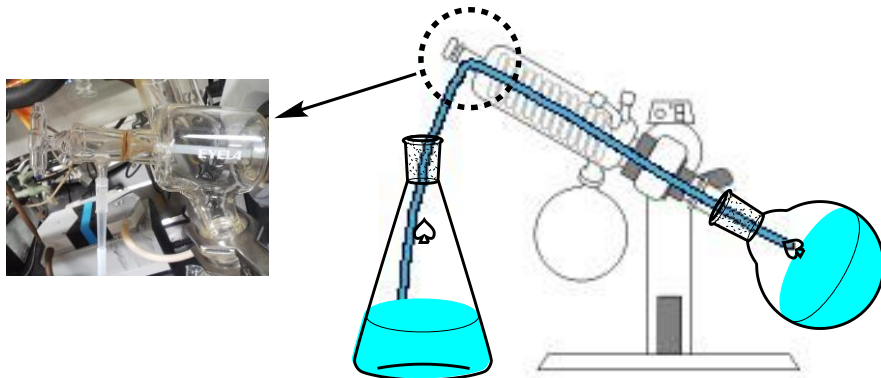


When the solution contains **Acid**, **Thiol** or **Sulfide**, **KOH trap** or **HClO<sub>2</sub> trap** should be used to prevent the damage to aspirator.

## After using acid

1. **Wash the inside of evaporator with acetone.**
2. **Keep the aspirator idling for more than 5 minutes.**

## Large scale evaporation



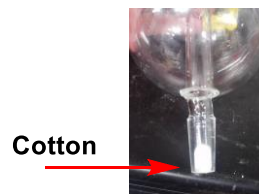
### Principle

- ◆ The solution in outer flask is drawn into inner flask under reduced pressure.
- ◆ Removal of solvents and injection of the solution are in equilibrium.

### Attention

Large flasks ( $\geq 1$  L) should be attached directly (without using the bump trap) and put in the water bath from the beginning not to damage the rotary axis.

## Solvent containing powder



Block the bump trap with **cotton** when the solution contains powder such as silica gel.

## Azeotropic data

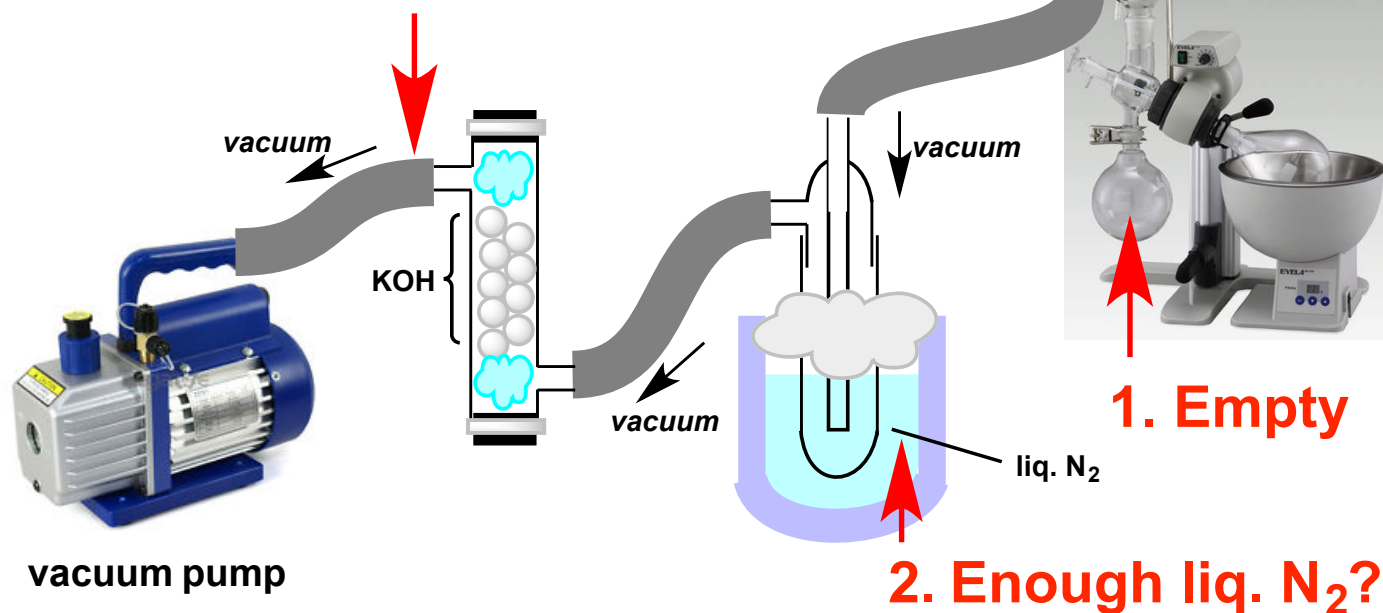
High boiling point solvent can be removed by mixing with another solvent which makes azeotropic mixture.

$t_{az}$  = Azeotropic point,  $X_1$  = mole fraction of component 1

Component 1	Component 2	$t_{az}$ (°C)	$X_1$
Water	Toluene	84.1	0.44
	Benzene	69.3	0.30
	2-Propanol	80.4	0.35
	1-Butanol	92.7	0.75
	2-Butanol	87.0	0.60
Acetic acid	Hexane	68.3	0.13
	Heptane	91.7	0.45
Pyridine	Toluene	110.1	0.25
DMF	Heptane	97.0	0.08

## Evaporate high-boiling point solvent with rotavap

### 3. Release the vacuum & turn off the pump after use



## Purpose of protocol

# When you want to evaporate high-boiling point solvent, such as DMF, you may use a vacuum pump connected to the rotavap instead of a diaphragm pump.

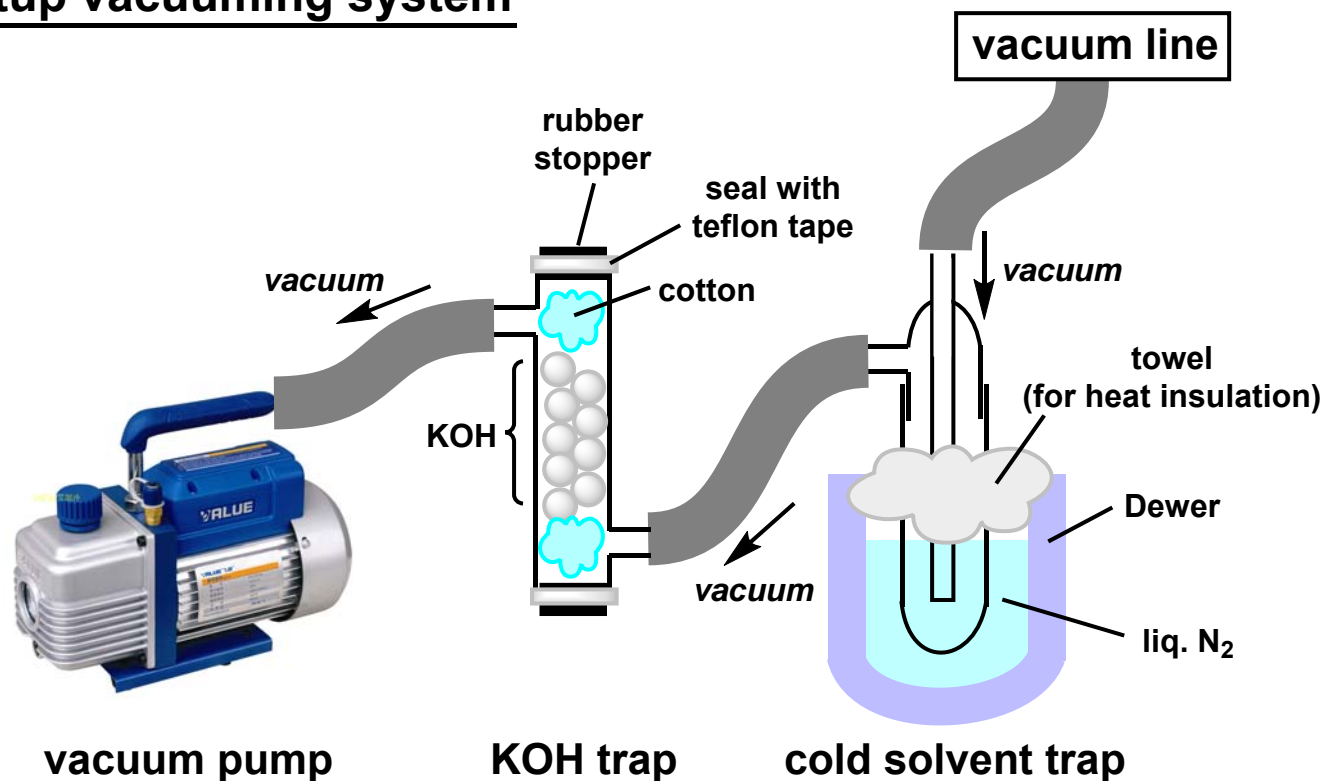
**Prevent damage** on the vacuum pump by **carefully trapping solvent vapor**.

## TIPS

1. **Empty the 1st solvent trap** to prevent vacuuming low-boiling point solvent, such as AcOEt.
2. Check whether there is **enough liq. N<sub>2</sub>** in the dewar.
3. **Make sure to turn off** the pump after you use, otherwise liq. N<sub>2</sub> will evaporate to eventually cause inflow of the trapped solvent vapor to the pump and damage it.

# Temp. of water bath: For water, ~30 °C. For DMF, ~40 °C.

## How to setup vacuuming system

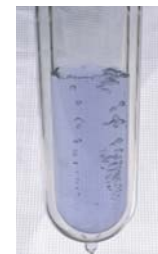


## Purpose of protocol

# **Cold solvent trap** is used for capturing solvent vapor. **KOH trap** is used for capturing moisture and acidic volatiles. They often damage vacuum pump severely if directly entered. You **ALWAYS** have to connect both between pump and vacuum line in order to minimize the damage.

## TIPS

# You have to evacuate inside of the trap **BEFORE** cool down in liq N<sub>2</sub>. If you cool down before vacuum, **explosive liquid O<sub>2</sub> will be generated** inside. You have to avoid such a very dangerous situation.



liquid O<sub>2</sub>  
(blue-colored)